

MODELLING OF EMULSION POLYMERIZATION IN A CONTINUOUS STIRRED TANK REACTOR

A Thesis Submitted
In Partial Fulfilment of the Requirements
for the Degree of

MASTER OF TECHNOLOGY

By

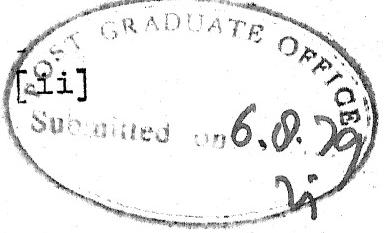
GIRISH DAMODAR BALLAL

to the

DEPARTMENT OF CHEMICAL ENGINEERING

INDIAN INSTITUTE OF TECHNOLOGY, KANPUR

AUGUST, 1979



CERTIFICATE

This is to certify that the present work 'MODELLING OF EMULSION POLYMERIZATION IN A CONTINUOUS STIRRED TANK REACTOR' has been carried out under my supervision and has not been submitted elsewhere for a degree.

Date: 4th August 1979

K. S. Gandhi

[K. S. Gandhi]

Professor

Department of Chemical Engineering
Indian Institute of Technology
Kanpur-208016, India

POST GRADUATE OFFICE

This thesis has been approved
for the award of the Degree of
Master of Technology (M.Tech.)
in accordance with the
regulations of the Indian
Institute of Technology Kanpur

Dated. 10.8.79 21

CHE-1979-M-BAL-MOD

I.I.T. KANPUR
CENTRAL LIBRARY
Acc. No. A 59503.

16 SEP 1979

CONTENTS

ABSTRACT	...	vi
CHAPTER 1 INTRODUCTION	...	1
CHAPTER 2 PHYSICAL MODEL AND POPULATION BALANCE EQUATIONS	...	5
CHAPTER 3 FORMULATION OF MODELLING EQUATIONS		14
CHAPTER 4 METHOD OF SOLUTION OF THE MODELLING EQUATIONS	...	34
CHAPTER 5 RESULTS AND DISCUSSION		56
CHAPTER 6 CONCLUSIONS	...	76
REFERENCES	...	77

LIST OF FIGURES

FIGURE		Page
1	Scheme for Solution of Modelling Equations ...	52
2	Particle Size Distribution	60
3	Effect of Mean Residence Time on Particle Size Distribution ...	61
4	Effect of Mean Residence Time on Number of Particles and Rate of Polymerization	63
5	Effect of Initiator Concentration on Number of Particles and Rate of Polymerization ...	66
6	Effect of Mean Residence Time on Molecular Weights ...	69
7	Effect of Initiator Concentration on Molecular Weights ...	71
8	Molecular Weight Distribution of Polymer (Weight Basis)	75

ABSTRACT

A general modelling framework was suggested by Min and Ray [17] for emulsion polymerization reactions. A simplified model based on the framework has been proposed to simulate the emulsion polymerization reaction in a continuous stirred tank reactor. The model uses the population balance equation approach. The method of moments has been used to solve the population balance equations. The model is capable of calculating conversion, rate of polymerization, number of particles, particle size distribution and the molecular weight distribution of resultant polymer latex. The variables studied were residence time and initiator concentration. The results obtained using this model were compared with the experimental data and the theoretical results of DeGraff and Poehlein [21]. The proposed model is found to predict the aforesaid quantities well at low residence times. The predictions of the model are in moderate agreement with data at high residence times. The theoretical approach by DeGraff and Poehlein behaves in the opposite way.

CHAPTER 1

INTRODUCTION

The 50 year old technique of emulsion polymerization is still being widely used in polymer industry. It often has significant advantages over homogeneous polymerization. Its final product is a latex and hence possesses the obvious advantage of direct application in cases like paints and coatings. Apart from this, there are distinct advantages in operating the reactor. The viscosity of a polymer latex in an emulsion polymerization reactor is low and hence the temperature control is relatively easy. In addition, a considerably high molecular weight product can be obtained at much higher reaction rates. For these reasons, emulsion polymerization has become commercially important in polymer industry. At present it is a major processing step in the manufacture of polymer products such as synthetic rubbers, paints, inks, coatings, adhesives, high impact strength co-polymers etc.

The commercial technique of emulsion polymerization can be divided into two classes: (1) batch and (2) continuous.

Batch systems are relatively simple and were developed first. These systems, although less efficient, are used for low production rates and for speciality applications. The advantages of this technique are the relatively simple equipment to be handled and the fact that a small amount

of material can be easily processed. This type of polymerization has been extensively studied in literature. Numerous articles are available in literature regarding the experimental and theoretical aspects of emulsion polymerization in batch. The more important of these works are by Smith and Ewart [1], Stockmayer [2], O'Toole [3], Gardon [4-10], Ugelstad [11], Medvedev [12], Min and Ray [13]. These and many other works have been comprehensively reviewed by Alexander and Nappel [14], Vander Hoff[15,16], Min and Ray [17].

As the annual output of the polymeric materials increases, more and more attention is being given to the use of continuous reactors. Although continuous stirred tank reactor (CSTR) is widely used commercially for emulsion polymerization, very little published information is available about these reactors. Experimental data is limited since the equipments are more costly and less flexible, experimental run times are longer and larger amounts of raw materials are consumed. Similarly the modelling of these reactors involves some conceptual problems which are not present in batch systems.

The earliest of the models developed for emulsion polymerization in a CSTR is by Greshberg and Longfield [18]. They have employed a residence time distribution model coupled with the Smith-Ewart kinetics. Omi and co-workers [19] and Nomura and co-workers [20] have employed Smith-Ewart kinetics for different reaction systems. They have also used

different models for initiation of particles. DeGraff and Poehlein [21] have employed Stockmayer's [2] theory coupled with residence time distribution for modelling a single stirred tank reactor. Recently population balance approach has been employed for modelling of continuous reactors.

Stevens and Funderburk [22] have employed a simple population balance approach to model a CSTR. Further work using this approach was done by Thompson and Stevens [23], Thompson and co-workers [24] and Cauley and Thompson [25]. These authors have used population balance equations to generate radical number distribution. They have used size independent particle growth rate. Cauley and Thompson [25] have further assumed instantaneous termination of radicals inside the particles. They have not developed material balance equations for various reactive species. The conversion and molecular weights of the polymer could not be calculated.

Min and Ray [17] have developed a comprehensive model for emulsion polymerization in batch and CSTR. They have taken into consideration a number of mechanisms like desorption of radicals from particles, chain transfer, initiation in aqueous phase, end-group stabilization of polymer particles, initiation by precipitation etc. They have, however, not presented any numerical result from this comprehensive model for CSTR.

The latest work on modelling of CSTR is by Kirilov and Ray [26]. They have used the general model of Min and Ray [17] for a specific case by making a number of simplifications. They have employed simple population balance equation for particle size distribution and have used Stockmayer's [2] relation for calculating the average number of radicals per particle. They have calculated average particle size and conversion obtained in the reactor. They did not calculate either the particle size distribution or the molecular weights of the polymer product.

In the present work, an attempt has been made to develop and solve modelling equations for emulsion polymerization in a CSTR. The population balance approach has been used. The population balance equations employed are more comprehensive. As a result, the molecular weight distribution of the resultant product can also be generated in addition to conversion and the particle size distribution.

CHAPTER 2

PHYSICAL MODEL AND POPULATION BALANCE EQUATIONS

Before starting the actual formulation of modelling equations, it is necessary to understand the physical picture and details of various mechanisms operating in an emulsion polymerization reactor. The currently accepted [14-17] picture is as follows:

A typical emulsion polymerization recipe consists of 100 parts of water, 50 parts of monomer, 2 parts of emulsifier, 0.1 part of initiator and small amounts of chain transfer agents and other additives. The soap is present in micelles, on the surface of monomer droplets and in the aqueous phase. The monomer is present mainly as droplets and a small amount (1 per cent) is solubilized in the micellar soap. A small portion of the monomer may also be dissolved in the aqueous phase. Generally, the initiator is present in the aqueous phase. In early stages of the reaction, the system consists of 3 types of particles dispersed in continuous aqueous phase: (1) emulsified monomer droplets of size $10000 \text{ } \text{\AA}$ and at a concentration of $10^{12}/\text{ml}$, (2) soap micelles with size $50-100 \text{ } \text{\AA}$ and at a concentration of $10^{18}/\text{ml}$ and (3) emulsifier stabilized monomer swollen polymer particles of size $500-1000 \text{ } \text{\AA}$, and at a concentration of $10^{14}-10^{15}/\text{ml}$ in later stages.

As the reaction proceeds the polymer particles grow in size, consuming the monomer present in the whole system, thus requiring more and more amount of soap to stabilize them. The soap is supplied by micelles which decrease in number and finally disappear. Similarly, the monomer droplets shrink in size as they supply monomer to the polymer particles and finally disappear, leaving the reaction system consisting of polymer particles dispersed in the aqueous phase.

Now some of the mechanisms operating in the reaction mixture will be discussed.

Nucleation of polymer particles is one of the least understood phenomena. It can occur in two different ways. The initiator present in the aqueous phase thermally decomposes into constituent radicals. (reaction temperature approximately $50-70^{\circ}\text{C}$) and these radicals grow by adding the monomer units from the monomer dissolved in the aqueous phase. During this process the radicals and the active polymer chains exhibit Brownian motion. During their motion they can collide with micelles, get absorbed into them and start the polymerization reaction thereby converting the micelles into polymer particles. They can also collide with and get absorbed in existing polymer particles. The radicals which are not captured by micelles or particles continue to add on monomer units and in the process decrease their

aqueous phase solubility. When a critical limit of chain size is reached they will precipitate and initiate new particles. Both these mechanisms of particle nucleation operate simultaneously. The former is more important for monomer scarcely soluble in water, while the latter is predominant for monomers which are relatively more soluble in water.

The particles thus initiated consist of monomer, polymer chains and the live polymer radicals. Polymerization continues in the particles, the monomer being supplied by monomer droplets. Radicals absorb into and desorb from the particles. Propagation, termination as well as chain transfer can take place inside particles. Soap required for stabilization is supplied by the micelles until they disappear. As more and more amount of monomer is consumed the monomer droplets decrease in size and number. Ultimately they disappear leaving only the polymer particles containing live radicals and dissolved monomer. As the reaction proceeds to higher range of conversion the monomer concentration in a particle decreases, viscosity of reaction medium increases and the reaction can become diffusion controlled. Gel effect may play an important role at higher conversion. Apart from these mechanisms, particles can also coagulate with each other to a limited extent. The extent of coagulation depends upon the particle size, amount of soap adsorbed on each particle, reaction temperature etc.

The polymer particles consist of monomer, polymer chains and live radicals as stated earlier. However, the exact structure of these particles is not yet known. The monomer and polymer may solubilize each other to form a homogeneous mixture. If that is the case, reaction will occur at a constant rate throughout the homogeneous particle. On the other hand, Williams [27] has postulated a non-uniform morphology for the particles, comprising of a polymer rich core and a monomer rich shell. The reaction is assumed to take place on the growing monomer polymer interface. Naper[28] has suggested exactly the opposite structure consisting of a monomer with core and a polymer rich shell.

This, in short, is a physical picture of the emulsion polymerization process. The aforesaid mechanisms operate simultaneously in a batch or continuous emulsion polymerization reactor. The objective of the present work is to use the knowledge of these basic mechanisms to model the characteristics of a polymer latex produced in a continuous stirred tank reactor.

The problem can be approached in two different ways:

In the residence time distribution approach the segregated flow model of a CSTR is employed. In this model the reactor is assumed to consist of a number of small independent reaction units. Each reaction unit is assumed to reside for different times inside the reactor. The reaction

units are assumed to operate as isolated batch reactors. The residence time distribution of different units combined with the knowledge of batch behaviour is believed to represent the CSTR characteristics. This method was most recently used by DeGraff and Poehlein [21] for the modelling of emulsion polymerization CSTR. Here a particle was identified as a reaction unit. They assumed the usual exponential residence time distribution. Another approach for modelling is the use of population balance equations which will be employed in this work. A short discussion about the theory of population balance equations is available in ref. 29. A brief introduction to this method is given below.

For any particulate system, a particle phase space consisting of a minimum number, say p , of independent co-ordinates, attached to a particle distribution, which completely describes the properties of the distribution, is postulated. The particle phase space may be divided into two subregions consisting of m internal and three external co-ordinates. External co-ordinates refer to the three spatial co-ordinates of the particle. Internal co-ordinates quantitatively represent the state of the particle independent of its spatial position.

Let us consider a $(m+3)$ dimensional particle distribution function $n(z, t)$ defined over region Z consisting of three external co-ordinates and m independent internal

co-ordinates. In a dynamic particulate system, individual particles change their position in the particle phase space. If the change of particle co-ordinates is gradual and continuous in any direction the rate of such 'motion' is called as convective particle velocity along that co-ordinate axis. Thus the particle phase-space velocity vector is defined as

$$\begin{aligned}\vec{v} &= v_x \delta_x + v_y \delta_y + \dots v_m \delta_m + v_1 \delta_1 + v_2 \delta_2 + v_3 \delta_3 \\ &= \vec{v}_{\text{int}} + \vec{v}_{\text{ext}}\end{aligned}\quad (2.1)$$

where v 's are the components of the particle velocity and δ 's are the unit vectors along the various co-ordinate axes. In general, the external velocity \vec{v}_{ext} can be calculated from the fluid velocity and momentum exchange relationship between fluid and particle. The internal particle velocity \vec{v}_{int} is assumed to be unique function of state of the particle.

There is also a possibility of particles suddenly appearing or disappearing in particle phase space along any co-ordinate. This possibility is represented by the birth and death functions of particle distribution. In general, these birth and death functions can be related to the state of the total system and the position of a particle in the particle phase space.

The population balance equation for particles in some fixed subregion Z_1 of particle phase space will be based on

the usual conservation law.

$$\text{Accumulation} = \text{Input} - \text{Output} + \text{net generation} \quad (2.2)$$

Consider the subregion to move convectively with particle phase space velocity \vec{v} . The population balance equation can be written as

$$\frac{d}{dt} \int_{z_1}^z n dz = \int_{z_1}^z (B-D) dz \quad (2.3)$$

where B and D represent birth and death functions.

Expanding L.H.S.

$$\begin{aligned} \frac{d}{dt} \int_{z_1}^z n dz &= \int_{z_1}^z \left(\frac{\partial n}{\partial t} dz + n \frac{d\vec{x}}{dt} \right) \\ &= \int_{z_1}^z \left[\frac{\partial n}{\partial t} + \vec{\nabla} \cdot \left(\frac{d\vec{x}}{dt} n \right) \right] dz \end{aligned}$$

where \vec{x} are the co-ordinates of system.

Replacing $\frac{d\vec{x}}{dt}$ by $(\vec{v}_{ext} + \vec{v}_{int})$ and combining with 2.3 gives the following expression:

$$\int_z \left[\frac{\partial n}{\partial t} + \vec{\nabla} \cdot (\vec{v}_{ext} \cdot n) + \vec{\nabla} \cdot (\vec{v}_{int} \cdot n) + D-B \right] dz = 0$$

The equation is valid for any arbitrary region Z_1 .

Hence the integrand vanishes for all z .

$$\text{Therefore } \frac{\partial n}{\partial t} + \vec{\nabla} \cdot (\vec{v}_{ext} \cdot n) + \vec{\nabla} \cdot (\vec{v}_{int} \cdot n) + D-B = 0 \quad (2.4)$$

This is a general expression which can be used to describe any particulate system, provided the number of

particles in the system is sufficiently large so that $n(z,t)$ can retain its statistical significance.

This equation will now be applied to a continuous well stirred particulate system of volume $V(t)$, having a population density function $n(z,t)$, p input and q output streams. Since in a well stirred system, tracing of external coordinates of particles is not of interest, this equation can be integrated over the range of external co-ordinates V or volume occupied by the system

$$\int_V \left(\frac{\partial n}{\partial t} + \vec{\nabla} \cdot \vec{v}_{int} \cdot n + D - B + \vec{\nabla} \cdot \vec{v}_{ext} \cdot n \right) dV = 0$$

$$\text{Therefore } V \left[\frac{\partial n}{\partial t} + \vec{\nabla} \cdot \vec{v}_{int} \cdot n + D - B \right] + \int_V \vec{\nabla} \cdot \vec{v}_{ext} \cdot n dV = 0 \quad (2.5)$$

The integral can be transformed into a surface integral of population flux over the surfaces of the system. The surfaces are: the cross-sections of input-output streams, free interface of liquid and total fluid-particle interface. The final expression is

$$\int \vec{\nabla} \cdot \vec{v}_{ext} \cdot n dV = n \frac{dV}{dt} - \sum_p Q_i n_i + \sum_q Q_o n \quad (2.6)$$

where Q 's are the flow rates of input-output streams.

Combining 2.5 and 2.6 and simplifying →

$$\frac{\partial n}{\partial t} + \vec{\nabla} \cdot \vec{v}_{int} \cdot n + n \frac{d(\log V)}{dt} = B - D + \sum_p \frac{Q_i n_i}{V} - \sum_q \frac{Q_o n}{V} \quad (2.7)$$

This expression represents the macroscopic population balance equation for a particulate, well mixed system. With the knowledge of input-output streams, expressions for velocity \vec{v}_i , birth and death functions this equation can be used to model emulsion polymerization in a CSTR mathematically. The values of population density $n(t)$ can be obtained and used to calculate the characteristics of the resultant polymer latex.

In next chapter, this equation will be used to develop the design equations for an emulsion polymerization CSTR.

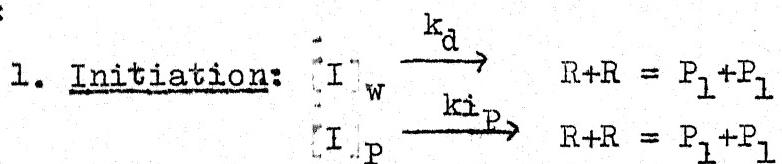
CHAPTER 3

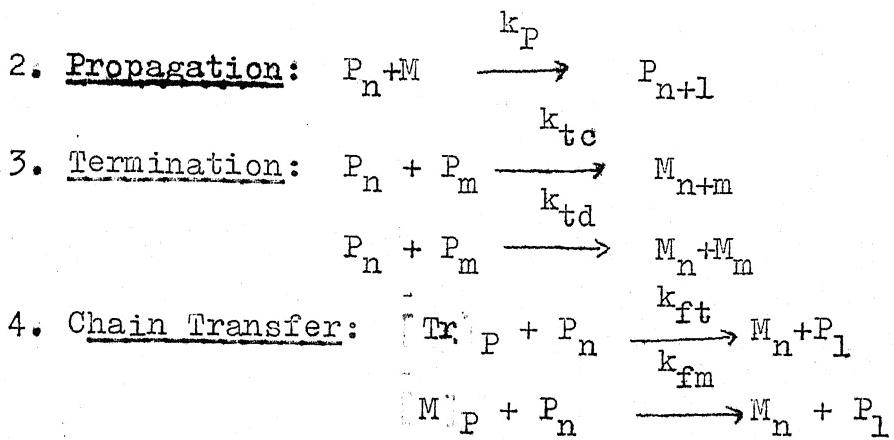
FORMULATION OF MODELLING EQUATIONS

From the previous chapters it can be seen that an emulsion polymerization latex is a particulate system comprising of a continuous aqueous phase and a particulate phase consisting of polymer particles, micelles and monomer droplets. A set of equations, consisting of material balance equations for continuous phase and population balance equations for particulate phase, completes the mathematical representation of physical polymerization system. Energy balance equations for the system are generally unnecessary because of low latex viscosity and high heat transfer rates, which result in an isothermal operation.

In the following pages development of the modelling equations will be given. This presentation closely follows the approach given by Min and Ray [17]. For convenience, nomenclature for this chapter is provided at the end of the chapter.

Kinetic Scheme: The usual kinetic scheme involving initiation, propagation, termination as well as chain transfer to monomer and transfer agents has been used. The scheme is given below:





Particulate Phase:

For characterizing a polymer latex, the polymer particles are of prime importance. Very little, if any, reaction takes place in the monomer droplets or micelles. Hence it is not necessary to develop separate population balance equations for these species. It will be assumed that these distributions can be represented by average micelle size (volume) v_m and average droplet size (volume) v_d respectively.

With the polymer recipe and average residence time of the reactor usually employed in practice, the number density of polymer particles is of the order of $10^{14}/\text{cm}^3$ emulsion. Hence we can attach an appropriate distribution function to the particles and use macroscopic population balance equation as given in the last chapter.

$$\frac{\partial n}{\partial t} + \vec{\nabla} \cdot \vec{v}_{int} \cdot n + n \frac{d}{dt}(\log V) = B - D + \sum_p \frac{Q_i n_i}{V} - \sum_q \frac{Q n}{V} \quad (2.7)$$

Consider a reactor of volume V , with an input and output stream, both having a volumetric flow rate of q . The

The contents of reactor are assumed to be completely mixed. Let r_i represent the net generation term (B-D) resulting in discrete changes of variables due to the various mechanisms. Let n be the distribution function under consideration. The aforesaid equation, when applied to such system becomes

$$\frac{\partial n}{\partial t} + \vec{V} \cdot \vec{v}_{int} \cdot n = \sum_{i=1}^m r_i + \frac{1}{\theta} n \quad \left. - \frac{1}{\theta} n \right|_{feed} \quad (3.1)$$

where θ represents the mean residence time and m depends upon the number of mechanisms to be included in the model.

Radical Number Distribution:

Consider a distribution function $f(i, v, t)$ which represents number of particles per cm^3 of emulsion, having volume v to $v+dv$ and i radicals, at any time t . The knowledge of this distribution is necessary to calculate the rate of polymerization and \bar{i}_{avg} . The population balance equation corresponding to this function is

$$\frac{\partial f(i, v, t)}{\partial t} + \frac{\partial}{\partial v} \cdot (r_v \cdot f(i, v, t)) = \sum_{i=1}^m r_i + \frac{1}{\theta} f(i, v, t) \quad \left. - \frac{1}{\theta} f(i, v, t) \right|_{feed} \quad (3.2)$$

First term represents rate of change of $f(i, v, t)$ with respect to time. r_v in the second term represents the rate of change of volume of particle with respect to time, [ie $\frac{dv}{dt}$].

The size of a particle varies due to the propagation reaction taking place inside the particle and due to particle swelling. This quantity will depend upon the model assumed for particle structure. For homogeneous particle morphology Gardon [4] has calculated this quantity by correlating the propagation mechanism with the monomer concentration and polymer and monomer densities. He obtained following relationship for the conditions when the monomer droplets are present in the system and the aqueous phase is saturated (with monomer).

$$\frac{dv}{dt} = r_v = \frac{k_p}{N_A} \cdot \frac{d_m}{d_p} \cdot \frac{\bar{\varphi}}{(1-\bar{\varphi})} \cdot i \quad (3.3)$$

The rate terms r_i in the RHS of 3.2 depend upon the particular mechanisms to be taken into consideration. These terms will be discussed below.

First term, r_1 represents the change in $f(i, v, t)$ due to absorption of radicals into the particles. Radicals can either propagate and precipitate as oligomers in the aqueous phase or can be absorbed into the particles. The relative importance of these two mechanisms depends upon the monomer solubility and the propagation rate constant in the aqueous phase. Gardon [4] has obtained an expression for rate of radical absorption by computing the probability of collision of a radical with micelles and particles. The final expression is

$$R_{ia} = \frac{2 f k_d [I]_w}{A_m + A_p} \quad \text{when } s \gg 4/L$$

$$= \frac{2 f k_d [I]_w}{(4/L)} \quad \text{when } s \ll 4/L$$

where L represents the average length travelled by a growing chain before it reaches the critical length at which it precipitates and is given by

$$L = (2D t_{prec})^{0.5}$$

Here t_{prec} represents the time required by radical to reach critical chain length for precipitation and is a function of the propagation rate constant and monomer concentration in the aqueous phase. D is the diffusion coefficient of radicals in the aqueous phase.

Rate r_1 in eqn.3.2 is then given by the following expression

$$r_1 = R_{ia} \times a_p N_A (f(i-l, v, t) - f(i, v, t)) \quad (3.4)$$

where the first term represents the formation of $f(i, v, t)$ by absorption of a radical into $f(i-l, v, t)$ type of particles and the second term represents destruction of $f(i, v, t)$ type of particle by absorption of a radical, thus transforming it into a $f(i+l, v, t)$ type particle.

Second rate term (r_2) represents the effect of formation of a radical pair inside the particle by the decomposition of monomer-soluble initiator.

$$r_2 = f k_{ip} [I]_P \cdot N_A \cdot v (f(i+2, v, t) - f(i, v, t)) \quad (3.5)$$

Radicals can also desorb from the particle. This effect can be accounted for by the following expression:

$$r_3 = \frac{k_o a_p}{v} [(i+1) f(i+1, v, t) - i f(i, v, t)] \quad (3.6)$$

First term represents the formation of $f(i, v, t)$ type particle by radical desorption from a $f(i+1, v, t)$ type particle and second term represents the destruction of $f(i, v, t)$ by desorption of a radical.

Rate of change of $f(i, v, t)$ by termination of two radicals inside the particle, is given by the following expression:

$$r_4 = \frac{k_{tc} + k_{td}}{2} \left(\frac{1}{N_A v} \right) [(i+2)(i+1) f(i+2, v, t) - (i)(i-1)f(i, v, t)] \quad (3.7)$$

The rate of formation of new particles by radical absorption in micelles is given by

$$r_5 = R_i \times \frac{a_m \cdot m}{A_m + A_p} \delta_{il} \delta(v - v_m) \quad (3.8)$$

δ functions on right hand side of the expression indicate that new particles, thus formed, will have volume v_m and will contain one radical.

Particle Size Distribution:

Let $F(v, t)$ be the particle size distribution function. This distribution is quite important since the particle size is one of the characterizing properties of a polymer latex.

Consider the balance equation for $F(v, t)$, which represents the number of particles having volume between v and $v+dv$ at any time t . It is related to radical number distribution by the following equation:

$$F(v, t) = \sum_{i=0}^{\infty} f(i, v, t) \quad (3.9)$$

By combining 3.2 to 3.9 the population balance equation for $F(v, t)$ can be derived to be

$$\frac{\partial F(v, t)}{\partial t} + \frac{\partial}{\partial v} (r_v F(v, t)) =$$

Rate of change of $F(v, t)$ with respect to time	+	Rate of change of $F(v, t)$ due to volume growth of particles
$R_i \frac{A_m}{A_m + A_p} \cdot \delta(v - v_m)$	+	$\frac{1}{\theta} [F(v, t) - F(v, t)]_{\text{feed}}$
Rate of change of $F(v, t)$ due to generation of new particles of size v_m	+	Change of $F(v, t)$ due to input and output

(3.10)

where

$$r_v = \frac{k_p}{N_A} \cdot \frac{d_m}{d_p} \cdot \frac{\phi}{(1-\phi)} \cdot i_{\text{avg}} \quad (3.11)$$

and i_{avg} [i.e. average number of radicals per particle] is given by

$$i_{\text{avg}} = \frac{\int_{v_m}^{\infty} \sum_{i=0}^{\infty} i f(i, v, t) dv}{\int_{v_m}^{\infty} \sum_{i=0}^{\infty} f(i, v, t) dv} \quad (3.12)$$

Live Polymer Chain Length Distribution:

In order to calculate the molecular weight distribution the chain length distribution of the polymer radicals present inside the particle is needed. This distribution is used to solve the dead polymer chain length distribution which is in turn used to obtain the m.w.d. of the polymer product.

Consider the particles having volume between v and $v+dv$ and i radicals at any time t . These i radicals will in general have different chain lengths. Amongst these radicals consider the ones having chain length n . Let $L(n,i,v,t)$ represent the total number of radicals of chain length n present in all the particles having volume between v and $v+dv$ and having i radicals, at any time t . The population balance for such radicals is given by the following expression.

$$\frac{\partial L(n,i,v,t)}{\partial t} + \frac{\partial}{\partial v}(r_v \cdot L(n,i,v,t)) = \sum_{i=1}^k r_i \\ + \frac{1}{Q} [L(n,i,v,t) \left. \right|_{\text{feed}} - L(n,i,v,t)] \quad (3.13)$$

First term represents change in $L(n,i,v,t)$ with respect to time. The second term represents change in $L(n,i,v,t)$ due to change in size of a particle, where r_v represents the rate of change of volume with respect to time.

The rate terms r_i on right hand side, as before, depend on specific assumptions made. Firstly consider the effect of absorption of radicals. It has been assumed that

only primary radicals can absorb into the particle and that the particle nucleation by precipitation in aqueous phase has been negligible. Let $g(n,i,v,t)$ represent number of radicals of chain length n in a $f(i,v,t)$ type particle. When a radical is absorbed in a $f(i-1,v,t)$ type particle it converts the particle into a $f(i,v,t)$ type, making in the process, a contribution of $g(n,i-1,v,t)$ radicals to $L(n,i,v,t)$. Similarly when a primary radical is absorbed in a $f(i,v,t)$ type particle $g(n,i,v,t)$ number of radicals will be lost from $L(n,i,v,t)$. When a primary radical is absorbed in a particle it adds only to $L(1,i,v,t)$. Thus the total effect, per cm^3 of emulsion can be represented as:

$$\begin{aligned}
 r_1 &= R_i \frac{a_p}{A_p + A_m} \cdot f(i-1,v,t) \cdot g(n,i-1,v,t) \\
 &\quad - R_i \frac{A_p}{A_p + A_m} f(i,v,t) \cdot g(n,i,v,t) \\
 &\quad + R_i \frac{A_p}{A_p + A_m} f(i-1,v,t) \delta n_1 \\
 &= R_i \left[\frac{a_p}{A_p + A_m} \right] [f(n,i-1,v,t) - f(n,i,v,t) + f(i-1,v,t) \delta n_1]
 \end{aligned} \tag{3.14}$$

From the definition

$$g(n,i,v,t) = \frac{L(n,i,v,t)}{f(i,v,t)} \tag{3.15}$$

The initiator dissolved in a polymer particle can decompose to give a pair of radicals of chain length one each. A contribution of $g(n,1-2,v,t)$ is made to $L(n,i,v,t)$

per generation of a pair due to such an event. In addition to this the radicals formed themselves contribute to addition of radicals with chain length one. The same process occurring in a $f(i, v, t)$ type particle would result in a loss to $L(n, i, v, t)$

$$\begin{aligned} r_2 &= k_{ip} [I]_P \cdot v \cdot N_A [(g(n, i-2, v, t) + 2\delta n_1) f(i-2, v, t) \\ &\quad - g(n, i, v, t) \cdot f(i, v, t)] \\ &= k_{ip} [I]_P \cdot v \cdot N_A [L(n, i-2, v, t) - L(n, i, v, t) + 2\delta n_1 f(i-2, v, t)] \end{aligned} \quad (3.16)$$

The effect of radical desorption from particles is given by

$$\begin{aligned} r_3 &= k_o \frac{a_p}{v} [(i+1) f(i+1, v, t) \cdot g(n, i+1, v, t) (1 - \frac{1}{i+1}) \\ &\quad - i f(i, v, t) g(n, i, v, t)] \\ &= \frac{k_o a_p}{v} [(i) L(n, i+1, v, t) - (i) L(n, i, v, t)] \end{aligned} \quad (3.17)$$

Here the first term represents the production of $L(n, i, v, t)$ by desorption of radicals from $f(i+1, v, t)$ except when a radical of length n desorbs. The second term represents loss due to desorption from radicals from $f(i, v, t)$ type particles.

The effect of the kinetic steps inside the polymer particle depends upon the structure of particle. Assuming homogeneous particle morphology, the effect of propagation and transfer reaction can be expressed as

$$\begin{aligned}
r_4 &= k_p[M]_P f(i, v, t) \cdot g(n-1, i, v, t) - k_p[M]_P f(i, v, t) g(n, i, v, t) \\
&\quad - k_{ft}[Tr]_P f(i, v, t) g(n, i, v, t) + k_{ft}[Tr]_P \cdot i f(i, v, t) \delta n_1 \\
&\quad - k_{fm}[M]_P f(i, v, t) g(n, i, v, t) + k_{fm}[M]_P i f(i, v, t) \delta n_1 \\
&= k_p[M]_P [L(n-1, i, v, t) - L(n, i, v, t)] + k_{ft}[Tr]_P [-L(n, i, v, t) \\
&\quad + f(i, v, t) \cdot i \cdot \delta n_1] + k_{fm}[M]_P [f(i, v, t) \cdot i \cdot \delta n_1 - L(n, i, v, t)] \\
&\tag{3.18}
\end{aligned}$$

The first two terms represent the effect of propagation; third and fourth term represent chain transfer to transfer agent and the fifth and sixth term represent chain transfer to monomer. The chain transfer reactions involve addition of one chain to $f(l, i, v, t)$. This effect has been accounted for in fourth and sixth term.

The effect of termination reactions can be written as

$$\begin{aligned}
r_5 &= \frac{k_{tc} + k_{td}}{N_A v} \left[\frac{(i+2)(i+1)}{2} - (i+1) \right] \cdot L(n, i+2, v, t) \\
&\quad - \frac{(i)(i-1)}{2} L(n, i, v, t) \\
&= \frac{k_{tc} + k_{td}}{N_A v} \left[\frac{(i+1)(i)}{2} L(n, i+2, v, t) - \frac{(i)(i-1)}{2} L(n, i, v, t) \right] \\
&\tag{3.19}
\end{aligned}$$

First and third terms in the initial expression represent change due to termination between any two radicals. Third term represents the termination between a radical of chain length n and remaining $(i+1)$ radicals. The possibility

of termination between two radicals of same chain length n , which will be a second order term, has been neglected.

The effect due to radical absorption in micelles is given by the following expression:

$$\begin{aligned} r_6 &= R_i \cdot \frac{a_m}{(A_p + A_m)} \cdot m \cdot l \cdot \delta n_l \cdot \delta(v - v_m) \cdot \delta_{il} \\ &= R_i \cdot \frac{A_m}{(A_p + A_m)} \cdot \delta n_l \cdot \delta(v - v_m) \cdot \delta_{il} \end{aligned} \quad (3.20)$$

Dead Polymer Chain Length Distribution:

This distribution is related to the chain length of dead polymer molecules inside the particles and is thus closely connected with m.w.d. of the resultant polymer latex. Consider the particles having volume between v and $v+dv$ and possessing i active radicals at any time t . Apart from these i active radicals, each particle contains some dead polymer chains. Let $G(n, i, v, t)$ represent total number of these dead chains contained in all $f(i, v, t)$ type particles at any time t . The population balance equation for this function can be written as

$$\begin{aligned} \frac{\partial G(n, i, v, t)}{\partial t} + \frac{\partial r_v \cdot G(n, i, v, t)}{\partial v} &= \sum_{i=1}^k r_i \\ + \frac{1}{v} [G(n, i, v, t) - G_{\text{feed}}(n, i, v, t)] \end{aligned} \quad (3.21)$$

First term represents the change in $G(n, i, v, t)$ with respect to time and the second term expresses the change due to change in volume of particle.

Rate terms on the right hand side depend upon the particular mechanisms to be considered in the model. Derivation of these terms is similar to that for corresponding terms in live polymer chain length distribution. These terms will be discussed briefly.

The first rate term r_1 represents the effect of radical absorption on $G(n,i,v,t)$. Let $g'(n,i,v,t)$ represent number of dead polymer chains per particle. Hence $G(n,i,v,t) = g'(n,i,v,t) f(i,v,t)$. When a new radical enters a polymer particle of type $f(i-1,v,t)$, it converts that particle into $f(i,v,t)$ type particle and a contribution of $g'(n,i-1,v,t)$ is made towards $G(n,i,v,t)$. Hence the total contribution of this mechanism towards change in $G(n,i,v,t)$ is given by

$$r_1 = R_i \frac{a_p}{(A_m + A_p)} f(i-1,v,t) g'(n,i-1,v,t) - R_i \frac{a_p}{(A_m + A_p)}$$

$$= R_i \frac{a_p}{(A_m + A_p)} [G(n-1,i,v,t) - G(n,i,v,t)] \quad (3.22)$$

Similarly the effect of radical initiation in particles and desorption from particles can be shown to be given by following expression. It is assumed here that dead polymer chains can not desorb from particles.

$$r_2 = f_{kd} \cdot [I]_p [G(n,i-2,v,t) - G(n,i,v,t)] \quad (3.23)$$

$$r_3 = \frac{k_o a_p}{v} [(i+1) G(n,i+1,v,t) - i G(n,i,v,t)] \quad (3.24)$$

The effect of the kinetic steps is more important in case of $G(n,i,v,t)$ and can be given by following expression:

$$\begin{aligned}
 r_4 = & k_{ft} [Tr]_P L(n,i,v,t) + k_{fm} [M]_P L(n,i,v,t) \\
 & + \frac{k_{tc} + k_{td}}{2} [(i+2)(i+1) G(n,i+2,v,t) - (i)(i-1) G(n,i,v,t)] \\
 & + \frac{k_{tc}}{2v} [f(i+2,v,t) \sum_{m=1}^{n-1} g(m,i+2,v,t) g(n-m,i+2,v,t)] \\
 & + \frac{k_{td}}{2v} [g(n,i+2,v,t) f(i+2,v,t)]
 \end{aligned} \quad (3.25)$$

In this expression the first two terms represent formation of dead polymer chains due to chain transfer with transfer agent and with monomer. Third term expresses the change in $G(n,i,v,t)$ by termination of any two radicals in $f(i+2,v,t)$ type particle (i.e. formation) and in $f(i,v,t)$ type particle (destruction). Fourth term represents the additional contribution when two terminating radicals form a polymer chain of length n. Last term represent the possibility of disproportionation involving a live polymer radical of chain length n.

Apart from these population balance equations, balances which define the environment of particles must also be written. These equations include material balances for various reactive species and balances for number of droplets, micelles etc.

Emulsifier Balances:

The total amount of soap present per cm^3 emulsion is given by following expression:

$$\frac{d(s)_T}{dt} = \frac{1}{\theta} [s]_T \left|_{\text{feed}} - [s]_T \right| \quad (3.26)$$

This soap may only be present in aqueous phase as micelles or may be absorbed on the particles and droplets. The process of transfer of emulsifier will be assumed to be extremely fast. Under these conditions the balance for number of micelles yields

$$\frac{m_a m}{CE} = [s]_T - s_{wc} - \frac{A_p}{CE} - \frac{m_d v_d}{CE} \quad (3.27)$$

The terms on right hand side represent total soap, the soap present in the aqueous phase, the soap adsorbed on particles and the soap adsorbed on the monomer droplet respectively.

Total Monomer Balance:

The balance for the total amount of monomer present in the reactor is given below:

$$\begin{aligned} \frac{d V_R [M]_T}{dt} &= \frac{V_R}{\theta} [M]_T \left|_{\text{feed}} - [M]_T \right| \\ &- V_R k_p [M]_P \left\{ \sum_{i=0}^{\infty} i f(i, v, t) dv \right\} \quad (3.28) \end{aligned}$$

The first two terms represent the effect due to input and output terms. The third term represents the loss of monomer due to propagation in particles. Here the small contributions due to chain transfer and propagation in aqueous phase are neglected.

Monomer Droplets Balance:

Monomer droplets balance can be written as follows. As in the previous case the dynamic of transfer of monomers will be assumed to be extremely fast.

$$m_d v_d = [M]_T V_{mol} - \int_0^\infty v [M]_P V_{mol} \cdot F(v) \cdot dv \\ = V_{mol} m \cdot [M]_m V_{mol} \quad (3.29)$$

The terms on right hand side of equation represent total monomer content, monomer in particles, and monomer in micelles.

Aqueous Phase Balances:

The balance for concentration of the initiator in aqueous phase is given below

$$\frac{d V_w [I]_w}{dt} = - V_w k_d [I]_w + \frac{V_w}{Q} \left[[I]_w \right]_{feed} - [I]_w \quad (3.30)$$

Lastly an expression for the monomer concentration in polymer (i.e. $[M]_P$ or $\bar{\phi}$) is required. The nature of the equation will, in general, depend upon the model for particle structure considered. For a homogeneous particle morphology the free energy of mixing of polymer can be equated to the surface free energy [4]. The resultant expression for $\bar{\phi}$ is:

$$(1-\bar{\phi}) + \ln(\bar{\phi}) + \frac{2\gamma V_{mol}}{RT r} (1-\bar{\phi})^2 = \frac{2\gamma V_{mol}}{RT r} \quad (3.31)$$

So far the population balance equations for particulate phase and the material balance for aqueous phase have been

discussed. These balances are assumed to represent the given system completely. These equations can be applied to a practical system and the results can be used to obtain desired characteristics of the resultant polymer latex.

NOMENCLATURE

CE	surface coverage by the emulsifier, cm^2/mole
a_m	surface area of micelles, cm^2
a_p	surface area of particle of size v , cm^2
A_m	total surface area of micelles, cm^2/cm^3 emulsion
A_p	total surface area of particles, cm^2/cm^3 emulsion
d_p	density of polymer, gms/cm^3
d_m	density of monomer, gms/cm^3
f	efficiency of initiator decomposition, dimensionless
$f(i, v, t)$	number of particles containing i radicals and possessing volume between v to $v+dv$ at time t , moles/cm^3 emulsion
$F(v, t)$	number of particles having volume between v and $v+dv$ at time t , moles/cm^3 emulsion
$g(n, i, v, t)$	number of live chains of chain length n present in a particle containing i radicals and possessing volume between v to $v+dv$ at time t , dimensionless
$g'(n, i, v, t)$	number of dead chains of chain length n present in particle containing i radicals and possessing volume between v to $v+dv$ at time t , dimensionless
$G(n, i, v, t)$	number of dead chains of chain length n present in all the particles having i radicals and possessing volume v to $v+dv$, at time t , moles/cm^3 emulsion
i	number of radicals in a particle, dimensionless
\bar{i}_{avg}	average number of radicals per particle, dimensionless
$[I]_w$	initiator concentration in aqueous phase, moles/cm^3 , water
$[I]_p$	initiator concentration in polymer particles, moles/cm^3

k_{ip}	rate constant for dissociation of initiator in polymer phase, l/cms
k_d	rate constant for dissociation of initiator in aqueous phase, l/cms
k_p	rate constant for propagation, $\text{cm}^3/\text{mole. sec.}$
k_{tc}	rate constant for termination by combination, $\text{cm}^3/\text{mole.sec.}$
k_{td}	rate constant for termination by disproportionation, $\text{cm}^3/\text{mole sec}$
k_{ft}	rate constant for chain transfer to transfer agent, $\text{cm}^3/\text{mole sec.}$
f_{fm}	rate constant for chain transfer to monomer, $\text{cm}^3/\text{mole sec.}$
k_o	rate constant for radical desorption, cms/sec
$L(n,i,v,t)$	number of live chains of chain length n in all the particles having i radicals and possessing volume between v and $v+dv$ at time, t , moles/ cm^3 emulsion
m	number of micelles, moles/ cm^3 emulsion
$[M]_P$	concentration of monomer in the particles, moles/ cm^3
$[M]_m$	concentration of monomer in the micelles, moles/ cm^3
$[M]_T$	total concentration of monomer in the reactor, moles/ cm^3 emulsion
n	chain length of polymer, dimensionless
r	radius of particle, cms
R_i	rate of initiation, moles/sec.
R_{ia}	rate of absorption of radicals in particles, and micelles, moles/sec. $\cdot \text{cm}^2$

$[S]_T$	total emulsifier concentration, moles/cm ³ emulsion
S_{wc}	critical miscelle concentration, moles/cm ³ water
t	time, sec.
$[Tr]_P$	concentration of chain transfer agent inside the particle, moles/cm ³
v	volume of particle, cm ³
v_m	volume of micelle, cm ³
v_d	volume of monomer droplets, cm ³
V_R	volume of reactor, cm ³
V_w	volume of aqueous phase per cm ³ emulsion, cm ³ /cm ³ emulsion
ϕ	monomer volume fraction in the particles, dimensionless
ψ	monomer polymer interaction parameter, dimensionless
γ	surface tension, dynes/cms
θ	mean residence time, sec.
δ_{ig}	Kronecker delta function, 1 for $i=J$, 0 for $i \neq J$
$\delta(v-v_m)$	Dirac delta function; 1 for $v=v_m$, 0 for $v \neq v_m$.

CHAPTER 4

METHOD OF SOLUTION OF THE MODELLING EQUATIONS

In the previous chapter, the modelling equations for emulsion polymerization in a CSTR were presented. These equations contain various material balance equations and the population balance equations. The material balance equations are ordinary differential or nonlinear algebraic equations and are relatively easy to handle. However, the various population balance equations involve different distribution functions like $F(v)$, $f(i,v)$, $L(n,i,v)$ and $G(n,i,v)$ and are complex coupled partial differential-difference equations. Special care is required to decide the line of approach to solve these equations. In this chapter normalisation procedures have been introduced and for convenience the additional nomenclature for this chapter is listed at the end of the chapter.

The aim of solving these equations is to calculate the values of the aforesaid distribution functions. However, generation of these functions numerically for different values of attributes may take exorbitant time. To avoid this, the problem can be approached by making use of the concept of moments of a distribution,

Let $f(x)$ represent a continuous distribution for the attribute x . Then the k th moment of $f(x)$ is defined as

$$\bar{f}_k = \int_0^\infty x^k f(x) dx \quad (4.1)$$

Similarly if $H(i)$ represents a distribution defined only for discrete values of i , then the k th moment of the discrete distribution is given by

$$\bar{H}_k = \sum_{i=i \text{ min}}^{i=i \text{ max}} i^k H(i) \quad (4.2)$$

It can be seen from these definition that, with the knowledge of a distribution function, all of its moments can be calculated. Obviously, a continuous distribution function, say $f(x)$, containing infinite information cannot be exactly specified by a finite set of moments. However, it is possible to represent a distribution function, in terms of finite number of moments to an accuracy sufficient for modelling purposes. One of the methods to achieve this is to represent the distribution function in a series expansion of orthogonal polynomials like Laguerre polynomial [17, 30]. The Laguerre polynomial $L_n^\alpha(x)$ has the following orthogonal property.

$$\int_0^\infty (x)^\alpha \exp(x) L_n^\alpha(x) L_m^\alpha(x) dx = \underbrace{\frac{\alpha + n + 1}{n}}_{\delta_{nm}} \delta_{nm} \quad (4.3)$$

$$\text{where } \delta_{nm} = 1 \quad \text{if } n=m \quad (4.4)$$

$$= 0 \quad \text{if } n \neq m$$

$$\text{where } L_n^\alpha(x) = \sum_{j=0}^n (-1)^j \binom{n+\alpha}{n-j} \frac{x^j}{j!} \quad (4.5)$$

Using the definition and orthogonal property of Laguerre polynomial, as stated above, a normalized distribution function $G(y)$ can be represented in terms of summations involving Laguerre polynomial [40].

$$G(y) = \frac{y^\alpha \exp(-y)}{\Gamma(\alpha+1)} \sum_{n=0}^m c_n^\alpha L_n^\alpha(y) \quad (4.6)$$

The coefficients c_n^α are related to the moments \bar{G}_k of the normalized distribution function $G(y)$ by following expression:

$$c_n^\alpha = \sum_{j=0}^n (-1)^j \frac{\underline{L}_n}{\underline{n-j}} \frac{\underline{j}}{\underline{j}} \frac{\underline{\alpha+1}}{\underline{\alpha+j+1}} \bar{G}_j \quad (4.7)$$

Equation 4.7 follows from the orthogonal properties of L_n^α . Following Min [30], the parameter α is chosen so that c_1^α and c_2^α are zero, which results in rapid convergence. The resulting expression for α is the following:

$$\alpha = -1 + \frac{(\bar{G}_1)^2}{\bar{G}_2 - (\bar{G}_1)^2} \quad (4.8)$$

The normalized function $G(y)$, used here is related to the absolute distribution function $f(x)$ as follows:

$$f(x)dx = G(y) dy \quad \text{and} \quad y = \frac{\bar{f}_1}{\bar{f}_0(\alpha+1)} \quad (4.9)$$

The moments of the two distributions, $f(x)$ and $G(y)$, are related to each other by the following expression

$$\bar{f}_k = \bar{f}_0 \left[\frac{\bar{f}_1}{\bar{f}_0 (\alpha+1)} \right]^{k-1} = G_k \quad (4.10)$$

It can be seen from the preceding lines that most of the information contained in a continuous distribution function can be obtained from a finite set of moments of that distribution. The number of moments required to specify a differential distribution depends on the complexity of its shape. It has been found that only 3 or 4 moments are often sufficient to represent a unimodal distribution satisfactorily.

Even in practical cases, it is often impossible to measure complete molecular weight distribution or particle size distribution. Generally it is convenient to measure and use the leading moments of these distributions for quality control. This method involves the use of M_n , M_w or polydispersity index to represent m.w.d. and average size to specify the particle size distribution.

For the reason stated above, the modelling equations given in the previous chapter involving differential distribution functions will be transformed to a set of equations involving finite number of moments of those distributions.

It should be noted that the method to represent a distribution in terms of its moments is useful only for continuous distributions. Obviously a discrete distribution containing a large number of closely spaced points can be approximated by a continuous distribution. Consequently, in

in the present study the live polymer chain distribution as well as the dead polymer chain length distributions which are actually defined for integer values of n only, are approximated by a continuous distribution in n and the aforesaid method is then made use of.

However, a distribution of polymer particles in i (the number of radicals contained in the particle) is defined only at a few points. It has been found from experiments that the number of live radicals contained in a particle rarely exceeds 10. As a result, the use of moments to represent the discrete distribution in i is not justifiable. Hence this distribution will be generated numerically at the discrete points at which it actually exists (i.e. at $i = 0, 1, 2, \dots$). Since the number of points at which it is defined is small this technique involves the solution of a relatively small number of equations.

To start with, the following assumptions will be made for mathematical simplicity.

1. A distribution in the volume of particles is generated. The fractional moments in volume (i.e. two third and one third moment) are used to calculate the total surface area and the average diameter of the particles. However, when 'v' is connected with the distribution functions in a more complex way (like the term arising due to termination of radicals in particle), the average value of the volume

of particles is used for mathematical simplicity. This approximation is likely to be more exact when the particle size distribution is relatively narrow.

2. The terms representing the chain transfer to transfer agent and initiation in particles will be maintained while developing the sets of equations involving moments of live and dead polymer chain length distribution. However, the values of $[\bar{Tr}]_P$ and $[\bar{I}]_P$ can be evaluated by addition of two material balance equations and thus these mechanisms can be easily accounted for.

3. The value of monomer concentration in particles ($\bar{\phi}$) can be calculated from equation 3.30. This equation shows that ($\bar{\phi}$) is in general a function of particle size. However, from experimental studies it has been found that the concentration of monomer in a particle is almost constant and independent of its volume. The values of the concentrations of monomer in a polymer particle are reported in literature for different monomers. Hence, these values can be directly used to solve remaining modelling equations. Most of the results from this study have been obtained by this method. However, the results obtained without making this assumption are also given for comparison.

The modelling equations given in the earlier chapters are not dimensionless. Hence the numerical values of various variables depend upon the system of units to be used. These

equations should firstly be transformed into sets of nondimensional modelling equations. This approach facilitates the use of these equations, since the variables are not confined to any particular system of units. Similarly it facilitates the numerical solution of these equations, since the range of numerical values of various dimensionless variables is relatively narrow. The dimensionless population balance equations should then be transformed into the sets of equations involving moments of the dimensionless distributions.

This procedure has been explained in the following lines, for the case of live polymer chain length distribution.

Consider the population balance equation (3.10) involving live polymer chain length distribution:

$$\begin{aligned}
 & \frac{\partial L(n,i,v,t)}{\partial t} + \frac{\partial}{\partial v} (r_v \cdot L(n,i,v,t)) = R_{ia} \left[\frac{a_p}{A_p + A_m} \right] \\
 & [L(n,i-1,v,t) - L(n,i,v,t) + f(i-1,v,t) \cdot \delta_{nl}] \\
 & + k_d [I]_P v \cdot N_A [L(n,i-2,v,t) - L(n,i,v,t) + 2\delta_{nl} \cdot f(i-2,v,t)] \\
 & + \frac{k_o}{v} \left[(i) L(n,i+1,v,t) - (i) L(n,i,v,t) \right] + k_p [M]_P \\
 & [L(n-1,i,v,t) - L(n,i,v,t)] + k_{ft} [Tr]_P [i f(i,v,t) \delta_{nl} \\
 & - L(n,i,v,t)] + k_{fm} [M]_P [i f(i,v,t) \delta_{nl} - L(n,i,v,t)] \\
 & + \frac{k_{tc} + k_{td}}{N_A v} \left[\frac{(i+1)(i)}{2} L(n,i+2,v,t) - \frac{(i)(i-1)}{2} L(n,i,v,t) \right] \\
 & + \frac{1}{\theta} \left[L(n,i,v,t) \underset{\text{feed}}{\uparrow} - L(n,i,v,t) \right] + R_{ia} \frac{A_m}{(A_p + A_m)} \delta_{nl} \delta_{il} \delta(v - v_m)
 \end{aligned} \tag{3.10}$$

At steady state the derivatives with respect to time can be equated to zero.

Similarly all the particles are assumed to have volume v_{avg} . Hence the derivatives with respect to v also reduce to zero, and the average volume of the particles v_{avg} can then be substituted in place of volume, v , in these equations. Using the definitions of various dimensionless variables and dimensionless constants given at the end of this chapter and after further simplifications this equation reduces to

$$\begin{aligned}
 0 = & C_7 [\bar{I}]_w \frac{(\bar{v})^{2/3}}{\bar{F}^{2/3} + \bar{m}} [\bar{L}(n, i-1) - \bar{L}(n, i) + \bar{f}(i-1) \delta_{nl}] \\
 & + C_2 [\bar{I}]_P \cdot v [\bar{L}(n, i-2) - \bar{L}(n, i) + 2\delta_{nl} \bar{f}(i-2)] \\
 & + C_3 \cdot (\bar{v})^{1/3} [i \bar{L}(n, i+1) - i \bar{L}(n, i)] \\
 & + C_{101} (\bar{\varphi}) [\bar{L}(n-1, i) - \bar{L}(n, i)] \\
 & + C_{63} [\bar{T}_r]_P [\delta_{nl} \cdot (i) \bar{f}(i) - \bar{L}(n, i)] \\
 & + C_{102} (\bar{\varphi}) [\delta_{nl} \cdot i \bar{f}(i) - \bar{L}(n, i)] + \frac{C_4}{(\bar{v})} [(i) (i+1) \\
 & \quad \bar{L}(n, i+2) - (i)(i-1) \bar{L}(n, i)] + \frac{1}{\bar{\theta}} [\bar{L}(n, i) - \bar{L}(n, i)]_{feed} \\
 & + C_7 [\bar{I}]_w \frac{\bar{m}}{(\bar{F}^{2/3} + \bar{m})} \delta_{il} \delta(v - v_m) \delta_{nl}.
 \end{aligned} \tag{4.11}$$

This dimensionless population balance equation is then converted into a set of equations containing moments of the distribution in n . To do this the equation is multiplied

by n^k and is summed over all possible values of n . After further simplifications the following equation is obtained:

$$\begin{aligned}
 0 = & \frac{c_7[\bar{I}]_w(\bar{v})^{2/3}}{(\bar{F}_{2/3} + \bar{m})} [\bar{L}_k(i-1) - \bar{L}_k(i) + \bar{f}(i-1)] \\
 & + c_2[\bar{I}]_P(\bar{v}) [\bar{L}_k(i-2) - \bar{L}_k(i) + 2\bar{f}(i-2)] \\
 & + c_3(\bar{v})^{1/3} [i \bar{L}_k(i+1) - (i) \bar{L}_k(i)] \\
 & + c_{101}(\bar{\varphi}) \left[\sum_{m=1}^k \frac{1}{m} \bar{L}_{k-m}(i) \right] + c_{63}[\bar{T}_r]_P[i \bar{f}(i) \\
 & - \bar{L}_k(i)] + c_{102}(\bar{\varphi}) [i \bar{f}(i) - \bar{L}_k(i)] + \frac{c_4}{\bar{v}} [(i+1)(i) \bar{L}_k(i+2) \\
 & - (i)(i-1) \bar{L}_k(i)] + \frac{1}{\theta} [\bar{L}_k(i)_{\text{feed}} - \bar{L}_k(i)] \\
 & + c_7[\bar{I}]_w \frac{\bar{m}}{\bar{F}_{2/3} + \bar{m}} \delta_{il} \quad (4.12)
 \end{aligned}$$

Other modelling equations can be rendered dimensionless and the population balance equations can be transformed to a set of equations involving moments of dimensionless distributions, by using the aforesaid approach. The resultant set of modelling equations is given on the following page.

Particle Size Distribution:

$$0 = \bar{x}_{avg} \cdot n \frac{\bar{\phi}}{1-\bar{\phi}} \bar{F}_{k-1} + C_7 \frac{[\bar{I}]_w (\bar{m})}{(\bar{F}_{2/3} + \bar{m})} + \frac{1}{\theta} [\bar{F}_k - \bar{F}_k]_{feed} \quad k=0 \text{ to } n \quad (4.13)$$

Radical Number Distribution:

$$0 = C_7 \frac{[\bar{I}]_w (\bar{v})^{2/3}}{(\bar{F}_{2/3} + \bar{m})} [\bar{f}(i-1) - \bar{f}(i)] + C_3 (\bar{v})^{-1/3} [\bar{f}(i+1) \bar{f}(i+1) - i \bar{f}(i)] + \frac{C_4}{(\bar{v})} [(\bar{v}-v_m) \delta_{il} \delta(v-v_m) + \frac{1}{\theta} [\bar{f}(i) - \bar{f}(i)]_{feed}] + C_2 (\bar{v}) [\bar{I}]_P [\bar{f}(i-2) - \bar{f}(i)] \quad i = 0 \text{ to } m \quad (4.14)$$

Live Polymer Chain Length Distribution:

$$0 = C_7 \frac{[\bar{I}]_w (\bar{v})^{2/3}}{(\bar{F}_{2/3} + \bar{m})} [\bar{L}_k(i-1) - \bar{L}_k(i) + \bar{f}(i-1)] + C_2 [\bar{I}]_P (\bar{v}) [\bar{L}_k(i-2) - \bar{L}_k(i) + 2 \bar{f}(i-2)] + C_3 (\bar{v})^{-1/3} [(i) \bar{L}_k(i+1) - (i) \bar{L}_k(i)] + C_{101} (\bar{\phi}) [\sum_{m=1}^k \frac{k}{m} \bar{L}_{k-m}(i)] + C_{63} [\bar{f}(i) - \bar{L}_k(i)] + C_{102} (\bar{\phi}) [(i) \bar{f}(i) - \bar{L}_k(i)] + \frac{C_4}{(\bar{v})} [(i+1) (i) \bar{L}_k(i+2) - (i) (i-1) \bar{L}_k(i)] + \frac{1}{\theta} [\bar{L}_k(i) - \bar{L}_k(i)]_{feed} + C_7 [\bar{I}]_w \frac{\bar{m}}{(\bar{F}_{2/3} + \bar{m})} \delta_{il} \quad k=0 \text{ to } n; i=0 \text{ to } m \quad (4.12)$$

Dead Polymer Chain Length Distribution:

$$\begin{aligned}
0 &= \frac{C_7 [\bar{I}]_W (\bar{v})^{2/3}}{(\bar{F}_{2/3} + \bar{m})} [\bar{\bar{G}}_k(i-1) - \bar{\bar{G}}_k(i)] \\
&+ \frac{C_3}{(\bar{v})^{1/3}} [(i+1) \bar{\bar{G}}_k(i+1) - (i) \bar{\bar{G}}_k(i)] + C_2 [\bar{I}]_P(\bar{v}) \\
&[\bar{\bar{G}}_k(i-2) - \bar{\bar{G}}_k(i)] + \frac{C_4}{(\bar{v})} [(i+2)(i+1) \bar{\bar{G}}_k(i+2) \\
&- (i)(i-1) \bar{\bar{G}}_k(i)] + \frac{C_{403}}{(\bar{v})} [f(i+2) \sum_{p=0}^{k-1} \frac{\underline{k}}{\underline{k-p}} \underline{\underline{g}}_p(i+2) g_{k-p}(i+2)] \\
&+ C_{102}(\bar{\varphi}) [\bar{L}_k(i)] + \frac{C_{404}}{(\bar{v})} [(i+1) \bar{\bar{L}}_k(i+2)] + C_{63} [\bar{T}r]_P \bar{\bar{L}}_k(i) \\
&+ \frac{1}{\theta} [\bar{\bar{G}}_k(i) \underset{\text{feed}}{-} \bar{\bar{G}}_k(i)]
\end{aligned}$$

$k=0 \text{ to } n$
 $i=0 \text{ to } m$
(4.15)

Emulsifier Balance:

$$O = \frac{1}{\theta} [[\bar{S}_T]_{feed} - [\bar{S}]_T] \quad (4.16)$$

Initiator Balance:

$$O = - [\bar{I}]_W + \frac{C_{56}}{\bar{a}} [\bar{I}]_W \left|_{\text{feed}} - [\bar{I}]_W \right| \quad (4.17)$$

Monomer Balance:

$$O = \frac{1}{\Phi} \left[[\bar{M}]_T - [\bar{M}]_{T_{feed}} \right] + C_{53}(\bar{\varphi}) F_o \cdot i_{avg} \quad (4.18)$$

Micelle Balance:

$$0 = -\bar{m} + c_{50} [\bar{s}]_T - c_{51} - \bar{F}_{2/3} - c_{52} \bar{m}_d \quad (4.19)$$

Monomer Droplets Balance:

$$0 = -\bar{m}_d + C_{54}[\bar{M}]_T - C_{55} \cdot \bar{F}_1 \cdot (\bar{\varphi}) - C_{556}(\bar{m}) \quad (4.20)$$

Expression for ($\bar{\varphi}$):

$$0 = (1-\bar{\varphi}) + \ln(\bar{\varphi}) + \frac{(1-\bar{\varphi})^2}{2} + C_{65} \frac{\bar{F}_{1/3}}{\bar{F}_{1/3}} \quad (4.21)$$

It should be noted that the expressions 4.12, 4.14, 4.15 are functions of k and i . Similarly the expression 4.13 is a function of k . Thus, these expressions indeed represent sets of equations depending on different values of k and i .

The particle size distribution is continuous and the distribution in n can be assumed to be similar to a continuous distribution. Hence the different moments are used in the aforesaid sets of equations to represent them. However, it can be seen from these expressions that different fractional moments with respect to volume (i.e. $\bar{F}_{1/3}$ and $\bar{F}_{2/3}$) need to be evaluated separately. This can be done by expressing the differential distribution in volume in terms of first 5 integer moments as described previously in equations 4.4 to 4.10. Any fractional moment \bar{F}_t can then be calculated from the differential distribution. The resulting equation expressing the fractional moment in terms of first 5 integer moments is given by

$$\begin{aligned} \bar{G}_t^x &= \frac{1}{\alpha+1} + \frac{1}{\alpha+1} \sum_{n=3}^m c_n^\alpha \sum_{k=0}^n (-1)^k \\ &\times \frac{(n+\alpha+1)}{(n-k+1)} \frac{(\alpha+t+k+1)}{(\alpha+k+1)} \frac{1}{k+1} \end{aligned} \quad (4.22)$$

The distribution in i , however, cannot be approximated to behave like a continuous distribution, as explained previously. Hence, the equations involving i (i.e. 4.12, 4.14 and 4.15) are solved for different discrete values of i . These equations, however, involve advanced terms calculated at $(i+1)$ and $(i+2)$. Thus the set of n equations will involve $n+2$ variables. However, as stated earlier, the number of radicals in a particle is generally limited to a small value. In other words, after a sufficiently high value of i the terms evaluated at $(i+1)$ and $(i+2)$ tend to become zero. Thus the advanced terms requiring evaluation of $(i+1)$ and $(i+2)$ can be approximated to zero and the set of n equations can be solved for remaining n variables.

Method of Solution:

The modelling equations given earlier can be divided into 3 groups. The first group of equations contains various material balance equations, the equation for $(\bar{\varphi})$, the equations involving moments of particle size distribution and the equations involving radical number distribution. Equations 4.13, 4.14, 4.18, 4.19, 4.20 and 4.21 are interdependent and should be solved simultaneously. Equations 4.16 and 4.17 can be solved separately. The solution of this first group of equations gives the values of the various dimensionless concentration terms, radical number distribution and moments of particle size distribution. Using these values the second

group of equations involving moments of live polymer chain length distribution can be solved to obtain $\bar{f}_k(i)$ for different values of i and k . These answers can further be utilized to solve the third group of equations involving dead polymer chain length distribution to obtain $\bar{G}_k(i)$ for different values of i and k .

The basic aim of solving these modelling equations was to evaluate the properties of the output stream from the reactor. These properties include conversion, rate of polymerization in the reactor, number of particles per cm^3 of emulsion, particle size distribution and the molecular weight of the resultant polymer. The conversion, rate and number of particles per cm^3 emulsion can be calculated by knowing the values of $[\bar{M}]_T$ and $[\bar{F}_o]$.

The expressions to obtain these are given below:

$$X = \frac{[\bar{M}]_T \text{ feed} - [\bar{M}]_T}{[\bar{M}]_T \text{ feed}} \quad (4.23)$$

$$\text{Rate} = \frac{[\bar{M}]_T \text{ feed} - [\bar{M}]_T}{\theta} \times \frac{[I_o]_w k_p d_m}{N_A \cdot d_p \cdot v_m} \quad (4.24)$$

$$N_o = [\bar{F}_o] \cdot m_o \cdot N_A \quad (4.25)$$

The particle size distribution of the latex can be generated from its moments \bar{F}_k by using equation 4.22. The average volume and average diameter of these latex particles can also be evaluated from these moments.

$$V_{avg} = \frac{\bar{F}_1}{\bar{F}_0} \quad (4.26)$$

$$D_{number} = \frac{\bar{F}_1^{1/3}}{\bar{F}_0} \times \left(\frac{6}{\pi}\right)^{1/3} \quad (4.27)$$

$$D_{vol} = (V_{avg} \times \frac{6}{\pi})^{1/3} \quad (4.28)$$

The number average as well as weight average molecular weight and the polydispersity index can be calculated from moments of dead polymer chain length distribution. These expressions are given below:

$$M_n = \frac{\bar{G}_1}{\bar{G}_0} \quad (4.29)$$

$$M_w = \frac{\bar{G}_2}{\bar{G}_1} \quad (4.30)$$

where $\bar{G}_s = \sum_{i=0}^m \bar{G}_s(i)$ (4.31)

$$PDI = \frac{M_w}{M_n} \quad (4.32)$$

The entire molecular weight distribution of the resultant polymer can also be generated from these moments by using the procedure outlined earlier in equations 4.4 to 4.10 and 4.22.

Description of Computer Program:

The equations in group 1 are subdivided in two sub-groups, first containing equations 4.16 to 4.21 and 4.13 while the second comprises of equation 4.14. For the solution

of the first subgroup the feed conditions and initial guesses for variables as well as initial guess for \bar{i}_{avg} is needed. This group of equations can then be solved to give values of different dimensionless concentrations and the moments of particle size distribution. These values are then fed to the second subgroup which now becomes a set of linear equations. The starting value of $m=5$ is used. The values of $\bar{f}(m+1)$ and $\bar{f}(m+2)$ are assumed to be zero and the equations are solved to get values of $\bar{f}(i)$; $i=1, m$. An increment of two is then made in m and $(m+2)$ equations are again solved, assuming $\bar{f}(m+3)$ and $\bar{f}(m+4)$ to be equal to zero. The values of $\bar{f}(i)$, $i=1, m+2$ are then obtained. The values of $\bar{f}(m+1)$ and $\bar{f}(m+2)$ are checked to be less than a small constant z (10^{-7} was used). If these values are not less than z an increment of two is again made in m and the set of equations is again solved. The process is repeated until the last two variable values of a set are less than z , thus checking the assumption made in the solution of previous set. The value of \bar{i}_{avg} is then calculated. This improved value of \bar{i}_{avg} is then compared with the previous value of \bar{i}_{avg} and is used together with the values of different variables as the improved guesses to solve the equations in the first subgroup. The process is repeated till the difference between two values of \bar{i}_{avg} is less than a particular convergence criterion.

The equations in the first subgroup are highly nonlinear in nature. The method suggested by Brown has been employed

to solve these equations. The details of this method were taken from reference [31]. It has been found that it is not possible to give good initial guesses for these equations, without which the method does not tend to converge to the solution. Because of this reason an objective function consisting of sums of absolute values of RHS of the aforesaid equations is formed. This objective function is then minimized with respect to the aforesaid variables using the multivariable constrained optimization technique of Box. The details of this technique are given in references [32] and [33]. Relatively worse initial guesses can be used for this optimization procedure. The values of different variables from the optimization procedure are then used as the initial guesses for solving the equations of first group. The solution of linear equations is done by standard matrix inversion procedure. The subprogrammes needed for solution of this group are NONLIN, BACK, AMAX1, AMIN1, FUNX, CONSX, CHECH, CENTR, FUNC, CONST, MATIN, SET1, LINEAR, GOLDEN, CONF1 and OBJECT.

The answer from first group of equations are then used for the solution of equations in second group. While solving the radical number distribution (i.e. second subgroup of group 1), the assumption made for solving a set of equations was immediately checked by solving an increased number of equations. Hence when the convergence was ultimately achieved, the assumptions made in the solution of last but one set of

equations (containing say, 'p' equations) was verified by the solution of last set of equations. Hence these 'p' equations should, in principle, be solved for live and dead polymer chain length distribution. However, the solution of live polymer chain length distribution for $f(o)$ type of particles is meaningless. Hence 'p-l' equations are solved for live polymer chain length distribution. However, for dead polymer chain length distribution 'p' equations need to be solved. The equations are linear and are solved by the same method employed earlier. The subprogrammes required for this part are SET2, MATIN and CONFR.

The answer from 2nd group of equations is then used to solve the equations in 3rd group. The method of solution of these equations is similar to that employed for the 2nd group. The answers from this part give the values of $\overset{=}{G}_k(i)$ for different values of i and k. The subprogrammes required for this portion are SET3, MATIN and CONFR.

After the solution of modelling equations various properties of output stream are calculated by using equations 4.23-4.32. Cumulative particle size distribution, live polymer chain length distribution, molecular weight distribution are then generated. The subprogrammes required for this portion include DISTR, DIFF2, CONFR and QGLO.

The diagram showing the scheme to solve the modelling equations is given on the next page.

I.I.T. KANPUR
CENTRAL LIBRARY
 Acc. No. A 59503

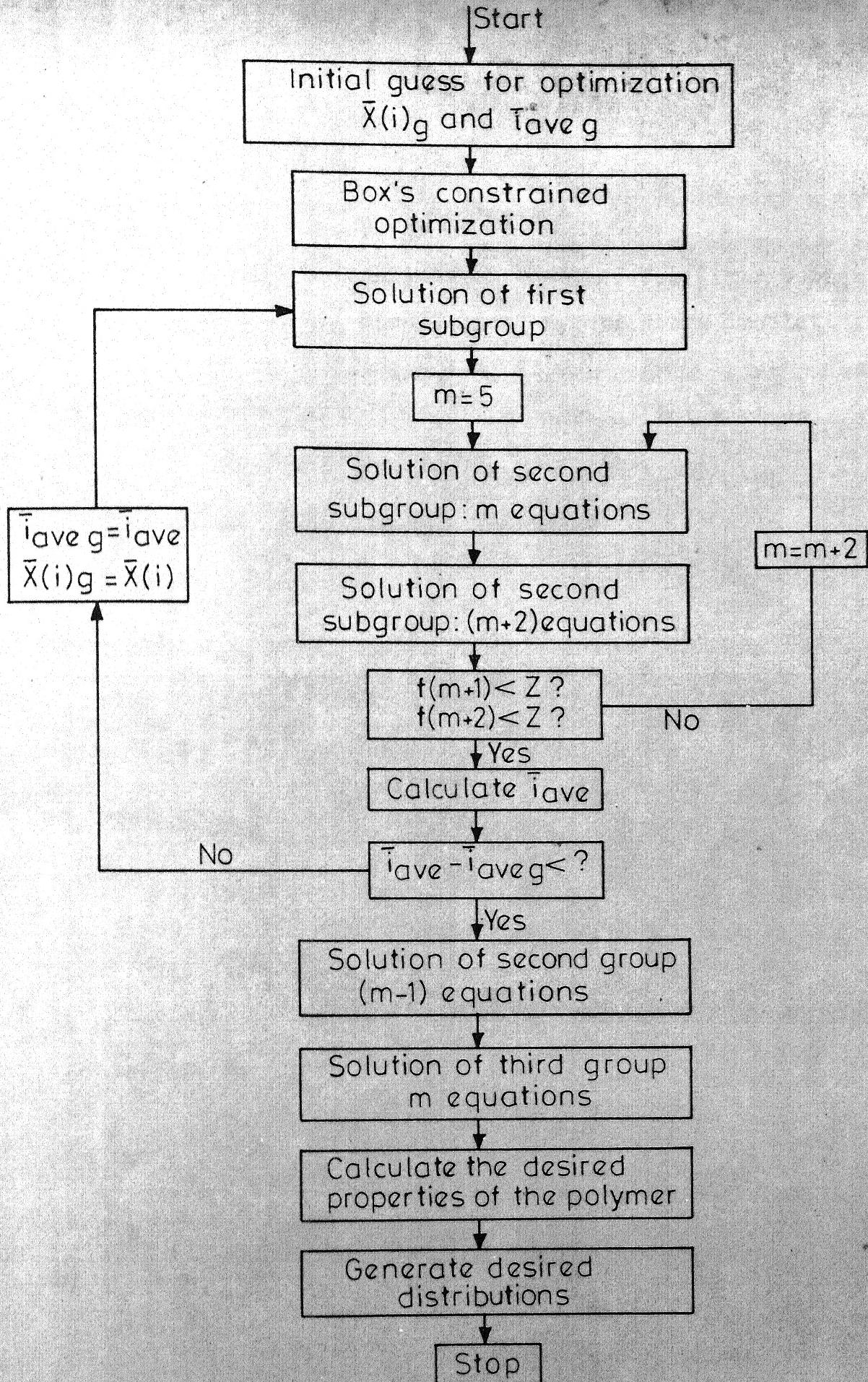


Fig. 1 -Scheme for solution of modelling equations.

NOMENCLATURE

- (1) m_o = nondimensionalization constant for distribution functions, micelle concentration and droplet concentration, moles/lit.
- (2) $[I_o]_w$ = nondimensionalization constant for various concentration terms, moles/lit.

Definitions of Dimensionless Variables:

1. $[\bar{S}]_T = \frac{[S]_T}{[I_o]_w}$
2. $[\bar{I}]_w = \frac{[I]_w}{[I_o]_w}$
3. $[\bar{M}]_T = \frac{[M]_T}{[I_o]_w}$
4. $[\bar{m}] = \frac{m}{m_o}$
5. $(\bar{m}_d) = \frac{m_d}{m_o}$
6. $\bar{f}(i) = \frac{f(i)}{m_o}$
7. $\bar{L}(n,i) = \frac{L(n,i,v_{avg})}{m_o}$
8. $\bar{\bar{L}}_k(i) = \sum_{n=1}^{\infty} n^k \bar{L}(n,i)$
9. $\bar{\bar{F}}_k = \bar{v}^k \bar{F}(\bar{v}) d\bar{v}$
10. $\bar{\bar{F}}(\bar{v}) = \frac{F(v)}{m_o}$

$$11. \bar{G}(n,i) = \frac{G(n,i, v_{avg})}{m_o}$$

$$12. \bar{G}_k(i) = \sum_{n=1}^{\infty} n^k \bar{G}(n,i)$$

Definitions of Dimensionless Constants:

$$1. C_7 = 2 \cdot k_d \cdot N_A \cdot d_P (I_o)_w \cdot v_m \cdot v_w \cdot f / (k_p \cdot d_m \cdot m_o)$$

$$2. C_2 = \frac{k_{ip}}{k_p} N_A^2 v_m^2 [I_o]_w \cdot \frac{d_p}{d_m} \cdot f$$

$$3. C_3 = \frac{k_o}{k_p} \cdot N_A \cdot \frac{d_p}{d_m} \cdot (v_m)^{2/3} (36\pi)^{1/3}$$

$$4. C_4 = \frac{k_{tc} + k_{td}}{2k_p} \cdot \frac{d_p}{d_m}$$

$$5. C_{50} = \frac{[I_o]_w}{(36\pi)^{1/3}} \cdot \frac{CE}{(36\pi)^{1/3} \cdot v_m^{2/3} m_o \cdot N_A}$$

$$6. C_{51} = \frac{s_{wc} CE}{(36\pi)^{1/3} \cdot v_m^{2/3} m_o} \cdot \frac{1}{N_A}$$

$$7. C_{52} = \frac{v_d^{2/3}}{v_m^{2/3}}$$

$$8. C_{53} = \frac{N_A \cdot v_m \cdot d_p \cdot m_o}{[I_o]_w M_w}$$

$$9. C_{54} = \frac{[I_o]_w V_{mol}}{N_A \cdot v_d \cdot m_o}$$

$$10. C_{556} = \frac{[M]_m \cdot v_m \cdot V_{mol}}{v_d}$$

$$11. C_{55} = \frac{v_m}{v_d}$$

$$12. C_{56} = \frac{k_p \cdot d_m}{k_d \cdot N_A \cdot d_p \cdot v_m}$$

$$13. \quad C_{63} = \frac{k_{ft}}{k_P} \cdot [I_o]_w \cdot N_A \cdot \frac{d_p}{d_m} \cdot v_m$$

$$14. \quad C_{65} = \left(\frac{4\pi}{3v_m} \right)^{1/3} \frac{2}{RT} \frac{V_{mol} \cdot \gamma}{}$$

$$15. \quad C_{101} = \frac{d_p \cdot v_M \cdot N_A}{d_m \cdot V_{mol}}$$

$$16. \quad C_{102} = \frac{k_{fm} N_A d_p v_m}{k_P d_m V_{mol}}$$

$$17. \quad C_{403} = \frac{k_{tc}}{2k_P} \cdot \frac{d_p}{d_m}$$

$$18. \quad C_{404} = \frac{k_{td}}{2k_P} \cdot \frac{d_p}{d_m}$$

CHAPTER 5

RESULTS AND DISCUSSION

The modelling equations developed in the earlier chapters have been solved by using the approach outlined in the previous chapter. The model has been applied for the reaction conditions reported by DeGraff and Poehlein [21]. Their work is one of the most readily available and comprehensive work on emulsion polymerization in a CSTR. The experimental set-up used by these authors consisted of a jacketed stirred tank reactor fitted with a temperature controller. Nitrogen atmosphere was maintained inside the reactor. The reactants were fed in two different streams; first stream comprising of monomer (styrene) and the other comprising of aqueous solution of initiator (Ammonium persulfate) and surfactant (sodium lauryl sulfate). The effluent stream was collected in a stirred vessel containing hydroquinone inhibitor and was then used for experimental measurements. They have also developed a theoretical model, using residence time distribution (RTD) approach, to obtain particle size distribution, rate of polymerization, number of particles and number average molecular weight of the polymer product. They have carried out experiments at various reaction conditions and have tried to compare the results with the values obtained from their model. Computations using the model developed here will be compared with their results.

Various constants used for the solution of the modelling equations are listed on the next page.

As has been stated earlier the chain transfer to transfer agent and the initiation in particles have been neglected in the present work. These mechanisms can, in general, be neglected if the monomer feed is free from transfer agent or trace impurities and if it is insoluble in water. Hence the values of k_{ft} and k_{iP} are assumed to be zero. k_{td} is also assumed to be zero, since for styrene the contribution due to termination by disproportionation is usually negligible compared to that with termination by combination [37]. It has found from earlier studies on batch systems that desorption from particles play a relatively small part in emulsion polymerization of styrene [15]. Hence the value of k_o is assumed to be zero. It has been reported that the micelles swell and approximately double their size when saturated with monomer [36]. This implies that approximately 75 per cent of the volume of a saturated micelle is filled with monomer which results in the value of $[M]_m$ as 0.0063.

The reaction was carried out by DeGraff and Poehlein [21] at several temperatures and surfactant concentrations. Only the runs at 70°C and with a surfactant concentration of 2.79 gms/100 gms water were selected for simulation. The monomer to water ratio in the reactor feed was 0.54. From this ratio, V_w is calculated to be 0.6494. The value of $(\bar{\varphi})$ was

Constant	Units	Value	Ref. No.
1. k_p	$\text{cm}^3/\text{mole.sec}$	2.650×10^5	[21]
2. k_o	cms/sec	0.0	assumed
3. k_d	l/sec	2.33×10^{-5}	[21]
4. k_{ip}	l/sec	0.0	assumed
5. k_{td}	$\text{cm}^3/\text{mole sec}$	0.0	assumed
6. k_{tc}	$\text{cm}^3/\text{mole sec}$	7.2×10^9	[21]
7. k_{ft}	$\text{cm}^3/\text{mole sec}$	0.0	assumed
8. k_{fm}	$\text{cm}^3/\text{mole sec}$	21.2	[34]
9. d_p	gms/cm^3	1.0526	[21]
10. d_m	gms/cm^3	0.8780	[21]
11. v_m	cm^3	65.4498×10^{-21}	[13]
12. V_{mol}	$\text{cm}^3/\text{gm.mole}$	118.45	[21]
13. v_d	cm^3	4.1888×10^{-12}	assumed
14. M_w	gm/mole	10 ⁴	-
15. $[M]_m$	moles/cm^3	0.0063	[36]
16. C_E	cm^2/mole	$3.672 \times 10^{+9}$	[21]
17. γ	dynes/cms	7.50	[10]
18. ψ	dimensionless	0.256	[10]
19. S_{wc}	$\text{moles/cm}^3 \text{ water}$	8.9×10^{-6}	[35]

reported by DeGraff and Poehlein [21] to be 0.615 and has been used for most of the simulations.

Calculations were made using the aforesaid constants at various reaction conditions and the results were compared with the theoretical predictions and experimental results obtained by DeGraff and Poehlein [21].

Particle Size Distribution:

The cumulative particle size distributions calculated at different conditions have been presented in Figures 2 and 3. It can be seen from these plots that the results are practically identical to the theoretical results obtained by DeGraff and Poehlein.

As can be seen from the figures, the breadth of the particle size distribution increases with increasing residence time. With the increase in the mean residence time, the distribution of times spent in the reactor broadens. Further the rate of volume growth (r_v) will also increase due to the increase in the value of i_{avg} . Both these effects cause a broader particle size distribution.

In the RTD approach it is assumed that a fraction of the micelles fed to the reactor do not get activated and the rest of the micelles are converted into particles at the same instant. Hence all the particles should have the same birth time. The RTD approach followed by DeGraff and Poehlein then requires the knowledge of only two mechanisms; (1) kinetics of volume growth of particles and (2) RTD of the reactor.

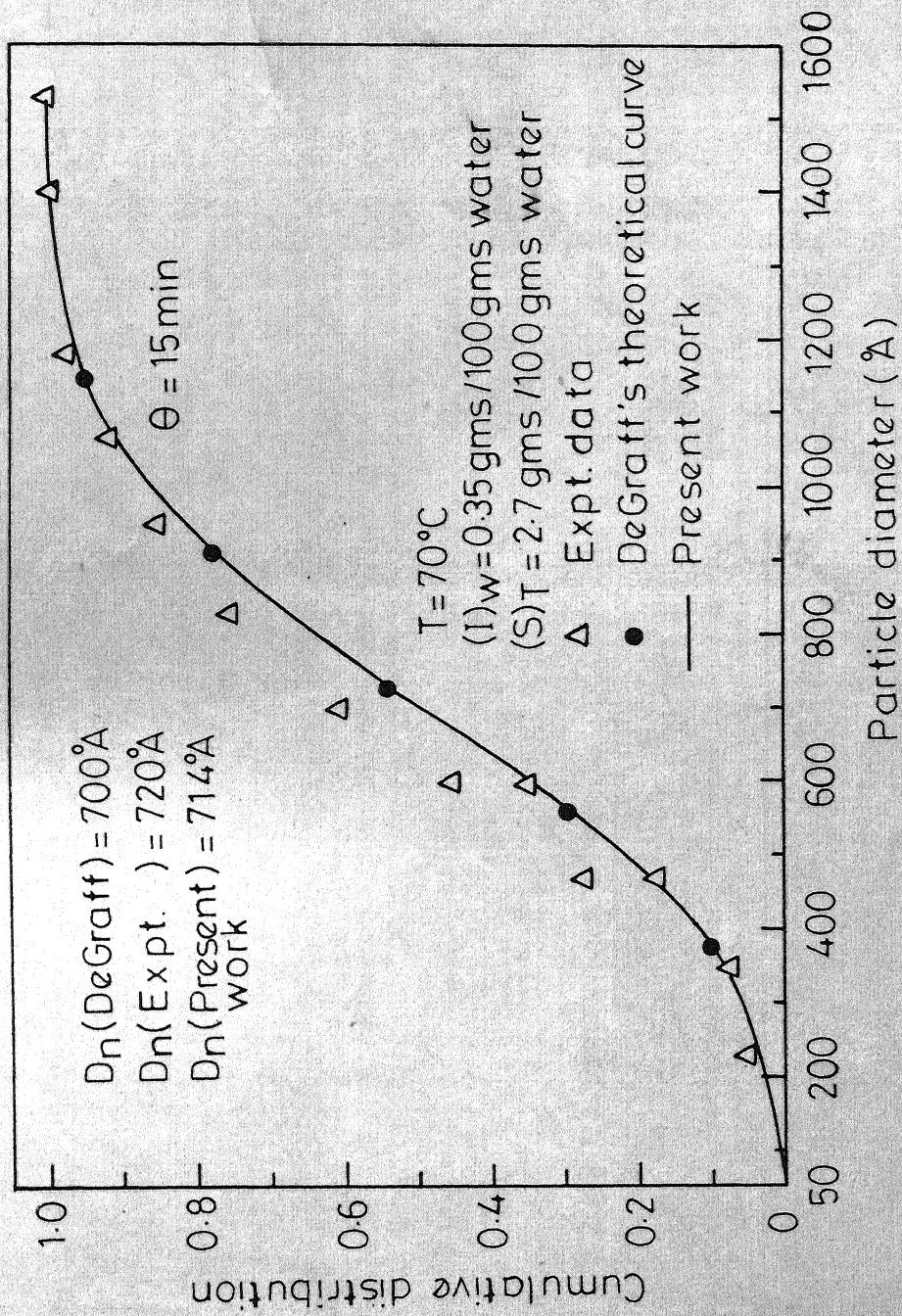


Fig. 2 - Particle size distribution.

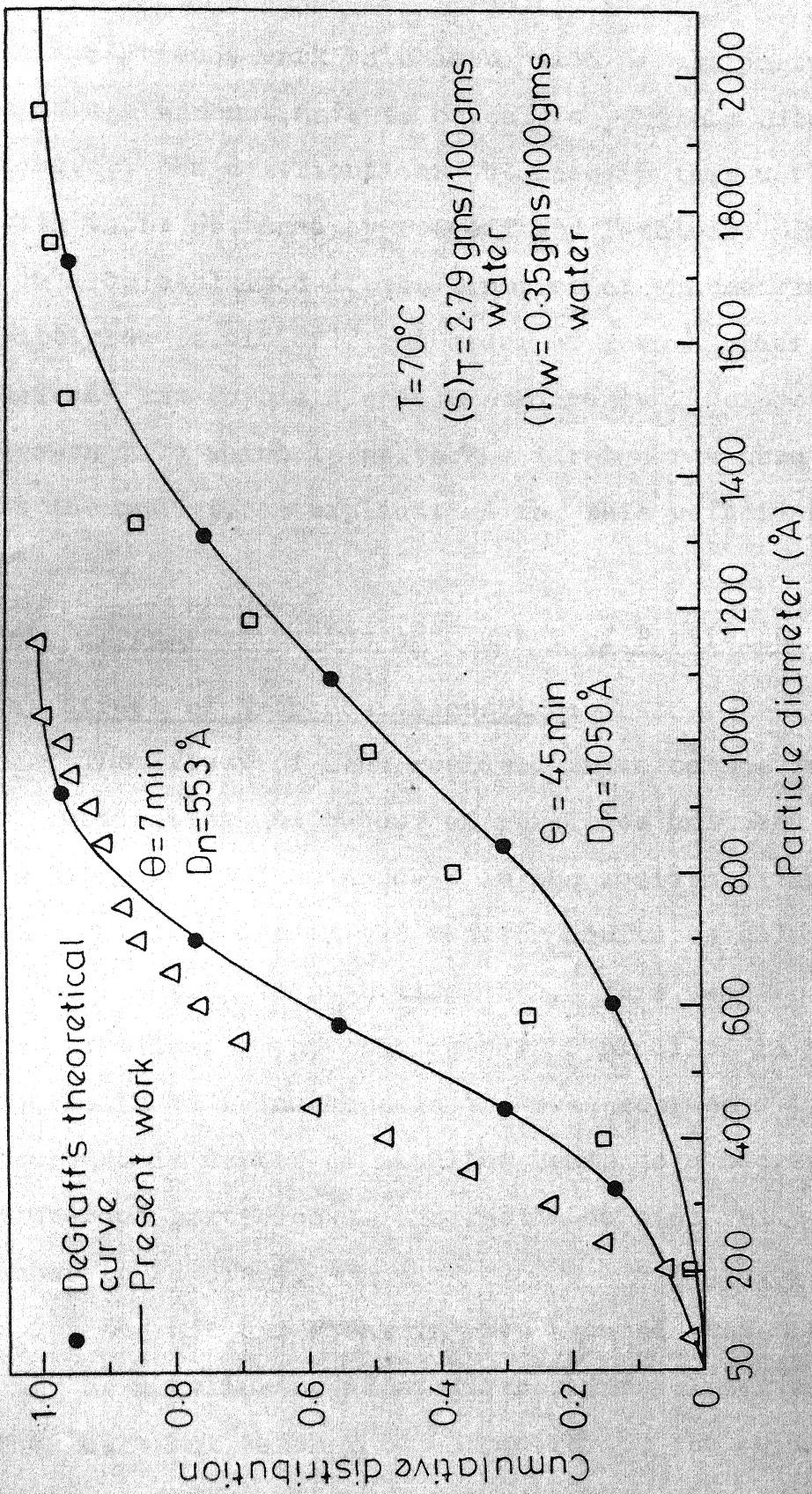


Fig. 3.—Effect of mean residence time on particle size distribution.

In the present work this assumption of all particles being of the same age was not made to calculate particle size distribution. However, the distributions obtained in this work coincide with those obtained by DeGraff and Poehlein. Such coincidence could be explained if the kinetics of volume growth of particles is either first order or zeroth order. However, DeGraff and Poehlein used Stockmayer's[2] expression for growth rate which is neither a first nor a zeroth order process. At the moment, an explanation for this coincidence cannot be offered.

Rate of Polymerization and Number of Particles:

(a) Effect of Mean Residence Time:

The effect of mean residence time on the rate of polymerization and number of particles has been presented in Figure 4. With increase in the residence time, the average size of particles in the reactor increases and requires more amount of soap to stabilize them. This soap is supplied by the micelles. Hence the number of micelles in the reactor decreases with increase in the mean residence time. The decrease in number of micelles leads to a decrease in the number of particles as the residence time is increased as shown in Figure 4.

DeGraff and Poehlein have assumed that soap is present only as micelles or as stabilizer for polymer particles. They have not taken into consideration the emulsifier present

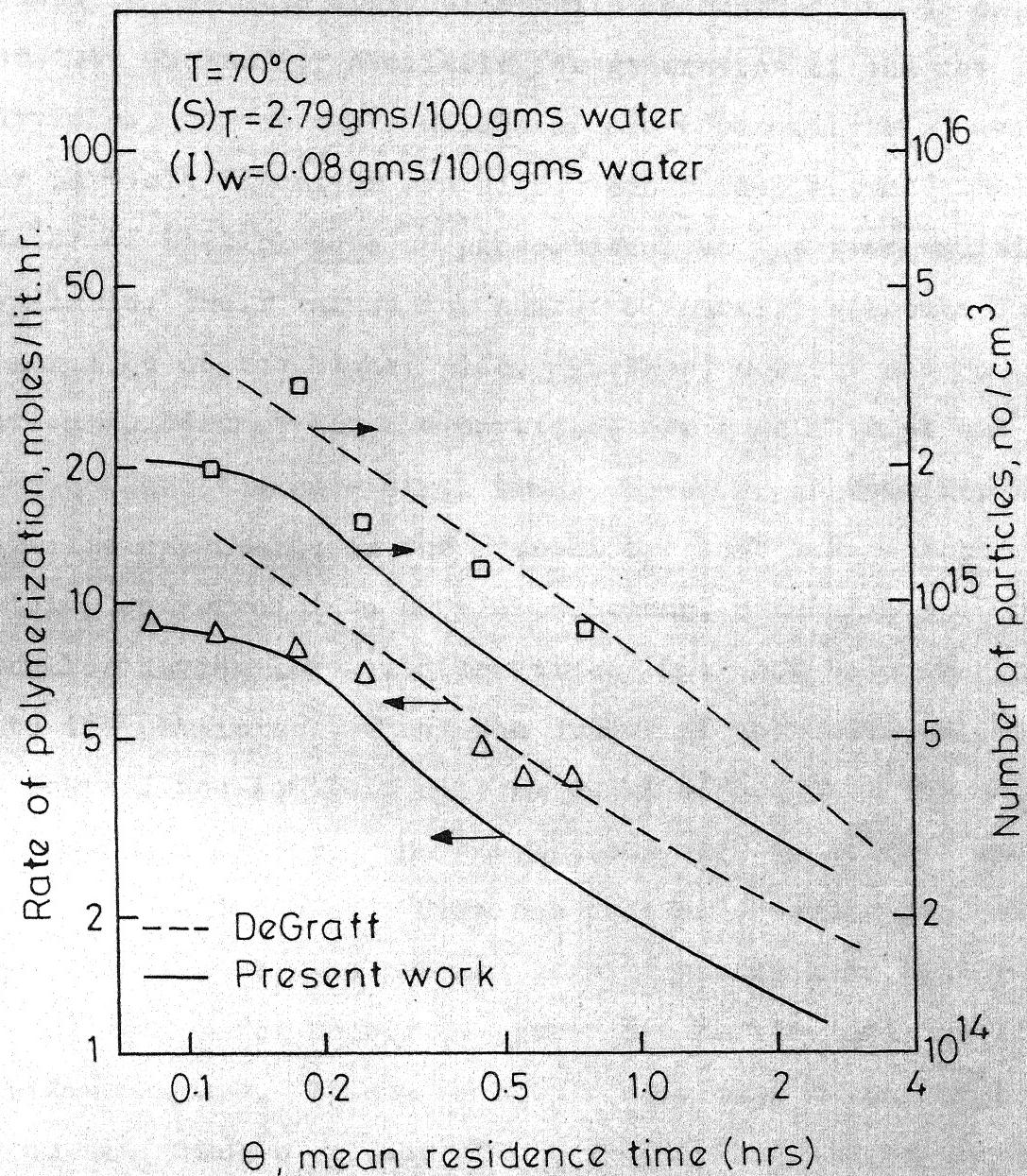


Fig. 4 - Effect of mean residence time on number of particles and rate of polymerization.

on the surface of monomer droplets and that present in aqueous phase upto critical micelle concentration, to calculate the free emulsifier available for generation of the new particles. Hence their values of the free emulsifier available for particle formation are high. Since they assumed the number of particles to be proportional to the free emulsifier available, their values for number of particles should in general be on the higher side. Further, DeGraff and Poehlein have assumed that all the particles are generated at the same time and have the same birth time. However, in reality, the micelles may reside in the reactor for some time before getting converted into particles. Thus, actually, some of the micelles having low residence times, will not be converted into the particles. Hence the number of particles calculated by DeGraff and Poehlein will be still higher than the actual number of particles. The assumptions made by DeGraff and Poehlein, stated above, have not been made in the present work. Therefore, the present results should give better approximation for number of particles, particularly at low residence times. It can be seen from Figure 4 that at low residence times a considerably better fit to the experimental data has been obtained using the present theory, while the results obtained by DeGraff and Poehlein are far from the experimental points. The deviation from theoretical predictions at high residence times is however inexplicable.

The rate of polymerization will be directly proportional to the number of particles times \bar{i}_{avg} as can be seen from 3.28 and 4.24. Hence an explanation similar to that given in the previous paragraph can be given about effect of mean residence time on the rate of polymerization also. The number of particles decreases while the average number of radicals per particle increases with increase in mean residence time. The variation of rate which involves the product of these two terms is given in Figure 4. The almost identical variation of the rate of polymerization and number of particles implies that the average number of radicals in a particle does not vary much with increase in the mean residence time.

(b) Effect of Initiator Concentration:

Figure 5 shows the effect of variation of initiator concentration on M_n and M_w . When the usual values of mean residence time and concentrations of initiator and soap are used the rate of initiation of radicals is considerably high compared to the free emulsifier available. Hence under these reaction conditions the number of particles in the reactor is considerably larger than the number of micelles. When higher concentrations of initiator are used higher rates of generation of radicals are obtained. However, most of these radicals will be captured by the particles since the total area of the particles is considerably larger than the total area of

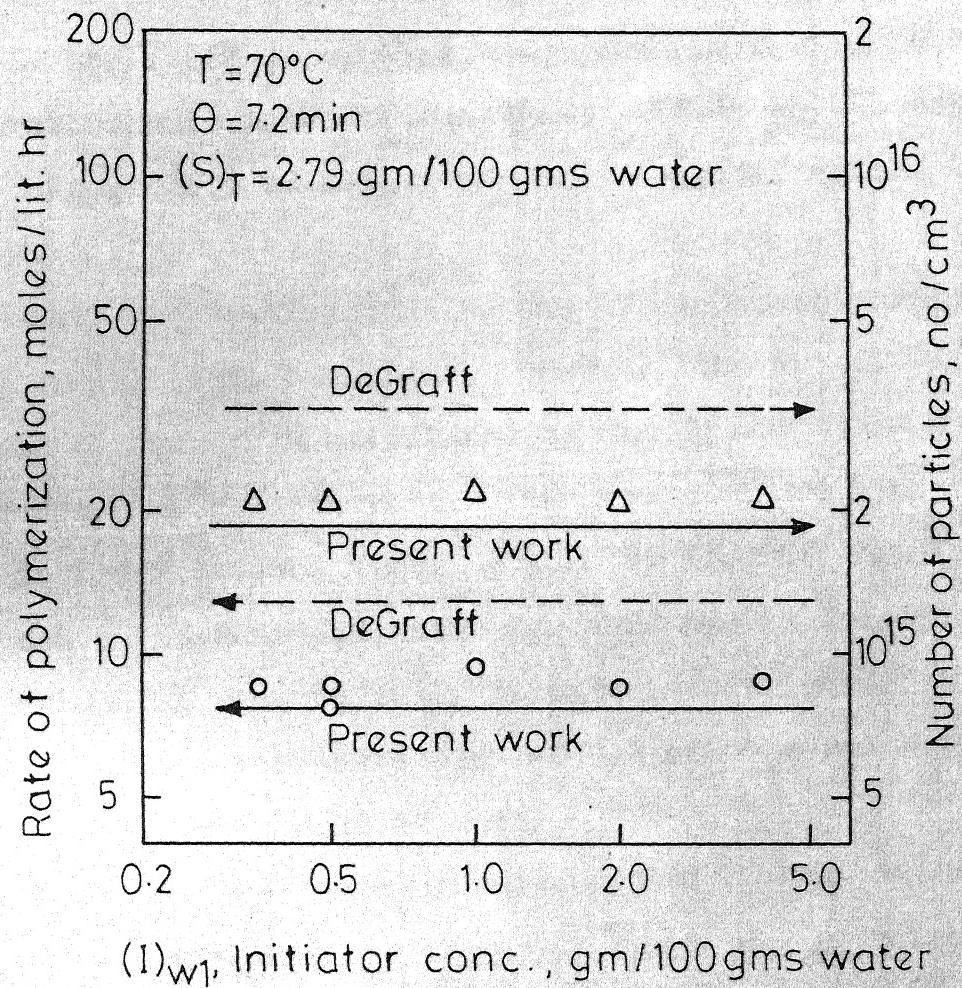


Fig. 5 - Effect of initiator conc. on rate of polymerization and number of particles.

the previous paragraph the value of number of particles is almost constant with respect to initiator concentration. Hence the rate of polymerization, which is proportional to the product of number of particles and the average number of radicals per particle, should not vary significantly with changes in initiator concentration. This behaviour can be seen from Figure 5. The present theory follows the experimental data reasonably well.

Molecular Weights:

(1) Effect of Mean Residence Time:

The effect of mean residence time on M_n and M_w is shown in Figure 6. With increase in mean residence time there is an increase in the particle size and hence the total rate of entry of radicals into particles increases. This decreases the time of growth of radicals inside the particles. Moreover, the average number of radicals present in a particle increases with size leading to the faster termination and lower times of chain growth. Due to these reasons the live chains present inside the particles grow for shorter times resulting in lower molecular weights. This effect can be seen from Figure 6.

DeGraff and Pochlein [21] did not consider the effect due to chain transfer to monomer for calculating M_n . However, this phenomena in general will always be operative and will tend to bring down the molecular weights of the polymer. This

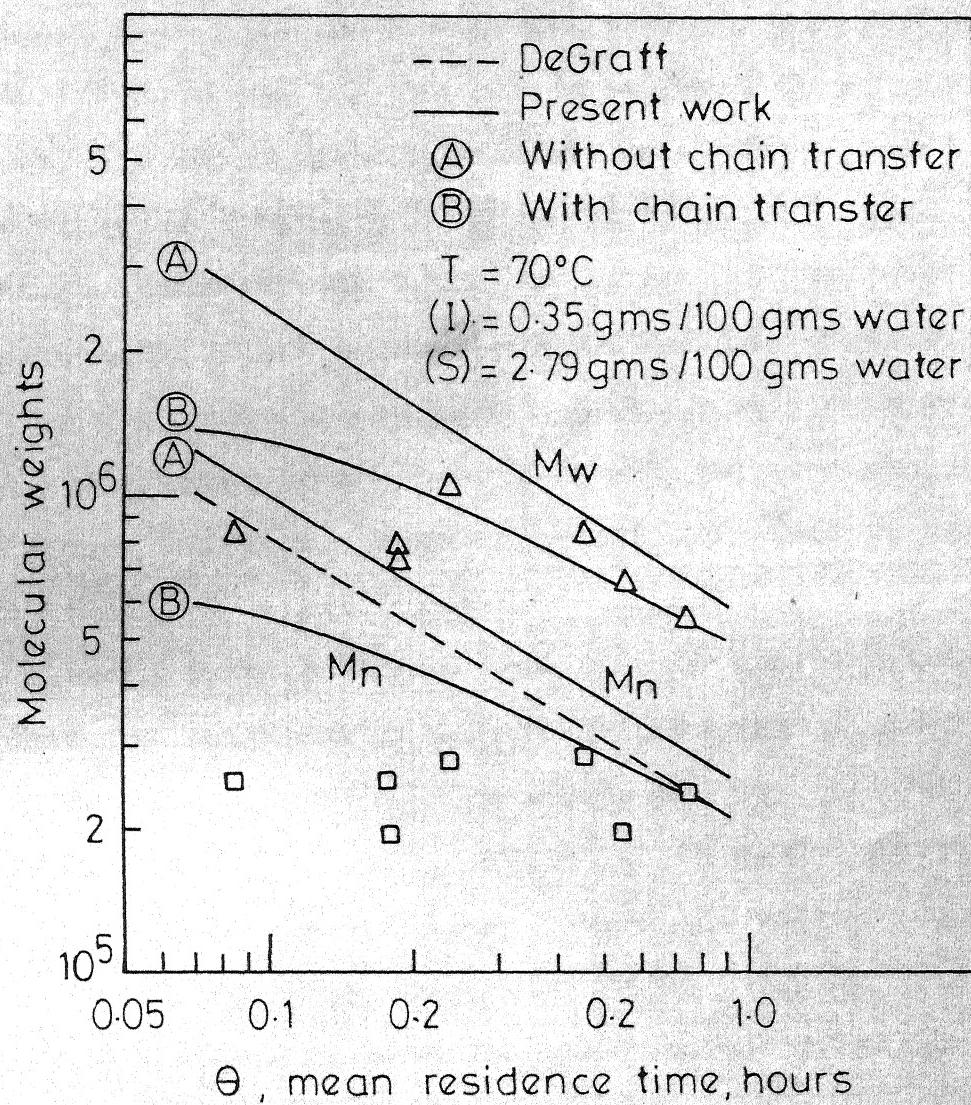


Fig. 6 - Effect of mean residence time on molecular weights.

can be seen from Figure 6 where the variation of molecular weights have been shown with and without considering chain transfer to monomer. The values obtained by the present theory are found to be better approximations to the experimental results than those obtained by DeGraff and Poehlein's theory. However, the values still do not agree with the experimental results satisfactorily at low residence times. Better fit could be obtained by taking into consideration the chain transfer to impurities and the termination by disproportionation. These processes will tend to bring down the molecular weight further.

DeGraff and Poehlein from the RTD approach could not obtain the weight average molecular weight of the product. However, the present theory allows the calculation of M_w from moments of dead polymer chain length distribution. These values are plotted in Figure 6 with and without considering chain transfer to monomer. It can be seen that these values compare fairly well with the experimental results.

(2) Effect of Initiator Concentration:

The effect of initiator concentration on molecular weights has been presented in Figure 7. As the initiator concentration increases the rate of radical absorption in the particles will increase. With the increase in the frequency of radical absorption in a particle, the live chains inside the particles are terminated faster, resulting

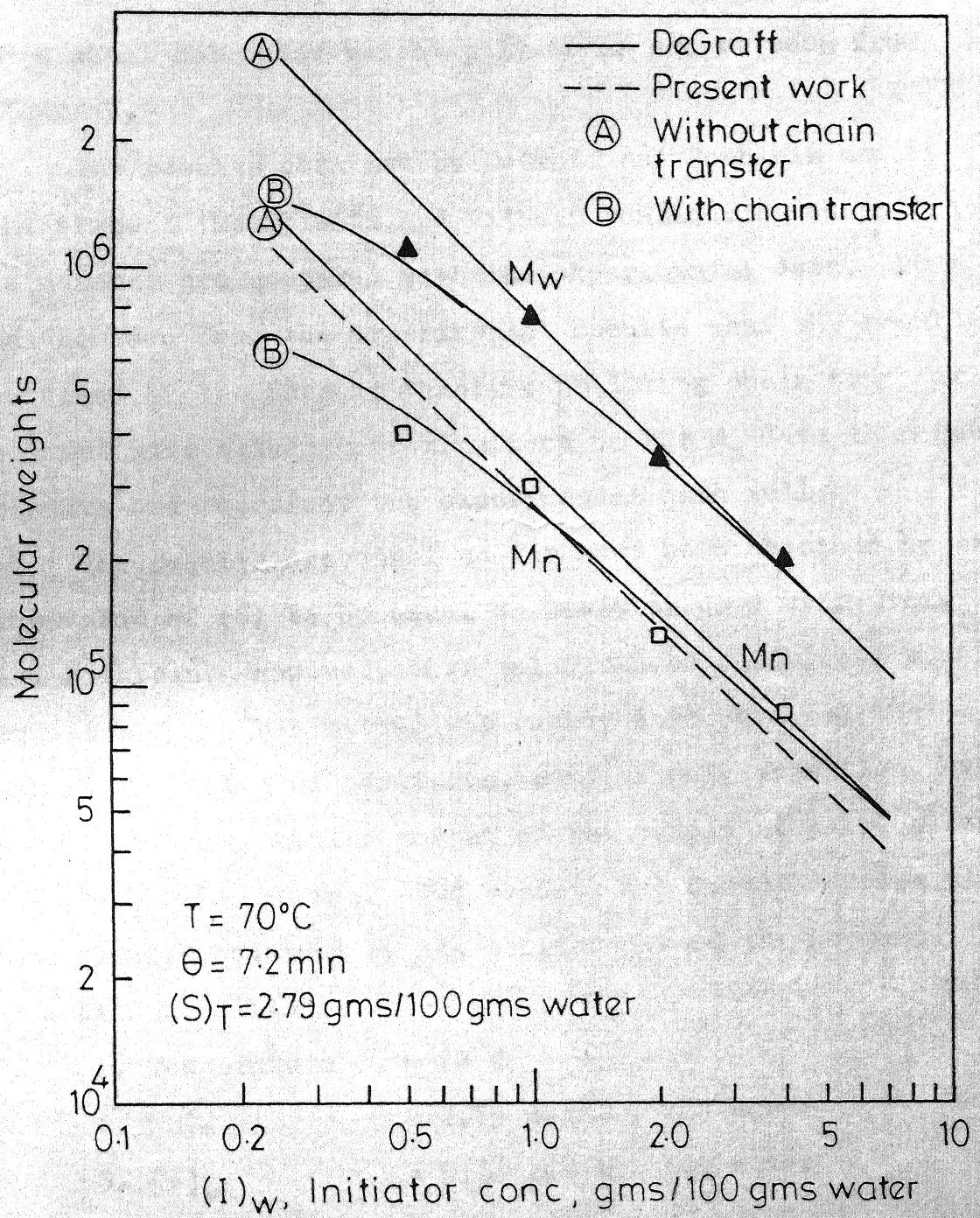


Fig. 7 - Effect of initiator concentration on molecular weights.

in a small molecular weight polymer as can be seen from Figure 7.

The results obtained by DeGraff and Poehlein and by the present theory with and without considering chain transfer to monomer are compared with the experimental data. It can be seen from the experimental results that the results obtained by the present theory considering chain transfer to monomer give better approximations to the M_n data than DeGraff's results and represent the experimental data well.

The results discussed so far have been obtained by using the value of ($\bar{\phi}$) to be equal to 0.615 as used by DeGraff and Poehlein. However, this value can be calculated by using equation 3.31. This method was employed to obtain the values of number of particles, rate of polymerization, average size and the molecular weight of the resultant polymer latex in simulating one run. The results are compared below with the results obtained by the earlier method at the same reaction conditions:

1. Temperature = 70°C
2. $[S]_T$ = 2.79 gms/100 gms water
3. $[I]_W$ = 0.35 gms/100 gms water
4. θ = 15 minutes

	First Approach	Second Approach
1. $\bar{\phi}$	0.615	0.801 (calculated from 4.21)
2. Average diameter (number basis Å)	729	986
3. Average diameter (volume basis Å)	817	1104
4. Number of particles per cm ³ of emulsion	1.09×10^{15}	6.12×10^{14}
5. Rate of polymerization moles/hr	4.7	3.47
6. M_n	5.46×10^5	4.006×10^5
7. M_w	1.32×10^6	9.392×10^5
8. Polydispersity index	2.43	2.344

It can be seen from the table given above that the value of $\bar{\phi}$ obtained by the latex method (i.e. $\bar{\phi} = 0.801$) is different from the one used in earlier method (i.e. $\bar{\phi} = 0.615$). Because of the higher value of $\bar{\phi}$ obtained in the latter method the rate of volume growth would be higher and hence the average particle size obtained is higher. Because of the higher size of particles, they require more soap for stabilization and hence less amount of soap is present as micelles. This results in lower value for number of particles by the latter method. Similarly, because of the larger size of the particles the frequency of radical entry in a particle is increased, leading to shorter growth times for the radicals.

present in the particle. This results in the lower values of molecular weights as reported in the table.

The aforesaid comparison is made without considering chain transfer to monomer. The comparison was also made by taking chain transfer into consideration. The results are similar to the case where chain transfer is not considered. The molecular weights obtained by both the approaches are, however, smaller than the first case.

	First Approach	Second Approach
1. $\bar{\phi}$	0.615	0.819 (Calculated from 4.21)
2. Average diameter (number basis Å)	714	986
3. Average diameter (volume basis Å)	799	1104
4. Number of particles per cm ³ emulsion	1.14×10^{15}	6.12×10^{14}
5. Rate of polymerization moles/hr	4.78	3.47
6. M_n	3.925×10^5	3.092×10^5
7. M_w	9.215×10^5	7.074×10^5
8. Polydispersity index	2.347	2.287

Apart from the results stated earlier, the present theory has the merit of ability to predict the entire molecular weight distribution of the polymer product. The molecular weight distribution on weight basis for a particular set of the reaction conditions is plotted in Figure 8.

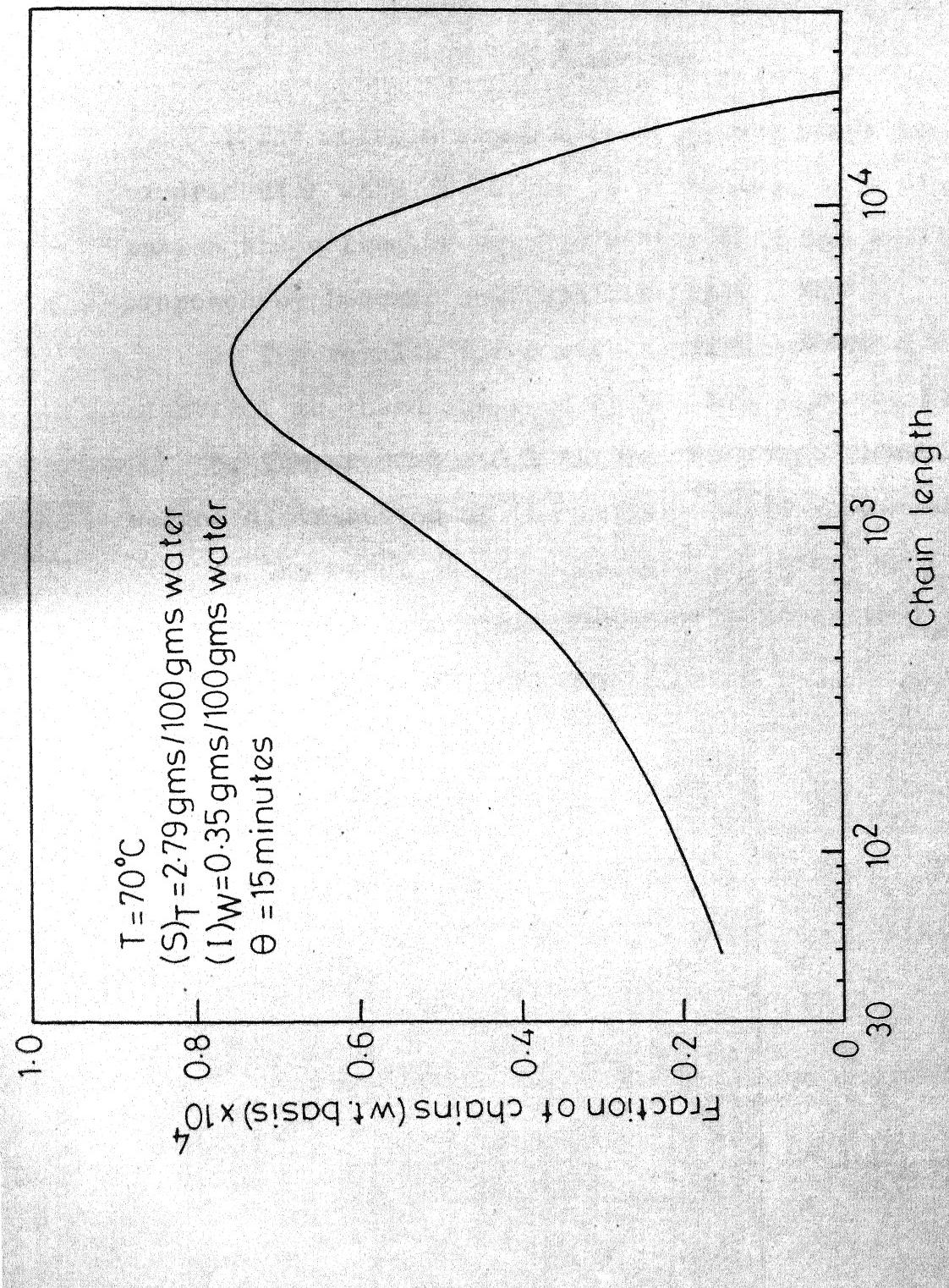


Fig. 8 - Molecular weight distribution of the polymer(wt.basis).

CHAPTER 6

CONCLUSIONS

1. The model presented in this work predicts the experimental data on number of particles, rate of polymerization and molecular weights better than the earlier model proposed by DeGraff and Poehlein [21].
2. The results for particle size distribution are identical to those obtained by the RTD approach [21].
3. The present model allows the complete molecular weight distribution of the polymer product to evaluated.
4. The model can be made more effective by including the phenomena like chain transfer to impurities and termination by disproportionation.

REFERENCES

- [1] W.V. Smith and R.H. Ewart, J. Chem. Phys., 16, 592 (1948).
- [2] W.H. Stockmayer, J. Polym. Sci., 24, 314 (1957).
- [3] J.T. O'Toole, J. Appl. Polym. Sci. 9, 1291 (1965).
- [4] J.L. Gardon, J. Polym. Sci., Part A-1, 6, 623 (1968).
- [5] J.L. Gardon, J. Polym. Sci., Part A-1, 6, 643 (1968).
- [6] J.L. Gardon, J. Polym. Sci., Part A-1, 6, 665 (1968).
- [7] J.L. Gardon, J. Polym. Sci., Part A-1, 6, 687 (1968).
- [8] J.L. Gardon, J. Polym. Sci., Part A-1, 6, 2853 (1968).
- [9] J.L. Gardon, J. Polym. Sci., Part A-1, 6, 2859 (1968).
- [10] J.L. Gardon, Br. Polym. J., 2, 1 (1970).
- [11] J. Ugelstad and F.K. Hanson, J. Polym. Sci., 16, 1953 (1978).
- [12] S.S. Medvedev, Proceedings of the International Symposium on Makromolecular Chemistry, Praugue, 1957, Pergamon, New York, 1958, p.174.
- [13] K.W. Min and W.H. Ray, J. Appl. Polym. Sci., 22, 89 (1978).
- [14] A.E. Alexander and D.H. Napper, Progress in Polymer Science (3) (A.D. Jenkins, ed.), Pergamon, 1975.
- [15] J.W. Vanderhoff, Vinyl Polymerization, Part II (G.E. Ham, Ed.), Dekker, New York, 1969, Chap. 1.
- [16] J.W. Vanderhoff, Advances in Emulsion Polymerization and Latex Technology, 8th Annual Short Course, Lehigh University (G. Poehlein, ed.), 1977.
- [17] K.W. Min and W.H. Ray, J. Macro. Sci.-Rev. Macromol. Chem., C11, 177 (1974).

- [18] D.B. Greshberg and J.F. Longfield, Unpublished Paper presented at the Symp. Poly. Kinetics Cat. Systems, 54th AIChE Meeting, New York, 1961, Preprint No.10.
- [19] T. Udea, S. Omi and H. Kubota, J.Chem. Eng. Japan, 2, 193 (1969); ibid, 4, 50(1971).
- [20] M. Nomura, H. Kojima, M. Harada, W. Eguchi and S. Nagata, J. Appl. Polym.Sci., 15, 675 (1971).
- [21] A.W. DeGraff and G.W. Poehlein, J. Polym.Sci., A-2, 9, 1955 (1971).
- [22] J.D. Stevens and J.O. Funderburk, I and E.C., Proc. Des. and Dev., 11, 360 (1972).
- [23] R.W. Thompson and J.W. Stevens, Chem.Eng. Sci., 32, 311 (1977).
- [24] R.W. Thompson, D.A. Cauley, A.J. Giglio and J.D. Stevens, Chem. Eng. Sci., 33, 383 (1978).
- [25] D. Cauley, A.J. Giglio and R.W. Thompson, Chem.Eng.Sci., 33, 979 (1978).
- [26] V.A. Kirilov and W.H. Ray, 33, 1499 (1978).
- [27] M.R. Grancio and D.J. Williams, J. Polym.Sci., A-1, 8, 2617 (1970).
- [28] D.H. Naper, J. Polym. Sci., A-1, 9, 2089 (1971).
- [29] A.D. Randolph and M.A. Larson, Theory of Particular Processes, Academic Press, New York, 1971.
- [30] K.W. Min, J.Appl. Polym.Sci., 22, 589 (1978).
- [31] Brown, Numerical Solution of Systems of Nonlinear Algebraic Equations [G.D. Byrne and C.A. Hall, Ed.], Academic Press, 1973.

- [32] J.L. Kuster and J.H. Mize, Optimization Techniques with Fortran, McGraw-Hill, New York, 1973.
- [33] G.S. Beveridge and R.S. Schechter, Optimization: Theory and Practice, McGraw-Hill, New York, 1970.
- [34] J. Brandrup, E.H. Immergut, Polymer Handbook, John Wiley and Sons, Inc., 1966.
- [35] Blackley, Emulsion Polymerization, Applied Science Publishers, 1975.
- [36] F.A. Bovey, Emulsion Polymerization, Interscience Publishers, 1955.
- [37] P.J. Flory, Principles of Polymer Chemistry, Cornell University Press, 1953.

```

00100 C ****
00200 C THIS PROGRAM IS USED TO SIMULATE A CONTINUOUS STIRRED TANK
00300 C REACTOR FOR EMULSION POLYMERIZATION. THE INPUT DATA FOR THE
00400 C PROGRAM CONSISTS OF VARIOUS CONSTANTS, FEED CONDITIONS AND
00500 C VARIOUS REACTOR PARAMETERS. THIS PROGRAM IS ABLE TO CALCULATE
00600 C THE CONVERSION, RATE OF POLYMERIZATION, AVERAGE PARTICLE SIZE,
00700 C NUMBER OF PARTICLES, PARTICLE SIZE DISTRIBUTION AND MOLECULAR
00800 C WT. DISTIBUTION OF THE RESULTANT POLYMER LATEX.
00900 C INITIALLY THE BOX'S OPTIMIZATION PROCEDURE IS USED TO
01000 C CALCULATE INITIAL GUESSES FOR THE SOLUTION OF FIRST SUBGROUP. THE
01100 C EQUATIONS OF FIRST SUBGROUP (I.E. MATERIAL BALANCE EQUATIONS,
01200 C Particle size distribution and radical number distribution) are
01300 C THEN SOLVED. THE EQUATIONS OF SECOND GROUP ARE THEN SOLVED TO
01400 C OBTAIN MOMENTS OF LIVE POLYMER CHAIN LENGTH DISTR. THESE RESULTS
01500 C ARE THEN USED TO SOLVE THIRD GROUP TO OBTAIN MOMENTS OF DEAD
01600 C POLYMER CHAIN LENGTH DISTR. VARIOUS OUTPUT VARIABLES ARE THEN
01700 C CALCULATED AND THE REQUIRED DISTRIBUTIONS ARE GENERATED.
01800 C THE DETAILS OF THE METHOD OF SOLUTION OF THE MODELLING
01900 C EQUATIONS IS GIVEN IN THESIS [PAGE ].
02000 C
02100 C NOMENCLATURE -----
02200 C VARIABLES -----
02300 C XSTART(1,1) ----- TOTAL SOAP CONCENTRATION
02400 C XSTART(1,2) ----- INITIATOR CONCENTRATION
02500 C XSTART(1,3) ----- TOTAL MONOMER CONCENTRATION
02600 C XSTART(1,4) ----- MICELLE CONCENTRATION
02700 C XSTART(1,5) ----- MONOMER DROPLETS CONCENTRATION
02800 C XSTART(1,6) ----- MONOMER VOLUME FRACTION IN
02900 C XSTART(1,7) ----- ZEROTH MOMENT W.R.T. VOLUME
03000 C XSTART(1,8) ----- FIRST MOMENT W.R.T. VOLUME
03100 C XSTART(1,9) ----- SECOND MOMENT W.R.T. VOLUME
03200 C XSTART(1,10) ----- THIRD MOMENT W.R.T. VOLUME
03300 C XSTART(1,11) ----- FOURTH MOMENT W.R.T. VOLUME
03400 C
03500 C XRAD(I) ----- RADICAL NUMBER DISTRIBUTION ; NUMBER OF
03600 C PARTICLES HAVING I RADICALS.
03700 C XLIVE(K,I) ----- MOMENTS OF LIVE CHAIN LENGTH DISTRIBUTION,
03800 C [K TH MOMENT ; I RADICALS IN PARTICLE,]
03900 C XDEAD(K,I) ----- MOMENTS OF DEAD CHAIN LENGTH DISTRIBUTION,
04000 C [K TH MOMENT ; I RADICALS IN PARTICLE,]
04100 C XFD(I) ----- FEED CONDITIONS FOR VARIABLES OF FIRST SUB-
04200 C GROUP.
04300 C FFD(I)
04400 C FDLIV(K,I) ----- FEED CONDITIONS OF XLIVE(K,I)
04500 C FDDED(K,I) ----- FEED CONDITIONS OF XDEAD(K,I)
04600 C
04700 C CONSTANTS -----
04800 C AKP ----- PROPATION RATE CONSTANT.
04900 C AKD ----- RATE CONSTANT FOR DISSOCIATION OF INITIATOR

```

5000 C IN AQUEOUS PHASE.
 5100 C AKIP-----RATE CONSTANT FOR DISSOCIATION OF INITIATOR
 5200 C IN POLYMER PHASE.
 5300 C AKZERO-----DESORPTION RATE CONSTANT.
 5400 C AKTC-----RATE CONSTANT FOR TERMINATION BY COMBINATIO
 5500 C AKTD-----RATE CONSTANT FOR TERMINATION BY DISPROPOR-
 5600 C TIONATION.
 5700 C AKFT-----RATE CONSTANT FOR CHAIN TRANSFER TO T
 5800 C RANSFER AGENT.
 5900 C AKFM-----RATE CONSTANT FOR CHAIN TRANSFER TO MONOMER
 6000 C DP-----DENSITY OF POLYMER.
 6100 C DM-----DENSITY OF MONOMER.
 6200 C VMOL-----MOLAR VOLUME OF MONOMER.
 6300 C SAI-----MONOMER-POLYMER INTERACTION PARAMETER.
 6400 C YETA-----SURFACE TENSION.
 6500 C AMW-----MOLECULAR WT. OF MONOMER.
 6600 C THETA-----MEAN RESIDENCE TIME.
 6700 C TEMP-----TEMPERATURE.
 6800 C VW-----WATER:EMULSION RATIO.
 6900 C AIOW-----(I_0)_W; SEE PAGE 53
 7000 C AMZERO-----(M_0)₀; SEE PAGE 53
 7100 C VD-----VOLUME OF DROPLETS.
 7200 C VM-----VOLUME OF MICELLES.
 7300 C CE-----AREA OCCUPIED BY SOAP.
 7400 C AMM-----MONOMER CONCENTRATION IN MICELLES.
 7500 C FEE-----EFFICIENCY OF DECOMPOSITION OF THE
 7600 C INITIATOR.
 7700 C SWC-----CRITICAL MICELLE CONCENTRATION.
 7800 C
 7900 C
 8000 C IPRIN-----PRINTING OPTION : IPRIN=1---INTERMIDIATE
 8100 C ITERATIONS ARE PRINTED ; IPRIN=0---ONLY
 8200 C DATA AND ANSWER FROM SIMULATION ARE PRINTE
 8300 C IPT-----PRINTING OPTION : IPT=0---CONCISE OUTPUT
 8400 C IPT=1---DETAILED OUTPUT [ENTIRE DISTR,
 8500 C ARE CALCULATED/PRINTED]
 8600 C MX-----INPUT UNIT NUMBER TO BE TYPED BY THE USER
 8700 C NX-----RESULT UNIT NUMBER TO BE TYPED BY THE USER
 8800 C NX=1,20,21,22,23,24----RESULT ON DISK
 8900 C NX=5----RESULT AT THE TERMINAL
 9000 C
 9100 C *****
 9200 C *****
 9300 C DOUBLE PRECISION XSTART(14,16),R(14,11),F(14),G(16),H(16)
 9400 C DOUBLE PRECISION XC(11),X(20),XTEMP(6),B(25,1),XRAD(0:20)
 9500 C DOUBLE PRECISION T(25,25)
 9600 C DOUBLE PRECISION FFD(0:20),FDLIV(0:5,20),FDDED(0:5,0:20),XFD(11)
 9700 C DOUBLE PRECISION C50,C51,C52,C53,C54,C556,C56,C7,CT1,C22,C33,C44
 9800 C DOUBLE PRECISION XLIVE(0:5,20),XDEAD(0:5,0:20),XCORD(0:50)
 9900 C DOUBLE PRECISION YCORD(0:50),GSMAL(0:5,20),XLIVET(0:5)

```

10000      DOUBLE PRECISION C79,C78,C65,YETA,XZER,DASHI,DASHI1,DASHI2
10100      DOUBLE PRECISION DIFF,ONETH,TWOTH,TRP,PHI,AM,AIP,SUM1,SUM2
10200      DOUBLE PRECISION BRACE1,BRACE2,BRACE3,SAI
10300      DOUBLE PRECISION THETR,DUMMY,THETVE,V,ANUMBR,AIW, XL1,PART1
10400      DOUBLE PRECISION ZCORD(0:50 ),XVOL(0:5),XDEADT(0:5),XCUM(0:50)
10500      DOUBLE PRECISION X01,X02,C,DAK,X66,X667,XMAX
10600      INTEGER GAMMA
10700      ACCEPT *,MX
10800      ACCEPT *,NX
10900  C      TO READ INPUT PARAMETERS FOR CONSTRAINED OPTIMIZATION
11000      READ(MX,*) NOPT,M,K,ITMAX,IPRIN
11100      READ(MX,*) ALPHA,BETA,DELTA,GAMMA,DASHI
11200      READ(MX,*) (XSTART(1,J),J=1,NOPT)
11300      DO 100 II=2,K
11400  100    READ(MX,*) (R(II,J),J=1,NOPT)
11500  C      TO READ INPUT PARAMETERS FOR LINEAR EQUATIONS AND
11600  C      CONVERGENCE CRITERION.
11700      READ(MX,*) MAXIT,NUMSIG,IPT,NUMEQ
11800      READ(MX,*) EPS,DELNON,PREC,ZETA
11900  C      TO READ INPUT CONSTANTS FOR THE REACTOR.
12000  C
12100  C      RATE CONSTANTS.---
12200      READ(MX,*) AKP,AKD,AKZERO,AKIP,AKTD,AKTC,AKFT,AKFM
12300  C      MONOMER-POLYMER CHARECTARISTICS.---
12400      READ(MX,*) AMW,DP,DM,VMOL,SAI
12500  C      REACTOR CHARECTARISTICS.---
12600      READ(MX,*) THETA,TEMP,VW
12700  C      OTHER CONSTANTS.---
12800      READ(MX,*) AIOW,AMZERO,VD,VM,CE,AMM,FEE,SWC,YETA
12900  C      TO READ FEED CONDITIONS.---
13000      READ(MX,*) (FFD(KI),KI=1,10)
13100      DO 181 KI=0,5
13200      READ(MX,*)(FDLIV(KI,IK),IK=1,10)
13300  181    CONTINUE
13400      DO 182 KI=0,5
13500      READ(MX,*)(FDDDED(KI,IK),IK=0,10)
13600  182    CONTINUE
13700      READ(MX,*)(XFD(KI),KI=1,NOPT)
13800      ANA=6.02E+23
13900      RR=8.2764E+07
14000  C
14100      ONETH=1.0/3.0
14200      TWOTH=2./3.0
14300  C      CALCULATION OF DIMENSIONLESS CONSTANTS.
14400      BRACE1=(36.0*3.142)**ONETH
14500      BRACE2=VD**TWOTH
14600      BRACE3=VM**TWOTH
14700  C
14800      C2=AKIP*ANA*ANA*VM*VM*AIOW*DP/(AKP*DM)
14900      C3=AKZERO*ANA*DP*BRACE3*BRACE1/(AKP*DM)

```

```

15000      C4=(AKTC+AKTD)*DP/(2.0*AKP*DM)
15100      C50=AIOW*CE/(BRACE1*BRACE3*AMZERO*ANA)
15200      C51=(VW*SWC*CE)/(BRACE1*BRACE3*AMZERO*ANA)
15300      C52=BRACE2/BRACE3
15400      C53=ANA*VM*DP*AMZERO/(AIOW*AMW)
15500      C54=AIOW*VMOL/(ANA*VD*AMZERO)
15600      C55=VM /VD
15700      C556=AMM*VM*VMOL/VD
15800      C56=AKP*DM/(AKD*ANA*DP*VM)
15900      CT1=1.0
16000      C7=2.0*AKD*ANA*DP*AIOW*VM*FEE*VW/(AKP*DM*AMZERO)
16100      THETR=THETA*AKP*DM/(ANA*DP*VM)
16200      DUMMY=1.0E+03
16300      C403=AKTC*DP/(2.0*AKP*DM)
16400      C404=AKTD*DP/(2.0*AKP*DM)
16500      THETVE=1.0/THETR
16600      C63=AKFT*AIOW*ANA*DP*VM/(AKP*DM)
16700      C101=DP*VM*ANA/(DM*VMOL)
16800      C102=AKFM*ANA*DP*VM/(AKP*DM*VMOL)
16900      DUMMY2=1.0E-02
17000      C65=((4.0*3.142/(3.0*VM))**ONETH)*(2.*VMOL*YETA/(RR*TEMP))
17100      DUMMY3=1.0E-03
17200      DUMMY4=4.2985E+03
17300 ****
17400 C
17500 C     THE FOLLOWING LINES SHOULD BE INSERTED FOR SECOND APPROACH.
17600 C     X01=0.01
17700 C     X02=0.99
17800 C     C=0.00001
17900 C     WRITE(NX,4189)
18000 C
18100 ****
18200
18300 C     PRINTING THE DATA_____
18400     WRITE(NX,3185)
18500     WRITE(NX,2001)
18600     WRITE(NX,3185)
18700     WRITE(NX,4181)
18800     WRITE(NX,2002) NOPT,M,K,ITMAX,IPRIN
18900     WRITE(NX,2003) ALPHA,BETA,DELTA,GAMMA,DASHI
19000     WRITE(NX,2004)
19100     WRITE(NX,2005) ((J,XSTART(1,J)),J=1,NOPT)
19200     WRITE(NX,2006)
19300     WRITE(NX,2007) MAXIT,NUMSIG,IPT,NUMEQ
19400     WRITE(NX,2008) EPS,DELNON,PREC,ZETA
19500     WRITE(NX,2009) AKP,AKD,AKZERO,AKIP,AKTD,AKTC,AKFT,AKFM
19600     WRITE(NX,2010)
19700     WRITE(NX,2011) AMW,DP,DM,VMOL,SAI
19800     WRITE(NX,2012)
19900     WRITE(NX,2013) THETA,TEMP,VW

```

```

20000      WRITE(NX,2014)
20100      WRITE(NX,2015) AIOW,AMZERO,VD,VM,CE,AMM,FEE,SWC,YETA
20200      WRITE(NX,3015)
20300      WRITE(NX,3016)((KI,FFD(KI)),KI=1,10)
20400      WRITE(NX,2113)
20500      DO 281 KI=0,5
20600      WRITE(NX,2114) (KI,IK,FDLIV(KI,IK),IK=1,10)
20700 281    CONTINUE
20800      WRITE(NX,2115)
20900      DO 282 KI=0,5
21000      WRITE(NX,2116) (KI,IK,FDDDED(KI,IK),IK=1,10)
21100 282    CONTINUE
21200      WRITE(NX,2487)
21300      WRITE(NX,2488) (KI,XFD(KI),KI=1,NOPT)
21400  C
21500  C
21600 2001  FORMAT(//10X,"DATA FOR THE PROBLEM")
21700 2002  FORMAT(//5X,"NOPT = ",I2,5X,"M = ",I2,5X,"K = ",I2,5X,
21800   1"ITMAX = ",I5,5X,"IPRIN = ",I2)
21900 2003  FORMAT(/5X,"ALPHA = ",F10.6,5X,"BETA = ",F10.6,5X,"DELTA = ",
22000   1F10.6//5X,"GAMMA = ",I2,13X,"DASHI = ",F10.6)
22100 2004  FORMAT(//5X,"INITIAL GUESSES-----")
22200 2005  FORMAT(/2(5X,"XSTART(",I2,") = ",E15,8))
22300 2006  FORMAT(//5X,"INPUT PARAMETERS FOR LINEAR EQUATIONS AND
22400   1THE CONVERGENCE CRITERION")
22500 2007  FORMAT(/5X,"MAXIT = ",I5,5X,"NUMSIG = ",I2,5X,"IPT = ",
22600   1I2/5X,"NUMEQ = ",I2)
22700 2008  FORMAT(/5X,"EPS = ",E15,8,3X,"DELNON = ",E15,8,3X,
22800   1"PREC = ",E15,8/5X,"ZETA = ",E15,8)
22900 2009  FORMAT(/5X,"RATE CONSTANTS----"/5X,"KP = ",E15,8,5X,
23000   1"KD = ",E15,8,5X,"KZERO = ",E15,8/5X,"KIP = ",E15,8,5X,
23100   1"KTD = ",E15,8,5X,"KTC = ",E15,8/5X,"KFT = ",E15,8,5X,
23200   1"KFM = ",E15,8 )
23300 2010  FORMAT(/5X,"MONOMER_POLYMER CHARECTARISTICS---")
23400 2011  FORMAT(/5X,"MW = ",E15,8,5X,"DP = ",E15,8,5X,"DM = ",E15,8/
23500   15X,"VMOL = ",E15,8,5X,"SAI = ",E15,8)
23600 2012  FORMAT(/5X,"REACTOR CHARECTARISTICS---")
23700 2013  FORMAT(/5X,"THETA = ",E15,8,5X,"TEMP = ",E15,8,5X,
23800   1"VW = ",E15,8)
23900 2014  FORMAT(/5X,"OTHER CONSTANTS---")
24000 2015  FORMAT(/5X,"AIOW = ",E15,8,4X,"AMZERO = ",E15,8,4X,"VD = ",
24100   1E15,8/5X,"VM = ",E15,8,4X,"CE = ",E15,8,4X,"AMM = ",
24200   1E15,8 /5X,"FEE = ",E15,8,4X,"SWC = ",E15,8,4X,"YETA = ",E15,8)
24300 3015  FORMAT(/5X,"FEED CONDITIONS---"/5X,"RADICAL NUMBER DISTRIBUTI
24400   1ON-----")
24500 3016  FORMAT(/2(5X,"FFD(",I2,") = ",F10,6))
24600 2113  FORMAT(/5X,"LIVE POLYMER CHAINS-----",/)
24700 2114  FORMAT(/3(4X,"FDLIV(",I2,1H,,I2,") = ",F8,4))
24800 2115  FORMAT(/5X,"DEAD POLYMER CHAINS-----",/)
24900 2116  FORMAT(/3(4X,"FDDDED(",I2,1H,,I2,") = ",F8,4))

```

```

25000 2487  FORMAT(//5X,"FEED COND. FOR VARIABLES OF INITIAL SET")
25100 2488  FORMAT(/3(5X,I2,4X,E11.4))
25200 4181  FORMAT(/7X,"DATA FOR CONSTRAINED OPTIMIZATION-----")
25300 ****
25400 C
25500 C     THE FOLLOWING LINE SHOULD BE INSERTED FOR SECOND APPROCH,
25600 C4189  FORMAT(5X,"OUTPUT OBTAINED BY EVALUATING PHI FROM EQUATION
25700 C     4.21)
25800 C
25900 ****
26000 C
26100 C
26200 CALL CONSX(NOPT,M,K,ITMAX,ALPHA,BETA,GAMMA,DELTA,XSTART,R,F,IT,
26300 1IEV2,NX,G,H,XC,IPRIN,DASHI,ONETH,TWOTH,BRACE1,BRACE2,
26400 1BRACE3,C50,C51,C52,C53,C54,C55,C556,C56,CT1,C7,THETR,DUMMY,XFD,
26500 1C65,SAI)
26600 IF(IT=ITMAX) 20,20,30
26700 20  IF(IPRIN.EQ.0) GO TO 4501
26800 WRITE(NX,2117) F(IEV2)
26900 2117 FORMAT(//5X,"FINAL FUNCTION VALUE =",D18.10,//)
27000 DO 300 J=1,NOPT
27100 WRITE(NX,16)J,XSTART(IEV2, J)
27200 16  FORMAT(2X,"X("I2,') = "E20.8)
27300 300 CONTINUE
27400 4501 GO TO 899
27500 30  WRITE(NX,17)ITMAX
27600 17  FORMAT(2X,"WARNING MAIN;THE NUMBER OF ITERATIONS HAS EXCEEDED"
27700 12X,I4)
27800 C
27900 ****
28000 C
28100 899  DO 568 KI=1,NOPT
28200 X(KI)=XSTART(IEV2,KI)
28300 568  CONTINUE
28400 999  DASHI1=DASHI
28500 XZER=X(7)
28600 DO 167 KI=1,4
28700 XTEMP(KI)=X(KI+7)
28800 167  CONTINUE
28900 NNON=NOPT+1
29000 X(NNON)=CONFR(TWOTH,XTEMP,XZER)
29100 CALL NONLIN(NNON,X,NUMSIG,MAXIT,IPRIN,EPS,DASHI1,NX,DELNON,PREC,
29200 1ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,C556,
29300 1C56,C7,CT1,C65,DUMMY,THETR,SAI,XFD,X66)
29400 DO 1111 IK=1,NNON
29500 IF(X(IK).LT.0.0) X(IK)=0.0
29600 1111 CONTINUE
29700 V=X(8)*DUMMY4/X(7)
29800 AIW=X(2)
29900 AM=X(4)

```

```

30000      AIP=0.0
30100      ANUMBR=X(7)/DUMMY3
30200      F23=X(12)*(DUMMY4**TWOOTH)/DUMMY3
30300      C22=C2*V*AIP
30400      C33=C3/(V**ONETH)
30500      C44=C4/V
30600      C78=C7*(V**TWOOTH)*AIW/(AM+F23)
30700      C79=C7*AM*AIW/(AM+F23)
30800      C
30900      ****
31000      C
31100      C      INSERT THE FOLLOWING LINES IN PLACE OF THE ENCLOSED SET OF LINES
31200      C      GIVEN ABOVE FOR THE SECOND APPROACH.
31300      C
31400      C899    DO 568 KI=1,5
31500      C      X(KI)=XSTART(IEV2,KI)
31600      C568    CONTINUE
31700      C      X667=XSTART(IEV2,6)
31800      C      DO 569 KI=7,NOPT
31900      C      X(KI-1)=XSTART(IEV2,KI)
32000      C      XFD(KI-1)=XFD(KI)
32100      C569    CONTINUE
32200      C999    DASHI1=DASHI
32300      C      XZER=X(6)
32400      C      DO 167 KI=1,4
32500      C      XTEMP(KI)=X(KI+6)
32600      C167    CONTINUE
32700      C      DAK=CONFR(ONETH,XTEMP,XZER)
32800      C      CALL GOLD(X01,X02,C,X667,DAK,XZER,SAI,C65,DUMMY4,ONETH)
32900      C      NNON=NOPT
33000      C      X(NNON)=CONFR(TWOOTH,XTEMP,XZER)
33100      C998    X66=X667
33200      C      CALL NONLIN(NNON,X,NUMSIG,MAXIT,IPRIN,EPS,DASHI1,NX,DELNON,PREC,
33300      C      1ONETH,TWOOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,C556,
33400      C      1C56,C7,CT1,C65,DUMMY,THETR,SAI,XFD,X66)
33500      C      DO 1111 KI=1,NNON
33600      C      IF(X(KI).LT.0.0) X(KI)=0.0
33700      C1111   CONTINUE
33800      C      XZER=X(6)
33900      C      DO 168 KI=1,4
34000      C      XTEMP(KI)=X(KI+6)
34100      C168    CONTINUE
34200      C      DAK=CONFR(ONETH,XTEMP,XZER)
34300      C      CALL GOLD(X01,X02,C,XMAX,DAK,XZER,SAI,C65,DUMMY4,ONETH)
34400      C      X667=XMAX
34500      C      DIFF=X667-X66
34600      C      IF(IPRIN.EQ.0) GO TO 4938
34700      C      WRITE(NX,4086) DIFF,X667
34800      C4086   FORMAT(/5X,"DIFFERANCE IN X66 = ",D15.8/5X,"X66 = ",D15.8/)
34900      C4938   IF(ABS(DIFF).LT.0.0001) GO TO 987

```

```

35000 C      GO TO 998
35100 C987    V=X(7)*DUMMY4/X(6)
35200 C      AIW=X(2)
35300 C      AM=X(4)
35400 C      AIP=0.0
35500 C      ANUMBR=X(6)/DUMMY3
35600 C      F23=X(11)*(DUMMY4**TWOOTH)/DUMMY3
35700 C      C22=C2*V*IP
35800 C      C33=C3/(V**ONETH)
35900 C      C44=C4/V
36000 C      C78=C7*(V**TWOOTH)*AIW/(AM+F23)
36100 C      C79=C7*AM*AIW/(AM+F23)
36200 C
36300 C*****CALL LIN(NUMEQ,FFD,ANUMBR,THETR,C22,C33,C78,C79,C44,NX,DASHI2,
36400      1NTEMP3,B,IPRIN)
36500      DIFF=DASHI2-DASHI1
36600      IF(IPRIN.EQ.0) GO TO 4502
36700      WRITE(NX,4002) DIFF
36800      FORMAT(/2X,"DIFF   ",E15.8)
36900 4002    IF(ABS(DIFF).LT.ZETA) GO TO 98
37000 4502    DASHI=DASHI2
37100      DASHI2
37200      GO TO 999
37300 98      IF(IPRIN.EQ.0) GO TO 4503
37400      WRITE(NX,2017)
37500 2017    FORMAT(/" FUNCTION VALUES AFTER FIRST SET")
37600      WRITE(NX,2018)((I,X(I)),I=1,NNON)
37700 2018    FORMAT(1(5X,"X(",I2,") = ",E15.8))
37800 4503    DO 187 IK=1,NTEMP3
37900      KI=IK
38000      XRAD(KI-1)=B(KI,1)
38100 187     CONTINUE
38200      IF(IPRIN.EQ.0) GO TO 4504
38300      WRITE(NX,2019)((KI,XRAD(KI)),KI=0,NTEMP3)
38400 2019    FORMAT(1(5X"XRAD(",I2,") = ",E15.8))
38500 C      LIVE POLYMER CHAIN LENGTH DISTRIBUTION STARTS.
38600 4504    NTEMP2=NTEMP3-1
38700      PHI=X(6)
38800 C
38900 C*****C
39000 C
39100 C      INSERT THE FOLLOWING LINE INSTEAD OF THE PREVIOUS LINE FOR
39200 C      SECOND APPROACH.
39300 C      PHI=X66
39400 C
39500 C*****C
39600 C
39700      DO 878 KI=1,NTEMP2
39800      XLIVE(0,KI)=KI*XRAD(KI)
39900 878     CONTINUE

```

```

40000      DO 857 MONT=1,4
40100      MOMENT=MONT
40200      CALL SET2(NTEMP3,C22,C33,C44,C78,C79,C63,C102,C101,TRP,
40300      1MOMENT,PHI,THETVE,FDLIV,XRAD,XLIVE,T,B)
40400      M=1
40500      CALL MATIN(T,NTEMP2,B,M,DETER)
40600      DO 857 KI=1,NTEMP2
40700      XLIVE(MOMENT,KI)=B(KI,1)
40800 857   CONTINUE
40900      DO 862 MONT=0,4
41000      IF(IPRIN.EQ.0) GO TO 4505
41100      WRITE(NX,4250)
41200 4250   FORMAT(5X,'LIVE POLYMER CHAIN LENGTH DISTRIBUTION_____')
41300      WRITE(NX,2499) MONT
41400 2499   FORMAT(5X,'MOMENT =',I2)
41500      WRITE(NX,2500)(KI,XLIVE(MONT,KI),KI=1,NTEMP2)
41600 2500   FORMAT(2(5X,I2,5X,E15.8))
41700 862    CONTINUE
41800 C      DEAD POLYMER CHAIN LENGTH DISTRIBUTION STARTS.
41900 4505   DO 911 MOMENT=0,4
42000      DO 911 IRAD=1,NTEMP3
42100      GSMAL(MOMENT,IRAD)=XLIVE(MOMENT,IRAD)/XRAD(IRAD)
42200 911    CONTINUE
42300      DO 912 MONT=0,4
42400      MOMENT=MONT
42500      CALL SET3(NTEMP3,C22,C33,C78,C44,C102,C63,C403,C404,V,
42600      1TRP,PHI,THETVE,XRAD,FDDED,XLIVE,GSMAL,MOMENT,T,B)
42700      M=1
42800      CALL MATIN(T,NTEMP3,B,M,DETER)
42900      DO 912 KI=1,NTEMP3
43000      XDEAD(MOMENT,KI-1)=B(KI,1)
43100 912    CONTINUE
43200 C
43300 C
43400      IF(IPRIN.EQ.0) GO TO 4506
43500      WRITE(NX,4251)
43600 4251   FORMAT(5X,'DEAD POLYMER CHAIN LENGTH DISTRIBUTION_____')
43700      DO 963 MONT=0,4
43800      WRITE(NX,2499) MONT
43900      WRITE(NX,2500) (KI,XDEAD(MONT,KI),KI=0,NTEMP2)
44000 963    CONTINUE
44100 4506   DO 872 MOMENT=0,4
44200      SUM1=0.0
44300      DO 871 KI=1,NTEMP2
44400      SUM1=SUM1+XLIVE(MOMENT,KI)
44500 871    CONTINUE
44600      XLIVET(MOMENT)=SUM1
44700 872    CONTINUE
44800      DO 874 MOMENT=0,4
44900      SUM2=0.0

```

```

45000      DO 873 KI=0,NTEMP2
45100      SUM2=SUM2+XDEAD(MOMENT,KI)
45200 873    CONTINUE
45300      XDEADT(MOMENT)=SUM2
45400      TYPE *,SUM2
45500 874    CONTINUE
45600      AMNWT=AMW*XDEADT(1)/XDEADT(0)
45700
45800      AMWWT= AMW*XDEADT(2)/XDEADT(1)
45900      PDI=AMWWT/AMNWT
46000      VAVE=X(8)*DUMMY4*VM/X(7)
46100 C
46200 C*****
46300 C
46400      DAVE2=((VAVE*6.0/3.142)**ONETH)*(1.0D+08)
46500      DAVE1=(CONFR(ONETH,XTEMP,XZER)*(DUMMY4**ONETH)*((6.0*VM/3.142)
46600      1**ONETH)/X(7))*1.00D+08
46700      PATNUM=X(7)*AMZERO*ANA/DUMMY3
46800 C
46900 C*****
47000 C
47100 C      INSERT THE FOLLOWING LINES IN PLACE OF THE SET OF LINES
47200 C      ENCLOSED ABOVE FOR THE SECOND APPROACH.
47300 C      VAVE=X(7)*DUMMY4*VM/X(6)
47400 C      DAVE2=((VAVE*6.0/3.142)**ONETH)*(1.0D+08)
47500 C      DAVE1=(CONFR(ONETH,XTEMP,XZER)*(DUMMY4**ONETH)*((6.0*VM/3.142)
47600 C      1**ONETH)/X(6))*1.00D+08
47700 C      PATNUM=X(6)*AMZERO*ANA/DUMMY3
47800 C
47900 C*****
48000 C
48100 C      CONVER=(XFD(3)-X(3))/XFD(3)
48200 C      RATE=(XFD(3)-X(3))*AIOW*3.6E+06/(THETA*DUMMY2)
48300 C
48400 C
48500 C
48600 C
48700 C      OUTPUT SECTION BEGINS_____
48800      WRITE(NX,3186)
48900 3186    FORMAT(1H1)
49000      WRITE(NX,3185)
49100 3185    FORMAT(1X,'*****')
49200      *****')
49300      WRITE(NX,2147)
49400 2147    FORMAT(/10X,'OUTPUT FROM THE PROGRAM.'//)
49500      WRITE(NX,3185)
49600      WRITE(NX,2148) PATNUM,VAVE,DAVE2,DAVE1,CONVER,RATE
49700 2148    FORMAT(/5X,'(1) NUMBER OF PARTICLES (/CC EMULSION) ____',E15.8/
49800      15X,'(2) AVERAGE VOLUME OF PARTICLES (CC) ____',E15.8/
49900      15X,'(3) AVERAGE DIA ;VOLUME BASIS ("A) ____',E15.8/

```

```

50000      15X,'(4) AVERAGE DIA ;NUMBER BASIS ("A")_____,E15.8//'
50100      15X,'(5) CONVERSION OBTAINED_____,'E15.8/
50200      15X,'(6) RATE (MOLES/(LIT.HOUR))_____,E15.8//'
50300      WRITE(NX,2149) AMNWT,AMWWT,PDI
50400 2149   FORMAT(5X,'(7) NUMBER AVERAGE MOLECULAR WEIGHT_____,E15.8/
50500      15X,'(8) WEIGHT AVERAGE MOLECULAR WEIGHT_____,E15.8/
50600      15X,'(9) POLYDISPERSITY INDEX_____,E15.8//')
50700      WRITE(NX,3185)
50800      IF(IPT.EQ.0) GO TO 4510
50900      WRITE(NX,3186)
51000      WRITE(NX,3187)
51100 3187   FORMAT(/5X,'DETAILS OF THE DISTRIBUTIONS FOLLOW_____/)
51200      WRITE(NX,3185)

51300 C
51400 C
51500      IWHICH=0
51600      XL1=1.0
51700      PART1=5000.0
51800      CALL DISTR(XL1,PART1,XLIVET,IWHICH,XCORD,YCORD,ZCORD,XCUM)
51900      WRITE(NX,2150)
52000 2150   FORMAT(/5X,'LIVE POLYMER CHAIN LENGTH DISTRIBUTION')
52100      WRITE(NX,3185)
52200      WRITE(NX,2151)
52300 2151   FORMAT(/5X,'NO',5X,'MOL. WT.',5X,'NUMB.BASIS',5X,'WT.BASIS
1(NORMAL)')
52400      SUM1=0.0
52500      PART1=XCORD(2)-XCORD(1)
52600      DO 870 KI=0,50
52700      SUM1=SUM1+ZCORD(KI)*PART1
52800
52900 870    CONTINUE
53000      DO 863 KI=0,50
53100      ZCORD(KI)=ZCORD(KI)/SUM1
53200 863    CONTINUE
53300      DO 879 KI=0,50
53400      WRITE(NX,2152) KI,XCORD(KI),YCORD(KI),ZCORD(KI)
53500 2152   FORMAT(5X,I2,3X,E11.4,3X,E11.4,3X,E11.4)
53600 879    CONTINUE
53700 C
53800 C
53900 C
54000      CALL DISTR(XL1,PART1,XDEADT,IWHICH,XCORD,YCORD,ZCORD,XCUM)
54100      SUM1=0.0
54200      PART1=XCORD(2)-XCORD(1)
54300      DO 4851 KI=0,50
54400      SUM1=SUM1+ZCORD(KI)*PART1
54500 4851   CONTINUE
54600      DO 4853 KI=0,50
54700      ZCORD(KI)=ZCORD(KI)/SUM1
54800 4853   CONTINUE
54900      WRITE(NX,3186)

```

```

55000      WRITE(NX,3188)
55100      WRITE(NX,3185)
55200 3188  FORMAT( //)
55300      WRITE(NX,2153)
55400 2153  FORMAT(5X,'DEAD POLYMER CHAIN LENGTH DISTRIBUTION')
55500      WRITE(NX,3185)
55600      WRITE(NX,2154)
55700 2154  FORMAT(5X,'NO',5X,'MOL.WT.',6X,'NUMB.BASIS',5X,'WT.BASIS
55800           1(NORMAL)')
55900      DO 891 KI=0,50
56000      WRITE(NX,2152) KI,XCORD(KI),YCORD(KI),ZCORD(KI)
56100 891   CONTINUE
56200 C
56300 C
56400      XL1=1.0
56500      PART1=100.0
56600      IWHICH=1
56700      DO 892 KI=0,4
56800      XVOL(KI)=X(KI+7)*(DUMMY4**KI)/DUMMY3
56900 892   CONTINUE
57000      CALL DISTR(XL1,PART1,XVOL,IWHICH,XCORD,YCORD,ZCORD,XCUM)
57100      DO 2176 KI=0,50
57200      XCORD(KI)=((XCORD(KI)*VM*6.0/3.142)**ONETH)*(1.0D+08)
57300 2176  CONTINUE
57400      WRITE(NX,3186)
57500      WRITE(NX,3188)
57600      WRITE(NX,3185)
57700      WRITE(NX,2155)
57800 2155  FORMAT(5X,'PARTICLE SIZE DISTRIBUTION [CUMULATIVE]')
57900      WRITE(NX,3185)
58000      WRITE(NX,2156)
58100 2156  FORMAT(5X,'NO',7X,'DIA("A")',9X,'FRACTION' )
58200      DO 890 KI=0,50
58300      WRITE(NX,2157) KI,XCORD(KI),XCUM(KI)
58400 890   CONTINUE
58500 2157  FORMAT(5X,I2,5X,E11.4,6X,E11.4)
58600 C
58700 4510  WRITE(NX,3186)
58800      STOP
58900      END
59000 C
59100 C
59200      SUBROUTINE NONLIN(N,X,NUMSIG,MAXIT,IPRIN,EPS,DASHI,NX,DELNON,
59300           1PREC,ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,
59400           1C556,C56,C7,CT1,C65,DUMMY,THETR,SAI,XFD,X66)
59500 C***** ****
59600 C
59700 C           THIS SUBROUTINE SOLVES A SET OF NONLINEAR ALGEBRAIC
59800 C           EQUATIONS. IT IS USED TO SOLVE THE EQUATIONS OF FIRST SUBGROUP
59900 C           OF GROUP 1. OTHER SUBROUTINES DIRECTLY CONNECTED WITH THIS

```

```

60000 C      SUBROUTINE ARE BACK,AMAX1,AMINI,FUNX.
60100 C      SEE REF 31 FOR DETAILS.
60200 C
60300 C      DISCRIPTION OF PRINCIPAL ARGUMENTS_____
60400 C      N_____ NUMBER OF EQUATIONS.
60500 C      X(I)_____ THE VECTOR OF VARIABLES.
60600 C      NUMSIG_____ CONVERGENCE CRITERION.
60700 C      EPS_____ CONVERGENCE CRITERION.
60800 C      MAXIT_____ MAXIMUM NUMBER OF ITERATIONS ALLOWED BY
60900 C                  THIS PROCEDURE.
61000 C      PREC_____ MINIMUM ALLOWABLE STEP SIZE FOR CALCULATING
61100 C                  PARTIAL DERIVATIVES.
61200 C      DELNON_____ MINIMUM ALLOWABLE ABSOLUTE VALUE OF PARTIAL
61300 C                  DERIVATIVES.
61400 C
61500 C*****
61600      DOUBLE PRECISION X(12),PART(12),TEMP(12),COE(12,13),XFD(11)
61700      DOUBLE PRECISION FMAX,FPLUS,F,ABSFF,HOLD,FACTOR,ETA,H
61800      DOUBLE PRECISION ABB,DERMAX,TEST,AA,CC,X66,DASHI,ONETH,TWOTH
61900      DOUBLE PRECISION BRACE1,BRACE2,BRACE3,THETR,DUMMY,SAI
62000      DOUBLE PRECISION C50,C51,C52,C53,C54,C55,C556,C56,C7,CT1,C65
62100      DIMENSION ISUB(12),LOKP(12,12)
62200      RELCON=10.0E+00**(-NUMSIG)
62300      JTEST=1
62400      DO 700 M=1,MAXIT
62500      DO 1111 KI=1,N
62600      IK=KI
62700      IF(X(IK).LE.0.0) X(IK)=0.0
62800 1111  CONTINUE
62900      LQUIT=0
63000      FMAX=0.0
63100      M1=M-1
63200      IF(IPRIN.NE.1) GO TO 9
63300      WRITE(NX,49)M1,(X(I),I=1,N)
63400 49      FORMAT(5X,'M1= ',I5/5(5X,E11.4))
63500      NW=1
63600 9       DO 10 J=1,N
63700 10      LOKP(1,J)=J
63800      DO 5555 K=1,N
63900      IF(K-1) 134,134,131
64000 131     KMIN=K-1
64100      KTEMP=K
64200      CALL BACK (KMIN,N,X,ISUB,COE,LOKP,KTEMP)
64300
64400 134     JJJ=K
64500      CALL FUNX(X,F,JJJ,DASHI,
64600      1ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,C556,
64700      1C56,C7,CT1,C65,DUMMY,THETR,SAI,XFD,X66)
64800      ABSFF=ABS(F)
64900      FMAX=AMAX1(FMAX,ABSFF)

```

```

65000 C
65100 C
65200 IF(ABS(F).GE.EPS) GO TO 1345
65300 LQUIT=LQUIT+1
65400 IF(LQUIT.NE.N) GO TO 1345
65500 GO TO 725
65600 1345 FACTOR=0.001E+00
65700 135 ITALLY=0
65800 DO 200 I=K,N
65900 ITEMP=LOKP(K,I)
66000 HOLD=X(ITEMP)
66100 ETA=FACTOR*ABS(HOLD)
66200 H=AMIN1(FMAX,ETA)
66300 CCC H=ETA
66400 IF(H.LT.PREC) H=PREC
66500 X(ITEMP)=HOLD+H
66600 IF(K-1) 161,161,151
66700 KTEMP=K
66800 151 CALL BACK(KMIN,N,X,ISUB,COE,LOKP,KTEMP)
66900 161 JJJ=K
67000 CALL FUNX(X,FPLUS,JJJ,DASHI,
67100 1ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,C556,
67200 1C56,C7,CT1,C65,DUMMY,THETR,SAI,XFD,X66)
67300 PART(ITEMP)=(FPLUS-F)/H
67400 X(ITEMP)=HOLD
67500 IF(ABS(PART(ITEMP)).LT.DELTA) GO TO 190
67600 C ABB=ABS(F/PART(ITEMP))
67700 IF(ABB.LE.1.E+15) GO TO 200
67800
67900
68000 190 ITALLY=ITALLY+1
68100 200 CONTINUE
68200 C NK=N-K
68300 IF(ITALLY.LE.NK) GO TO 202
68400
68500 C FACTOR=FACTOR*10.0E+00
68600 IF(FACTOR.GT.11.) GO TO 775
68700 GO TO 135
68800
68900 202 IF(K.LT.N) GO TO 203
69000 IF(ABS(PART(ITEMP)).LT.DELTA) GO TO 775
69100 NPL1=N+1
69200 COE(K,NPL1)=0.0
69300 KMAX=ITEMP
69400 GO TO 500
69500 203 KMAX=LOKP(K,K)
69600 DERMAX=ABS(PART(KMAX))
69700 KPLUS=K+1
69800 DO 210 I=KPLUS,N
69900 JSUB=LOKP(K,I)

```

```

70000      TEST=ABS(PART(JSUB))
70100      IF(TEST.LT.DERMAX) GO TO 209
70200      DERMAX=TEST
70300      LOKP(KPLUS,I)=KMAX
70400      KMAX=JSUB
70500      GO TO 210
70600 209  LOKP(KPLUS,I)=JSUB
70700 210  CONTINUE
70800      IF(ABS(PART(KMAX)).EQ.0.0) GO TO 775
70900      ISUB(K)=KMAX
71000      COE(K,N+1)=0.0
71100      DO 220 J=KPLUS,N
71200      JSUB=LOKP(KPLUS,J)
71300      COE(K,JSUB)=PART(JSUB)/PART(KMAX)
71400 C
71500      NPL1=N+1
71600      COE(K,NPL1)=COE(K,NPL1)+PART(JSUB)*X(JSUB)
71700 220  CONTINUE
71800      NPL1=N+1
71900 500  COE(K,NPL1)=(COE(K,NPL1)-F)/PART(KMAX)+X(KMAX)
72000 C     WRITE(NX,6666) K,(ISUB(IK),IK=1,12),((LOKP(IJ,JI),IJ=1,12),
72100 C     1JI=1,12)
72200 C6666 FORMAT(I2/5(2X,E15.7)/5(2X,E15.7))
72300 5555  CONTINUE
72400      NPL1=N+1
72500      X(KMAX)=COE(N,NPL1)
72600      IF(N.EQ.1) GO TO 610
72700      N1=N-1
72800      CALL BACK(N1,N,X,ISUB,COE,LOKP,K)
72900 610  IF(M-1) 650,650,625
73000 C
73100 625  DO 630 I=1,N
73200      AA=ABS(TEMP(I)-X(I))
73300      CC=ABS(X(I))*RELCON
73400
73500      IF(AA.GT.CC) GO TO 649
73600 630  CONTINUE
73700 C
73800 C
73900      JTEST=JTEST+1
74000      IF(JTEST-3) 650,725,725
74100 649  JTEST=1
74200 650  DO 660 I=1,N
74300 660  TEMP(I)=X(I)
74400 700  CONTINUE
74500      WRITE(NX, 1753)
74600 1753 FORMAT(/2X,'WARNING NONLIN;NO CONVERGENCE MAXIMUM NO OF
74700          ITERATION USED')
74800      IF(IPRIN.NE.1) GO TO 800
74900      WRITE(NX, 1763)

```

```

75000 1763  FORMAT('FUNCTION VALUE AT LAST APPROXIMATION FOLLOW')
75100
75200
75300 725   IF(IPRIN.NE.1) GO TO 800
75400
75500 7777  DO 750 IK=1,N
75600
75700
75800
75900
76000
76100 750   CONTINUE
76200
76300
76400 7788  FORMAT(3F15.8)
76500
76600 8777  WRITE(NX, 751)
76700 751   FORMAT(//5X,'CONVERGENCE HAS BEEN ACHIEVED .THE FUNCTI')
76800
76900 7515  FORMAT(5X,'VALUES AT FINAL APPROXIMATION FOLLOW'//
77000
77100
77200 8888  FORMAT(5X,'ANSWER '/3(2X,F16.8))
77300
77400 775   WRITE(NX,752)
77500 752   FORMAT(//5X,'WARNING NONLIN;MODIFIED JACOBIAN IS SINGULAR TRY ')
77600
77700 7525  FORMAT(5X,'DIFFERENT INITIAL APPROXIMATIOS ')
77800
77900
78000 4005  CONTINUE
78100
78200 4500  FORMAT(/,3X,6(2X,E11.4))
78300 C      MAXIT=M+1
78400 800   RETURN
78500
78600 C
78700 C
78800 SUBROUTINE BACK(KMIN,N,X,ISUB,COE,LOKP,K)
78900 ****
79000 C
79100 C      THIS SUBROUTINE IS CALLED FROM NONLIN.IT IS USED FOR
79200 C      BACKSOLVING THE SETS OF EQUATIONS OBTAINED IN NONLIN AT VARIOUS
79300 C      STAGES OF ITERATION.
79400 C
79500 ****
79600 DOUBLE PRECISION X(12),COE(12,13)
79700 DIMENSION ISUB(12),LOKP(12,12)
79800 DO 200 KK=1,KMIN
79900 KM=KMIN-KK+2

```

```

80000 C -
80100      KM1=KM-1
80200      KMAX=ISUB(KM1)
80300 C
80400      X(KMAX)=0.0
80500      DO 100 J=KM,N
80600      JSUB=LOKP(KM,J)
80700      KM1=KM-1
80800      X(KMAX)=X(KMAX)+COE(KM1,JSUB)*X(JSUB)
80900 100    CONTINUE
81000      KM1=KM-1
81100      NPL1=N+1
81200      X(KMAX)=X(KMAX)+COE(KM1,NPL1)
81300 200    CONTINUE
81400      RETURN
81500      END
81600 C
81700 C
81800      FUNCTION AMAX1(A,C)
81900 C***** ****
82000 C
82100 C      THIS SUBROUTINE IS CALLED FROM NONLIN. IT REPORTS THE
82200 C      GREATER OF THE TWO NUMBERS REPRESENTING THE ARGUMENTS(I.E.A AND
82300 C      C).
82400 C
82500 C***** ****
82600      DOUBLE PRECISION A,C
82700      IF(A.GT.C) C=A
82800      AMAX1=C
82900      RETURN
83000      END
83100 C
83200 C
83300      FUNCTION AMIN1(P,Q)
83400 C***** ****
83500 C
83600 C      THIS SUBROUTINE IS CALLED FROM NONLIN. IT REPORTS THE
83700 C      SMALLER OF THE TWO NUMBERS REPRESENTING THE ARGUMENTS(I.E.P AND
83800 C      Q).
83900 C
84000 C***** ****
84100      DOUBLE PRECISION P,Q
84200      IF(P.GT.Q) P=Q
84300      AMIN1=P
84400      RETURN
84500      END
84600 C
84700 C
84800      SUBROUTINE FUNX(X,RES,K,DASHI,
84900      1ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,C556,

```

```

85000      1C56,C7,CT1,C65, DUMMY,THETR,SAI,XFD)
85100 C*****
85200 C
85300 C      THIS SUBROUTINE IS CALLED FROM NONLIN. IT REPORTS THE
85400 C      VALUE OF K TH EQUATION OF THE SET OF ALGEBRAIC EQUATIONS TO
85500 C      BE SOLVED BY NONLIN.
85600 C
85700 C*****
85800      DOUBLE PRECISION X(12),XFD(12),F(6),FFD(6),FOLD(6)
85900      DOUBLE PRECISION RES,DASHI,ONETH,TWOTH
86000      DOUBLE PRECISION BRACE1,BRACE2,BRACE3,DUMMY,THETR,SAI
86100      DOUBLE PRECISION C50,C51,C52,C53,C54,C55,C556,C56,C7,CT1,C65
86200 C      DOUBLE PRECISION FZERO,FFDZER
86300 C      COMMON ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,
86400 C      1C556,C56,CT1,C7,THETR,DUMMY
86500      NMAX=4
86600 C
86700 C
86800 C
86900      W1=8.08E-04
87000      W2=47.69
87100      W3=3.807E-02
87200      W4=2.937E+06
87300      W5=7.0998E-01
87400      W6=0.63
87500      W7=0.1981
87600      FOLD(1)=8.5111E+02
87700      FOLD(2)=7.3E+06
87800      FOLD(3)=9.432E+10
87900      FOLD(4)=1.622E+15
88000 C
88100      DUMMY1=1.0E+06
88200      DUMMY2=1.0E-02
88300      DUMMY3=1.0E-03
88400      DUMMY4=4.2985E+03
88500 C
88600 C      X ARE THE OUTPUT VARIABLES FROM THE REACTOR
88700 C      XFD ARE THE FEED CHARECTARISTIC OF AFORESAID VARIABLES
88800      N=K=8
88900      FZERO=X(7)
89000      FFDZER=XFD(7)
89100      DO 210 INTEMP=1,NMAX
89200      NITEM=INTEMP
89300      F(NITEM)=X(NITEM+7)
89400      FFD(NITEM)=XFD(NITEM+7)
89500 210      CONTINUE
89600      IF(K.GT.7) GO TO 205
89700      GO TO (1,2,3,4,5,6,7),K
89800 1      CONH=CONFR(TWOTH,F,FZERO)
89900      RES=X(12)-CONH

```

```

90000      RETURN
90100 2      RES=(XFD(1)-X(1))/THETR
90200      RES=RES/W1
90300      RETURN
90400 3      RES=C56*(XFD(2)-X(2))/THETR-X(2)
90500      RES=RES/W2
90600      RETURN
90700 4      RES=(XFD(3)-X(3))/(THETR*DUMMY2)-C53*X(6)*(FZERO/DUMMY3)*DASHI
90800      RES=RES/W3
90900      RETURN
91000 5      RES=C50*X(1)-X(4)-C51-X(12)*(DUMMY4**TWOOTH)/
91100 1DUMMY3-C52*X(5)
91200      RES=RES/W4
91300      RETURN
91400 6      RES=C54*X(3)/DUMMY2-(X(5))-C55*(F(1)*DUMMY4/DUMMY3)*
91500 1X(6)-C556*X(4)
91600      RES=RES/W5
91700      RETURN
91800 7      RES=X(6)=0.615
91900      RES=RES/W6
92000      RETURN
92100 205     IF (K.GT.8) GO TO 206
92200      RES=C7*X(2)*X(4)/(X(4)+(X(12)*(DUMMY4**TWOOTH)/
92300 1DUMMY3))+(FFDZER-FZERO)/(THETR*DUMMY3)
92400 CCC      RES=RES/W7
92500      RETURN
92600 206     IF(K.GT.9)GO TO 207
92700      RES=C7*X(2)*X(4)/(X(4)+(X(12)*(DUMMY4**TWOOTH)/
92800 1DUMMY3))+CT1*DASHI*N*X(6)*FZERO/(DUMMY3*(1-X(6)))+
92900 1(FFD(N)-F(N))*(DUMMY4**N)/(THETR*DUMMY3)
93000 CCC      RES=RES/FOLD(N)
93100      RETURN
93200 207     RES=C7*X(2)*X(4)/(X(4)+(X(12)*(DUMMY4**TWOOTH)/
93300 1DUMMY3))+CT1*DASHI*N*X(6)*F(N-1)*(DUMMY4**N-1)/(DUMMY3*1(1-X(6)))+
93400 1(FFD(N)-F(N))*(DUMMY4**N)/(THETR*DUMMY3)
93500 CCC      RES=RES/FOLD(N)
93600      RETURN
93700      END
93800 C
93900 C
94000      FUNCTION CONFR(FF,YY,YY0)
94100 ***** ****
94200 C
94300 C      THE INPUT FOR THIS SUBROUTINE ARE YY AND YY0,YY(I)
94400 C      REPRESENT THE MOMENTS OF THE DISTRIBUTION(I=1,2,----).YY0
94500 C      REPRESENTS ZERO TH MOMENT OF THE DISTRIBUTION .THE FUNCTION
94600 C      RETURNS THE FF TH MOMENT OF THE DISTRIBUTION,FOR DETAILS OF
94700 C      THE PROCEDURE EMPLOYED SEE CHAPTER 4.
94800 C
94900 C***** ****

```

```

95000      DOUBLE PRECISION Y(6),YY(6),CCN(6),SSN(6),CALPHA(6),SALPHA(6)
95100      DOUBLE PRECISION YY0,AA,A,ALPHA,ALPHA1,ALKKA,ALPHN1,ALPHF1
95200      DOUBLE PRECISION ALPHFJ,ALPHJ1,FF,RATIO1
95300      NMAX=4
95400      AA=YY(2)*YY0/(YY(1)*YY(1))
95500      Y(1)=1.0/(AA-1.0)
95600      Y(2)=AA*Y(1)*Y(1)
95700      ALPHA=Y(1)-1.
95800      ALPHA1=Y(1)
95900      IF(ALPHA1.LT.0) GO TO 450
96000 C
96100 C
96200      DO 118 IK=3,NMAX
96300      KI=IK
96400      Y(KI) =((ALPHA1*YY0/YY(1))**KI)*YY(KI)/YY0
96500 118      CONTINUE
96600      ALPHF1=ALPHA+FF+1.0
96700 C
96800      DO 100 NMNM=3,NMAX
96900      N=NMM
97000      CALPHA(N)=1
97100      DO 100 KKJJ=1,N
97200      KK=KKJJ
97300      NKK=N-KK
97400      ALKKA=ALPHA+KK+1.
97500      CCN(KK)=((-1)**KK)*FACTO(N)*GAMM(ALPHA1)/(FACTO(KK)*
97600      1FACTO(NKK)*GAMM(ALKKA))
97700      CALPHA(N)=CALPHA(N)+CCN(KK)* Y(KK)
97800 100      CONTINUE
97900      DO 201 NMNM=3,NMAX
98000      N=NMM
98100      SALPHA(N)=0.0
98200      ALPHN1=N+ALPHA+1.
98300      SALPHA(N)=GAMM(ALPHN1)*GAMM(ALPHF1)/(FACTO(N)*GAMM(ALPHA1))
98400      DO 201 JJJK=1,N
98500      JJ=JJJK
98600      ALPHFJ=ALPHA+FF+JJ+1.0
98700      ALPHJ1=ALPHA+JJ+1.0
98800      ALPHN1=ALPHA+N+1.0
98900      NJJ=N-JJ

00100      SSN(JJ)=((-1.0)**JJ)*GAMM(ALPHN1)*GAMM(ALPHFJ)/(FACTO(NJJ)*
00200      1FACTO(JJ)*GAMM(ALPHJ1))
00300      SALPHA(N)=SALPHA(N)+SSN(JJ)
00400
00500
00600 201      CONTINUE
00700      A=GAMM(ALPHF1)
00800      DO 400 NMNM=3,NMAX
00900      N=NMM

```

```

01000      A=A+CALPHA(N)*SALPHA(N)
01100 400      CONTINUE
01200      A=A/GAMM(ALPHA1)
01300      CONFR=((YY(1)/(YY0*ALPHA1))**FF)*A*YY0
01400      RETURN
01500 450      RATIO1=YY(1)/YY0
01600      CONFR=YY0*(RATIO1**FF)
01700      RETURN
01800      END
01900 C
02000 C
02100      FUNCTION FACTO(N)
02200 C*****THIS FUNCTION REPORTS THE VALUE OF FACTORIAL OF AN
02300 C
02400 C      THIS FUNCTION REPORTS THE VALUE OF FACTORIAL OF AN
02500 C      INTEGER VARIABLE N.
02600 C
02700 C*****THIS FUNCTION REPORTS THE VALUE OF GAMMA FUNCTION OF
02800      FACTO=1
02900      IF(N,EQ.0) GO TO 2
03000      DO 1 JK=1,N
03100      FACTO=FACTO*JK
03200 1      CONTINUE
03300 2      RETURN
03400      END
03500 C
03600 C
03700      FUNCTION GAMM(XX)
03800 C*****THIS FUNCTION REPORTS THE VALUE OF GAMMA FUNCTION OF
03900 C
04000 C      THIS FUNCTION REPORTS THE VALUE OF GAMMA FUNCTION OF
04100 C      THE VARIABLE XX.
04200 C
04300 C*****DOUBLE PRECISION XX,X,Y
04400      DOUBLE PRECISION XX,X,Y
04500      IF(XX-57.0) 6,6,4
04600
04700 4      TER=2
04800      GAMM=1.E+30
04900      RETURN
05000 6      X=XX
05100      ERR=1.0E-06
05200      TER=0
05300      GAMM=1.0
05400      TF(X-2.0)50,50,15
05500 10     IF(X-2.0) 110,110,15
05600 15     X=X-1.
05700      GAMM=GAMM*X
05800      GO TO 10
05900 50     IF(X-1.0) 60,120,110

```

```

06000 C SEE IF X IS NEAR NEGATIVE INTEGER OR ZERO
06100 60 IF(X=ERR) 62,62,80
06200 62 Y=FLOAT(INT(X))-X
06300 IF (ABS(Y)=ERR) 130,130,64
06400 64 IF(1.0-Y=ERR) 130,130,70
06500 C X NOT NEAR A NEGATIVE INTEGER OR ZERO
06600 70 IF(X=1.0) 80,80,110
06700 80 GAMM=GAMM/X
06800 X=X+1.0
06900 GO TO 70
07000 110 Y=X-1.0
07100 GY=1.0+Y*(-0.5771017+Y*(+0.9858540+Y*(-0.8764218+Y*
07200 1(+0.8328212+Y*(-0.5684729+Y*(0.2548205+Y*(-.05149930)))))))
07300 GAMM=GAMM*GY
07400 120 RETURN
07500 130 IER=1
07600 RETURN
07700 END
07800 C
07900 C
08000 SUBROUTINE SET1 (N,FFD,ANUMBR,THETR,C22,C33,C78,C79,C44,T,B)
08100 ****
08200 C
08300 C THIS SUBROUTINE CALCULATES THE ELEMENTS OF COEFFICIENT
08400 C MATRIX OF THE SET LINEAR EQUATIONS IN SECOND SUBGROUP OF THE
08500 C GROUP 1. COEFFICIENT MATRIX IS REPRESENTED BY T.B(I,1) IS THE
08600 C VECTOR OF CONSTANTS .
08700 C
08800 ****
08900 DOUBLE PRECISION T(25,25),B(25, 1),FFD(0:20)
09000 DOUBLE PRECISION ANUMBR,THETR,C22,C33,C78,C79,C44,THETVE,BRACES5
09100 N1=N-1
09200 THETIVE=1./THETR
09300 DO 104 IROW=1,N
09400 DO 104 ICOL=1,N
09500 T(IROW,ICOL)=0.0
09600 104 CONTINUE
09700 C
09800 DO 105 IROW=1,N
09900 IF(IROW.EQ.N) GO TO 433
10000 ICOL=IROW
10100 ICOLM2=ICOL-2
10200 ICOLM1=ICOL-1
10300 ICOLP1=ICOL+1
10400 ICOLP2=ICOL+2
10500 C
10600 NEST=ICOLM2
10700 IF(NEST.LE.-1) GO TO 102
10800 IF(NEST.LE.0) GO TO 103
10900 T(IROW,NEST)=C22

```

```

11000 103      NEST=ICOLM1
11100          T(IROW,NEST)=C78
11200 102      NEST=ICOL
11300          T(IROW,NEST)==C78-C33*(IROW-1)-C44*(IROW-1)*(IROW-2)-THETVE
11400          1-C22
11500          NEST=ICOLP1
11600          IF(NEST.GT.N) GO TO 105
11700          T(IROW,NEST)=C33*ICOL
11800          NEST=ICOLP2
11900          IF(NEST.GT.N) GO TO 105
12000          T(IROW,NEST)=ICOL*ICOLP1*C44
12100          GO TO 105
12200 433      DO 434 KI=1,N
12300          IK=KI
12400          T(IROW,IK)=1.0
12500 434      CONTINUE
12600 105      CONTINUE
12700 C
12800          BRACE5=THETR*ANUMBR
12900          DO 439 IROW=1,N
13000          IF(IROW.EQ.N) GO TO 438
13100          IF(IROW.EQ.2) GO TO 437
13200          B(IROW,1)=-FFD(IROW-1)/BRACE5
13300          GO TO 439
13400 437      B(IROW,1)=-FFD(IROW-1)/BRACE5-C79/ANUMBR
13500          GO TO 439
13600 438      B(IROW,1)=+1.0
13700 439      CONTINUE
13800          RETURN
13900          END
14000 C
14100 C
14200          SUBROUTINE LIN(NUMEQ,FFD,ANUMBR,THETR,C22,C33,C78,C79,C44,NX,
14300          1DASHI2,NTEMP3,B,IPRIN)
14400 C*****C*****C*****C*****C*****C*****C*****C*****C*****C*****
14500 C
14600 C          THIS SUBROUTINE SOLVES THE EQUATIONS OF SECOND SUBGROUP
14700 C          OF THE FIRST GROUP OF EQUATIONS. INITIALLY 'NUMEQ' EQUATIONS ARE
14800 C          SOLVED BY CALLING SET1 TO OBTAIN COEFFICIENT MATRIX AND MATIN
14900 C          TO THIS EQUATIONS. THE ANSWER IS OBTAINED AS B(I,1) I=1,---NUMEQ.
15000 C          THE LAST TWO VALUES OF THE SET ARE TESTED TO BE LESS THAN
15100 C          1.0E-07 AND THE PROCESS IS REPEATED BY INCREASING THE NUMBER OF
15200 C          EQUATIONS BY 2. FOR DETAILS SEE CHAPTER 4.
15300 C
15400 C*****C*****C*****C*****C*****C*****C*****C*****C*****
15500          DOUBLE PRECISION T(25,25),B(25, 1),FFD(0:20)
15600          DOUBLE PRECISION ANUMBR,THETR,C22,C33,C78,C79,C44,DASHI2
15700          DOUBLE PRECISION SUM3,SUM4
15800          NTEMP=NUMEQ
15900 334      CALL SET1(NTEMP,FFD,ANUMBR,THETR,C22,C33,C78,C79,C44,T,B)

```

```

16000      M=1
16100      CALL MATIN(T,NTEMP,B,M,DETER)
16200      NTEMP1=NTEMP-1
16300      NTEMP2=NTEMP-2
16400      IF(B(NTEMP1,1).GE.1.0E-07) GO TO 338
16500      IF(B(NTEMP2,1).GE.1.0E-07) GO TO 338
16600      NTEMP3=NTEMP-3
16700      IF(IPRIN.EQ.0) GO TO 339
16800      WRITE(NX,1119) (NTEMP3,(B(IK,1),IK=1,NTEMP3))
16900 1119      FORMAT(5X,'ANSWER FROM LIN'//5X,'NTEMP3= ',I2//,2(5X,E15.8 ))
17000      GO TO 339
17100 338      NTEMP=NTEMP+2
17200      GO TO 334
17300 339      SUM3=0.0
17400      ,SUM4=0.0
17500      DO 335 IK=1,NTEMP3
17600      KI=IK
17700      SUM3=SUM3+B(KI,1)*(KI-1)
17800      SUM4=SUM4+B(KI,1)
17900 335      CONTINUE
18000      DASHI2=SUM3/SUM4
18100      RETURN
18200      END
18300 C
18400 C
18500      SUBROUTINE MATIN(A,N,B,M,DETER)
18600 C***** ****
18700 C
18800 C      THIS SUBROUTINE INVERTS THE COEFFICIENT MATRIX OF A SET
18900 C      OF LINEAR EQUATIONS.A REPRESENTS THE COEFFICIENT MATRIX .N IS
19000 C      THE NUMBER OF EQUATIONS.PUT M=1 IF ONLY SOLUTION OF EQUATIONS IS
19100 C      NEEDED.THE SOLUTION IS REPRESENTED BY B(I,1),I=1,----,N .
19200 C
19300 C***** ****
19400      DOUBLE PRECISION B(25,1)
19500      DOUBLE PRECISION A(25,25)
19600      DOUBLE PRECISION AMAX,SWAP,PIVOT,DETRM,T
19700      DIMENSION IPIVT(40),INDEX(40,2)
19800      EQUIVALENCE(IROW,JROW),(ICOLM,JCOLM),(AMAX,T,SWAP)
19900      10      DETER=1.0
20000      15      DO 20 J=1,N
20100      20      IPIVT(J)=0
20200      30      DO 550 I=1,N
20300      40      AMAX=0.0
20400      45      DO 105 J=1,N
20500      50      IF(IPIVT(J)-1) 60,105,60
20600      60      DO 100 K=1,N
20700      70      IF(IPIVT(K)-1) 80,100,740
20800      80      IF(AMAX-ABS(A(J,K))) 85,100,100

```

```

00900    85  IROW=J
01000    90  JCOLUMN
01100   100  CONTINUE
01200   105  CONTINUE
01300   110  IPIVT(ICOLM)=IPIVT(ICOLM)+1
01400   130  IF(IROW=ICOLM) 140,260,140
01500   140  DETER=-DETER
01600   150  DO 200 L=1,N
01700   160  SWAP=A(IROW,L)
01800   170  A(IROW,L)=A(ICOLM,L)
01900   200  A(ICOLM,L)=SWAP
02000   205  IF(M) 260,260,210
02100   210  DO 250 L=1,M
02200   220  SWAP=B(IROW,L)
02300   230  B(IROW,L)=B(ICOLM,L)
02400   250  B(ICOLM,L)=SWAP
02500   260  INDEX(I,1)=IROW
02600   270  INDEX(I,2)=ICOLM
02700   310  PIVOT=A(ICOLM,ICOLM)
02800   320  DETERM=DETER*PIVOT
02900   330  A(ICOLM,ICOLM)=1.0
03000   340  DO 350 L=1,N
03100   350  A(ICOLM,L)=A(ICOLM,L)/PIVOT
03200   355  IF(M) 380,380,360
03300   360  DO 370 L=1,M
03400   370  B(ICOLM,L)=B(ICOLM,L)/PIVOT
03500   380  DO 550 L1=1,N
03600   390  IF(L1=ICOLM) 400,550,400
03700   400  T=A(L1,ICOLM)
03800   420  A(L1,ICOLM)=0.0
03900   430  DO 450 L=1,N
04000   450  A(L1,L)=A(L1,L)-A(ICOLM,L)*T
04100   455  IF(M) 550,550,460
04200   460  DO 500 L=1,M
04300   500  B(L1,L)=B(L1,L)-B(ICOLM,L)*T
04400   550  CONTINUE
04500   600  DO 710 I=1,N
04600   610  L=N+1-I
04700   620  IF(INDEX(L,1)=INDEX(L,2)) 630,710,630
04800   630  JROW=INDEX(L,1)
04900   640  JCOLUMN=INDEX(L,2)
05000   650  DO 705 K=1,N
05100   660  SWAP=A(K,JROW)
05200   670  A(K,JROW)=A(K,JCOLUMN)
05300   700  A(K,JCOLUMN)=SWAP
05400   705  CONTINUE
05500   710  CONTINUE
05600   710  DO 11 K=1,N
05700           IF(IPIVT(K)=1) 12,11,12
05800   11  CONTINUE

```

```

10900      SUM1=0.0
11000      DO 467 KI1=0,MOMENT
11100      KI=KI1
11200      IMONT=MOMENT-KI
11300      SUM1=SUM1+FACTO(MOMENT)*GSMAL(KI,IRAD)*GSMAL(IMONT,IRAD)/
11400      1(FACTO(IMONT)*FACTO(KI))
11500 467    CONTINUE
11600      TERM(IRAD)=C403*SUM1*XRAD(IRAD)/V
11700 468    CONTINUE
11800      DO 9 IROW=1,NTEMP3
11900      IF(IROW.GT.1) GO TO 7
12000      B(IROW,1)==TERM(2)=C404*XLIVE(MOMENT,2)/V

00100      1=FDDED(MOMENT,0)*THETVE
00200      GO TO 9
00300 7      B(IROW,1)==TERM(IROW+1)=C404*IROW*
00400      1XLIVE(MOMENT,IROW+1)/V=C63*TRP*XLIVE(MOMENT,IROW-1)
00500      1=C102*PHI*XLIVE(MOMENT,IROW-1)-FDDED(MOMENT,IROW-1)*THETVE
00600 9      CONTINUE
00700      RETURN
00800      END
00900 C      SUBROUTINE SET2(NTEMP3,C22,C33,C44,C78,C79,C63,C102,C101,TRP,
01000      1MOMENT,PHI,THETVE,FDLIV,XRAD,XLIVE,T,B)
01300 C*****THIS SUBROUTINE CALCULATES THE COEFFICIENT MATRIX OF THE
01400 C      SET OF SET LINEAR EQUATIONS OF GROUP 2.(I.E. LIVE POLYMER CHAIN
01500 C      LENGTH DISTRIBUTION) COEFFICIENT MATRIX REPRESENTED T.VECTOR OF
01600 C      CONSTANTS IS B(I,1) .
01700 C
01800 C
01900 C
02000 C*****DOUBLE PRECISION T(25,25),B(25,1),XRAD(0:20)
02100      DOUBLE PRECISION T(25,25),B(25,1),XRAD(0:20)
02200      DOUBLE PRECISION FDLIV(5,20),XLIVE(0:5,20),TERM(20)
02300      DOUBLE PRECISION C22,C33,C44,C78,C63,C101,C102,PHI,TRP,THETVE
02400      DOUBLE PRECISION SUM1,C79
02500      NTEMP2=NTEMP3-1
02600      DO 4 IROW=1,NTEMP2
02700      DO 4 ICOL=1,NTEMP2
02800      T(IROW,ICOL)=0.0
02900 4      CONTINUE
03000 C      DO 5 IROW=1,NTEMP2
03100      ICOL=IROW
03200      ICOLM2=ICOL-2
03300      ICOLM1=ICOL-1
03400      ICOLP1=ICOL+1
03500      ICOLP2=ICOL+2
03600
03700 C

```

```

03800      NEST=ICOLM2
03900      IF(NEST.LE.-1) GO TO 2
04000      IF(NEST.LE.0)  GO TO 3
04100      T(IROW,NEST)=C22
04200  3    NEST=ICOLM1
04300      T(IROW,NEST)=C78
04400  2    NEST=ICOL
04500      T(IROW,NEST)==C78-C22-C33*ICOL=C63*TRP-C102*PHI-C44*ICOL*ICOLM1
04600      1=THETVE
04700      NEST=ICOLP1
04800      IF(NEST.GT.NTEMP2) GO TO 5
04900      T(IROW,NEST)=C33*ICOL
05000      NEST=ICOLP2
05100      IF(NEST.GT.NTEMP2) GO TO 5
05200      T(IROW,NEST)= C44*ICOLP1*ICOL
05300  5    CONTINUE
05400      DO 468 IRAD1=1,NTEMP2
05500      IRAD=IRAD1
05600      SUM1=0.0
05700      DO 467 KI1=1,MOMENT
05800      KI=KI1
05900      IMONT=MOMENT-KI
06000      SUM1=SUM1+FACTO(MOMENT)*XLIVE(IMONT,IRAD)/(FACTO(KI)*
06100      1FACTO(IMONT ))
06200  467  CONTINUE
06300      TERM(IRAD)=C101*PHI*SUM1
06400  468  CONTINUE
06500  C
06600      DO 9 IROW=1,NTEMP2
06700      IF(IROW.GT.1) GO TO 7
06800      B(IROW,1) ==C78*XRAD(0)-C102*PHI*1.0*XRAD(1)-THETVE*
06900      1FDLIV(MOMENT,1)-TERM(1)=C63*TRP*1.0*XRAD(1)-C79
07000      GO TO 9
07100  7    B(IROW,1)==C78*XRAD(IROW-1)-2.0*C22*XRAD(IROW-2)-C102*PHI*
07200      1IROW*XRAD(IROW)-THETVE*FDLIV(MOMENT,IROW)-TERM(IROW)
07300      1-C63*TRP*IROW*XRAD(IROW)
07400  9    CONTINUE
07500      RETURN
07600      END
07700  C
07800  C
07900      FUNCTION DIFF2(AMV,V)
08000  C***** ****
08100  C
08200  C      THE INPUT FOR THIS FUNCTION CONSISTS OF AMV AND V,AMV(I)
08300  C      ARE THE MOMENTS OF A DISTRIBUTION,(I=0,1,2----) THE FUNCTION
08400  C      RETURNS THE VALUE OF DISTRIBUTION FUNCTION EVALUATED AT V.
08500  C      FOR DETAILS SEE CHAPTER 4.
08600  C
08700  C***** ****

```

```

08800      DOUBLE PRECISION AMV(0:5),GX(6),CALPHA(6),ALPHAL(6)
08900 C      AMV ARE FIRST FOUR MOMENTS OF DISTRIBUTION
09000 C      AMVZER IS THE ZEROTH ABSOLUTE MOMENT OF DISTRIBUTION
09100 C      X IS THE X CO-ORDINATE OF WHICH
09200 C      THE FUNCTION IS TO BE EVALUATED,
09300 C      RESULT IS THE FUNCTION VALUE CALCULATED AT
09400 C      AFORESAID X CO-ORDINATE @VE,
09500 C
09600 C      INPUT PORTION-----
09700 C      CONVERT MV INTO GX(I,E, MOMENTS OF NORMALIEED
09800 C      DISTRIBUTION.)
09900 C
10000      DOUBLE PRECISION AMVZER,PDI,ALPHA,ALPHA1,ALPIT1,ALPIK1,XMINU,SUM
10100      DOUBLE PRECISION V
10200      AMVZER=AMV(0)
10300      PDI=AMV(2)*AMVZER/(AMV(1)*AMV(1))
10400      GX(1)=1.0/(PDI-1.0)
10500      GX(2)=PDI*GX(1)*GX(1)
10600      ALPHA=GX(1)-1.0
10700      ALPHA1=GX(1)
10800      NMAX=4
10900      DO 118 KI=3,NMAX
11000      IK=KI
11100      GX(IK)=((ALPHA1*AMVZER/AMV(1))**IK)*AMV(IK)/AMVZER
11200 118  CONTINUE
11300 C      CONVERT ABSOLUTE CO-ORDINATE V TO THE CORRESSPONDING
11400 C      NORMALISED CO-ORDINATE X,
11500 X=(ALPHA1*AMVZER/AMV(1))*V
11600 C      CALCULATE CALPHA.
11700      DO 100 KI=3,NMAX
11800      IK=KI
11900      CALPHA(IK)=1.0
12000      DO 100 ITAM=1,IK
12100      ITEMP=ITAM
12200      IKITAM=IK-ITEMP
12300      ALPIT1=ALPHA1+ITEMP
12400      CALPHA(IK)=CALPHA(IK)+((-1)**ITEMP)*FACTO(IK)*GAMM(ALPHA1)*
12500      1GX(ITEMP)/(FACTO(IKITAM)*FACTO(ITEMP)*GAMM(ALPIT1))
12600 100  CONTINUE
12700 C      CALCULATION OF MODIFIED LAGURE@ POLYNOMIAL
12800      DO 200 KI=3,NMAX
12900      IK=KI
13000      ALPIK1=ALPHA1+IK
13100      ALPHAL(IK)=GAMM(ALPIK1)/(GAMM(ALPHA1)*FACTO(IK))
13200      DO 200 ITAM=1,IK
13300      ITEMP=ITAM
13400      IKITAM=IK-ITEMP
13500      ALPIT1=ALPHA1+ITEMP
13600      ALPHAL(IK)=ALPHAL(IK)+((-1)**ITEMP)*GAMM(ALPIK1)*
13700      1(X**ITEMP)/(FACTO(IKITAM)*GAMM(ALPIT1)*FACTO(ITEMP))

```

```

13800 200  CONTINUE
13900 C   CALCULATION OF NORMALISED DISTRIBUTION FUNCTON.
14000 SUM=1.0
14100 DO 300 KI=3,NMAX
14200 IK=KI
14300 SUM=SUM+CALPHA(IK)*ALPHAL(IK)
14400 300  CONTINUE
14500 XMINU=-X
14600 GNORM=SUM*EXP(XMINU)*(X**ALPHA)/GAMM(ALPHA1)
14700 C   CALCULATION OF VALUE OF TJE ABSOLUTE FUNX.
14800 DIFF2=GNORM*ALPHA1*AMVZER*AMVZER/AMV(1)
14900 RETURN
15000 END
15100 C
15200 C
15300 C
15400 C
15500 SUBROUTINE CENTR(N,M,K,IEV1,I,XC,X,K1)
15600 ****
15700 C
15800 C   THIS SUBROUTINE IS REQUIRED FOR OPTIMIZATION, IT
15900 C   CALCULATES THE COEFFICIENTS OF THE CENTROID OF THE COMPLEX.
16000 C
16100 ****
16200 DOUBLE PRECISION X(K,M),XC(N)
16300 DO 20 J=1,N
16400 XC(J)=0.0
16500 DO 10 IL=1,K1
16600 XC(J)=XC(J)+X(IL,J)
16700 10  CONTINUE
16800 RK=K1
16900 XC(J)=(XC(J)-X(IEV1,J))/(RK-1.0)
17000 20  CONTINUE
17100 RETURN
17200 END
17300 C
17400 C
17500 SUBROUTINE CHECK(N,M,K,X,G,H,I,KODE,XC,DELTA,K1)
17600 ****
17700 C
17800 C   THIS SUBROUTINE IS REQUIRED FOR OPTIMIZATION, IT CHECKS
17900 C   IF A GIVEN POINT SATISFIES THE GIVEN SET OF CONSTRAINTS AND
18000 C   AND CALCULATES A FEASIBLE VERTEX FOR THE COMPLEX.
18100 C
18200 ****
18300 DOUBLE PRECISION X(K,M),G(M),H(M),XC(N)
18400 10  KT=0
18500 CALL CONST(N,M,K,X,G,H,I)
18600 DO 50 J=1,N
18700 IF(X(I,J)-G(J)) 20,20,30

```

```

18800    20   X(I,J)=G(J)+DELTA
18900          GO TO 50
19000    30   IF(H(J)=X(I,J)) 40,40,50
19100    40   X(I,J)=H(J)-DELTA
19200    50   CONTINUE
19300          NO=5
19400 C      WRITE (NO,*) ((J,X(1,J),XC(J),X(2,J)) ,J=1,N)
19500          IF(KODE) 110,110,60
19600    60   NN=N+1
19700          DO 100 J=NN,M
19800          CALL CONST(N,M,K,X,G,H,I)
19900          IF(X(I,J)=G(J)) 80,70,70
20000    70   IF(H(J)=X(I,J)) 80,100,100
20100    80   IEV1=I
20200          KT=1
20300          CALL CENTR(N,M,K,IEV1,I,XC,X,K1)
20400          DO 90 JJ=1,N
20500          X(I,JJ)=(X(I,JJ)+XC(JJ))/2.0
20600    90   CONTINUE
20700   100   CONTINUE
20800          IF(KT) 110,110,10
20900   110   RETURN
21000          END
21100 C
21200 C
21300          SUBROUTINE CONSX(N,M,K,ITMAX,ALPHA,BETA,GAMMA,DELTA,X,R,F,
21400          1IT,IEV2,NO,G,H,XC,IPRIN,DASHI,ONETH,TWOTH,BRACE1,BRACE2,
21500          1BRACE3,C50,C51,C52,C53,C54,C55,C556,C56,CT1,C7,THETR,DUMMY,XFD,
21600          1C65,SA1)
21700 ****
21800 C
21900 C          THIS SUBROUTINE IS USED TO FIND MAXIMA/MINIMA OF A
22000 C          CONSTRAINED MULTIVARIABLE FUNCTION BY BOX'S METHOD. IT IS USED
22100 C          TO FIND INITIAL GUESSES TO SOLVE THE EQUATIONS OF FIRST SUBGROUP
22200 C          OF GROUP 1. FOR DETAILS SEE CHAPTER 4 AND REF 32
22300 C
22400 C          DISCRIPTION OF PRINCIPAL VARIABLES
22500 C          ALPHA-----COMPLEX EXPANSION PARAMETER.
22600 C                      (USUALLY 1.3)
22700 C          BETA-----CONVERGENCE CRITERION
22800 C          GAMMA-----CONVERGENCE CRITERION
22900 C          DELTA-----USED TO FIND A FEASIBLE POINT
23000 C                      WHEN THE CALCULATED VERTEX OF TH
23100 C                      COMPLEX IS FOUND TO BE OUTSIDE
23200 C                      ANY CONSTRAINT.
23300 C          N-----NUMBER OF VARIABLES.
23400 C          M-----NUMBER OF CONSTRAINTS.
23500 C          K-----NUMBER OF VERTICES.
23600 C          ITMAX-----MAXIMUM NUMBER OF ITERATIONS
23700 C                      ALLOWED FOR OPTIMIZATION.

```

```

23800 C      R----- RANDOM NUMBERS.
23900 C      F----- OBJECTIVE FUNCTION.
24000 C      X(I)----- VECTOR OF VARIABLES
24100 C
24200 ****
24300      DOUBLE PRECISION ONEFH,TWOTH,BRACE1,BRACE3,BRACE2,THETR,DUMMY
24400      DOUBLE PRECISION C50,C51,C52,C53,C54,C55,C556,C56,C7,CT1,C65
24500      DOUBLE PRECISION DASHI,DASHI1,DASHI2,SAI
24600      INTEGER GAMMA
24700      DOUBLE PRECISION X(K,M),F(K),G(M),H(M),XC(N),XFD(N)
24800      DIMENSION R(K,N)
24900      IT=1
25000      IPRIN1=IPRIN
25100      KODE=0
25200      IF(M=N)20,20,10
25300      10      KODE=1
25400      20      CONTINUE
25500      DO 40 II=2,K
25600      DO 30 J=1,N
25700      X(II,J)=0.0
25800      30      CONTINUE
25900      40      CONTINUE
26000      DO 65 II=2,K
26100      DO 50 J=1,N
26200      I=II
26300      CALL CONST(N,M,K,X,G,H,I)
26400      X(II,J)=G(J)+R(II,J)*(H(J)-G(J))
26500      50      CONTINUE
26600      K1=II
26700      CALL CHECK(N,M,K,X,G,H,I,KODE,XC,DELTA,K1)
26800      IF(II=2)51,51,55
26900      51      IF(IPRIN)52,65,52
27000      52      WRITE(NO,018)
27100      18      FORMAT(1H1//2X,"COORDINATES OF INITIAL COMPLEX")
27200      IO=1
27300      WRITE (NO,159) (IO,J,X(IO,J),J=1,N)
27400      159     FORMAT(/3(2X,2HX,,I2,1H,,I2,4H) = ,E13,6))
27500      55      IF(IPRIN) 56,65,56
27600      56      WRITE (NO,159) (II,J,X(II,J),J=1,N)
27700      65      CONTINUE
27800      K1=K
27900      DO 70 IL=1,K
28000      I=IL
28100      CALL FUNC(N,M,K,X,F,I,DASHI,BRACE1,BRACE2,
28200      1BRACE3,C50,C51,C52,C53,C54,C55,C556,C56,CT1,C7,THETR,DUMMY,XFD,
28300      1C65,SAI)
28400      70      CONTINUE
28500      KOUNT=1
28600      IA=0
28700      IF(IPRIN) 72,80,72

```

```

28800 72      WRITE(NO,21)
28900 21      FORMAT(/2X,"VALUES OF THE FUNCTION")
29000      WRITE(NO,22) (J,F(J),J=1,K)
29100 22      FORMAT(/,3(2X,2HF(,I2,4H) = ,E13.6))
29200 80      IEV1=1
29300      DO 100 ICM=2,K
29400      IF(F(IEV1)=F(ICM)) 100,100,90
29500 90      IEV1=ICM
29600 100     CONTINUE
29700      IEV2=1
29800      DO 120 ICM=2,K
29900      IF(F(IEV2)=F(ICM)) 110,110,120
30000 110     IEV2=ICM
30100 120     CONTINUE
30200      IF(F(IEV2)=(F(IEV1)+BETA)) 140,130,130
30300 130     KOUNT=1
30400      GO TO 150
30500 140     KOUNT=KOUNT+1
30600      IF(KOUNT=GAMMA) 150,240,240
30700 150     CALL CENTR(N,M,K,IEV1,I,XC,X,K1)
30800      DO 160 JJ=1,N
30900      X(IEV1,JJ)=(1.0+ALPHA)*XC(JJ)-ALPHA*X(IEV1,JJ)
31000 160     CONTINUE
31100      I=IEV1
31200      CALL CHECK(N,M,K,X,G,H,I,KODE,XC,DELTA,K1)
31300      CALL FUNC(N,M,K,X,F,I,DASHI,BRACE1,BRACE2,
31400      1BRACE3,C50,C51,C52,C53,C54,C55,C556,C56,CT1,C7,THETR,DUMMY,XFD,
31500      1C65,SAI)
31600 170     IEV2=1
31700      DO 190 ICM=2,K
31800      IF(F(IEV2)=F(ICM)) 190,190,180
31900 180     IEV2=ICM
32000 190     CONTINUE
32100      IF(IEV2-IEV1) 220,200,220
32200 200     DO 210 JJ=1,N
32300      X(IEV1,JJ)=(X(IEV1,JJ)+XC(JJ))/2.0
32400 210     CONTINUE
32500      I=IEV1
32600      CALL CHECK(N,M,K,X,G,H,I,KODE,XC,DELTA,K1)
32700      CALL FUNC(N,M,K,X,F,I,DASHI,BRACE1,BRACE2,
32800      1BRACE3,C50,C51,C52,C53,C54,C55,C556,C56,CT1,C7,THETR,DUMMY,XFD,
32900      1C65,SAI)
33000      GO TO 170
33100 220     CONTINUE
33200      IF(IT/50*50.NE.IT) IPRIN=0
33300      MTYPE=1
33400      IF(IT/50*50.NE.IT) MTYPE=0
33500      IF(MTYPE.EQ.0) GO TO 4511
33600  C       IF(IT.GT.300) IPRIN=1
33700      TYPE *,IT

```

```

33800 4511 IF(IPRIN) 230,228,230
33900 230 WRITE(NO,23)IT
34000 23 FORMAT(//,2X,'ITERATION NUMBER',I5)
34100 23 WRITE(NO,24)
34200 24 FORMAT(/,2X,'CO-ORDINATES OF CORRECTED POINT')
34300 24 WRITE(NO,159) (IEV1,JC,X(IEV1,JC),JC=1,N)
34400 24 WRITE(NO,21)
34500 24 WRITE(NO,22)(I,F(I),I=1,K)
34600 24 WRITE(NO,25)
34700 25 FORMAT(/,2X,'CO-ORDINATES OF THE CENTROID')
34800 25 WRITE(NO,26)(JC,XC(JC),JC=1,N)
34900 26 FORMAT(/,3(2X,2HX(,I2,6H,C) = ,E13.6))
35000 228 IT=IT+1
35100 228 IPRIN=IPRIN1
35200 240 IF(IT=ITMAX) 80,80,240
35300 240 RETURN
35400 240 END
35500 C
35600 C
35700 SUBROUTINE CONST(N,M,K,X,G,H,I)
35800 DOUBLE PRECISION X(14,16),G(16),H(16)
35900 ****
36000 C
36100 C      THIS SUBROUTINE DEFINES VARIOUS EXPLICIT AND IMPLICIT
36200 C      CONSTRAINTS.
36300 C      G(I)-----LOWER CONSTRAINTS.
36400 C      H(I)-----UPPER CONSTRAINTS.
36500 C      I=1,---N-----EXPLICIT CONSTRAINTS.
36600 C      I=N+1,---M-----IMPLICIT CONSTRAINTS.
36700 C
36800 ****
36900 G(1)=0.0
37000 H(1)=4.9
37100 G(2)=0.0
37200 H(2)=1.0
37300 G(3)=0.0
37400 H(3)=1.92
37500
37600      G(4)=1.0E-06
37700 H(4)=1.0E+06
37800 G(5)=1.0E-06
37900 H(5)=1.0E+06
38000 G(6)=.01
38100 H(6)=1.0
38200 G(7)=1.0E-06
38300 H(7)=1.0E+03
38400 G(8)=1.0E-06
38500 H(8)=1.0E+04
38600 G(9)=1.0E-06
38700 H(9)=1.0E+05

```

```

38800      G(10)=1.0E-06
38900      H(10)=1.0E+06
39000      G(11)=1.0E-07
39100      H(11)=1.0E+07
39200      IF(N=M) 1,2,1
39300  1    X(I,12)=X(I,7)*X(I,9)/X(I,8)**2
39400      G(12)=1.0
39500      H(12)=10.0
39600      X(I,13)=X(I,8)/X(I,7)
39700      G(13)=0.5
39800      H(13)=10.0
39900      X(I,14)=X(I,9)/X(I,8)
40000      G(14)=1.0
40100      H(14)=10
40200      X(I,15)=X(I,10)/X(I,9)
40300      G(15)=1.0
40400      H(15)=10.0
40500      X(I,16)=X(I,11)/X(I,10)
40600      G(16)=1.0
40700      H(16)=10.0
40800  2    RETURN
40900      END
41000  C
41100  C
41200      SUBROUTINE FUNC (INN,M,K,XDUMY,F,J,DASHI,BRACE1,BRACE2,
41300      1BRACE3,C50,C51,C52,C53,C54,C55,C556,C56,CT1,C7,THETR,DUMMY,XFD)
41400  ****
41500  C
41600  C      THIS SUBROUTINE RETURNS THE VALUE OF THE OBJECTIVE
41700  C      FUNCTION TO BE OPTIMIZED. THE OBJECTIVE FUNCTION IS MADE FROM
41800  C      R.H.S. OF EQUATIONS OF FIRST SUBGROUP OF GROUP 1.
41900  C
42000  ****
42100      DOUBLE PRECISION XDUMY(K,M),F(K),XFD(11),Z(6),FFD(6),FOLD(6)
42200      DOUBLE PRECISION X(25),RUMY(6)
42300      DOUBLE PRECISION ONEFH,TWOTH,BRACE1,BRACE2,THETR,DUMMY
42400      DOUBLE PRECISION C50,C51,C52,C53,C54,C55,C556,C56,C7,CT1
42500      DOUBLE PRECISION DASHI,RES,FZERO,FFDZER,FIRST,SECOND,THIRD
42600      DOUBLE PRECISION FOURTH,FIFTH,SIX,SEVENT,SUM
42700      NMAX=4
42800  C
42900  C
43000      ONEFH=1.0/3.
43100      TWOTH=2.0/3.
43200  C
43300      W1=8.08E-04
43400      W2=47.69
43500      W3=3.807E-02
43600      W4=2.937E+06
43700      W5=7.0998E-01

```

```

43800      W6=0.63
43900      W7=0.1981
44000      FOLD(1)=8.5111E+02
44100      FOLD(2)=7.3E+06
44200      FOLD(3)=9.432E+10
44300      FOLD(4)=1.622E+15
44400 C
44500      DUMMY1=1.0E+06
44600      DUMMY2=1.0E-02
44700      DUMMY3=1.0E-03
44800      DUMMY4=4.2985E+03
44900 C
45000 C      X ARE THE OUTPUT VARIABLES FROM THE REACTOR
45100 C      XFD ARE THE FEED CHARECTARISTIC OF AFORESAID VARIABLES
45200      DO 450 IK=1,INN
45300      KI=IK
45400      X(KI)=XDUMMY(J,KI)
45500 450      CONTINUE
45600 C
45700      N=K=7
45800      FZERO=X(7)
45900      FFDZER=XFD(7)
46000      DO 210 INTEMP=1,NMAX
46100      NTEM=INTEMP
46200      Z(NTEM)=X(NTEM+7)
46300      FFD(NTEM)=XFD(NTEM+7)
46400 210      CONTINUE
46500 1      RES=(XFD(1)-X(1))/THETR
46600      FIRST=RES/W1
46700 2      RES=C56*(XFD(2)-X(2))/THETR-X(2)
46800      SECOND=RES/W2
46900 3      RES=(XFD(3)-X(3))/(THETR*DUMMY2)-C53*X(6)*(FZERO/DUMMY3)*DASHI
47000      THIRD=RES/W3
47100 4      RES=C50*X(1)-X(4)-C51-CONFR(TWOTH,Z,FZERO)*(DUMMY4**TWOTH)/
47200      1DUMMY3-C52*X(5)/DUMMY1
47300      FOURTH=RES/W4
47400 5      RES=C54*X(3)/DUMMY2-(X(5)/DUMMY1)-C55*(Z(1)*DUMMY4/DUMMY3)*
47500      1X(6)-C556*X(4)
47600      FIFTH=RES/W5
47700 6      RES=X(6)-0.615
47800      SIX=RES/W6
47900      RES=C7*X(2)*X(4)/(X(4)+(CONFR(TWOTH,Z,FZERO)*(DUMMY4**TWOTH)/
48000      1DUMMY3))+(FFDZER-FZERO)/(THETR*DUMMY3)
48100 C
48200 *****C*****C*****C*****C*****C*****C*****C*****C*****C*****
48300 C
48400 C      INSERT THIS LINE INSTEAD OF PREVIOUS LINE FOR SECOND APPROACH
48500 C6      RES=1.0-X(6)+DLOG(X(6))+SAI*(1.0-X(6))+C65*X(7)/
48600 C      1*(CONFR(ONETH,Z,FZERO)*(DUMMY4*ONETH))
48700 C

```

```

48800 C*****
48900 C
49000 SEVENT=RES/W7
49100 DO 650 N=1,NMAX
49200 IF(N.GT.1) GO TO 207
49300 RES=C7*X(2)*X(4)/(X(4)+(CONFR(TWOTH,Z,FZERO)*(DUMMY4**TWOTH)/
49400 1DUMMY3))+CT1*DASHI*N*X(6)*FZERO/(DUMMY3*(1-X(6)))+
49500 1(FFD(N)=Z(N))*(DUMMY4**N)/(THETR*DUMMY3)
49600 RUMY(N)=RES/FOLD(N)
49700 GO TO 650
49800 207 RES=C7*X(2)*X(4)/(X(4)+(CONFR(TWOTH,Z,FZERO)*(DUMMY4**TWOTH)/
49900 1DUMMY3))+CT1*DASHI*N*X(6)*Z(N-1)*(DUMMY4**N-1)/(DUMMY3*1
50000 1(1-X(6)))+(FFD(N)=Z(N))*(DUMMY4**N)/(THETR*DUMMY3)
50100 RUMY(N)=RES/FOLD(N)
50200 650 CONTINUE
50300 SUM=0.0
50400 DO 550 IK=1,NMAX
50500 KI=IK
50600 SUM=SUM+ABS(RUMY(KI))
50700 550 CONTINUE
50800 F(J)=SUM+ABS(FIRST)+ABS(SECOND)+ABS(THIRD)+ABS(FOURTH)+1
50900 ABS(FIFTH)+ABS(SIX)+ABS(SEVENT)
51000 F(J)==F(J)
51100 RETURN
51200 END
51300 C
51400 C
51500 SUBROUTINE QG10(XL,XU,Y,AMV)
51600 C*****
51700 C
51800 C
51900 C THIS SUBROUTINE RETURNS THE VALUE OF INTEGRAL OF A
52000 C FUNCTION EVALUATED IN THE RANGE XL TO XU. IT IS USED FOR
52100 C CALCULATING THE CUMULATIVE DISTRIBUTION FROM A DIFFERENTIAL
52200 C DISTRIBUTION.
52300 C*****
52400 DOUBLE PRECISION Y,XL,XU
52500 DOUBLE PRECISION AMV(0:5)
52600 DOUBLE PRECISION A,B,C
52700 A=0.5*(XU-XL)+XL
52800 B=(XU-XL)
52900 C=0.4869533*B
53000 Y=0.03333567*(DIFF2(AMV,A+C)+DIFF2(AMV,A-C))
53100 C=0.4325317*B
53200 Y=Y+0.07472567*(DIFF2(AMV,A+C)+DIFF2(AMV,A-C))
53300 C=0.3397048*B
53400 Y=Y+0.1095432*(DIFF2(AMV,A+C)+DIFF2(AMV,A-C))
53500 C=0.2166977*B
53600 Y=Y+0.1346334*(DIFF2(AMV,A+C)+DIFF2(AMV,A-C))
53700 C=0.07443717*B

```

```

53800      Y=B*(Y+0.1477621*(DIFF2(AMV,A+C)+DIFF2(AMV,A-C)))
53900      RETURN
54000      END
54100      C
54200      C
54300      SUBROUTINE DISTR(XL1,PART1,AMV,IWHICH,XCORD,YCORD,ZCORD,XCUM)
54400      ****
54500      C
54600      C      THIS SUBROUTINE GENERATES THE DISTRIBUTION FUNCTION FROM
54700      C      ITS MOMENT. INITIALLY THE FUNCTION IS EVALUATED AT STEPS OF
54800      C      PART1 STARTING WITH THE LOWER LIMIT OF XL1. THE USEFUL RANGE OF
54900      C      DISTRIBUTION IS THEN CALCULATED. THE USEFUL RANGE IS THEN
55000      C      DIVIDED INTO 50 INTERVALS AND THE DISTRIBUTION FUNCTION IS
55100      C      CALCULATED AT THESE 51 POINTS. XCORD(I) IS THE SET OF X
55200      C      CO-ORDINATES AT WHICH THE FUNCTION CALCULATED. YCORD(I) IS THE
55300      C      SET OF FUNCTION VALUES AT THESE POINTS ON NUMBER BASIS.
55400      C      ZCORD(I) IS THE SET OF FUNCTION VALUES ON WT. BASIS. XCUM
55500      C      REPRESENTS THE SET OF NORMALISED CUMULATIVE DISTRIBUTION
55600      C      FUNCTION CALCULATED AT THE AFORESAID POINTS
55700      C      IWHICH=0----NORMALISED CUMULATIVE DISTRIBUTION IS NOT CALCULATED
55800      C
55900      C
56000      ****
56100      DOUBLE PRECISION XCORD(0:50),YCORD(0:50),ZCORD(0:50),XCUM(0:50)
56200      DOUBLE PRECISION STAR(50),AMV(0:5)
56300      DOUBLE PRECISION XL,XL1,XU1,XU,PART1,PART2,Y,YMAX
56400      DOUBLE PRECISION RATIO,AREA,TOTAL
56500      XL=XL1
56600      C      TO FIND USEFUL RANGE OF DISTRIBUTION.
56700      YMAX=0.0
56800      1      Y=DIFF2(AMV,XL)
56900      IF(Y.GT.YMAX) GO TO 15
57000      RATIO=Y/YMAX
57100      IF(RATIO.GT.0.0001)GO TO 16
57200      GO TO 2
57300      15     YMAX=Y
57400      16     XL=XL+PART1
57500      GO TO 1
57600      C      UPPER USEFUL LIMIT FOR THE DISTRIBUTION HAS BEEN OBTAINED
57700      2      XU1=XL
57800      PART2=(XU1-XL1)/50.0
57900      XL=XL1
58000      C      TO FIND DIFFERENTIAL AND CUMULATIVE DISTRIBUTION
58100      XCORD(0)=XL1
58200      YCORD(0)=DIFF2(AMV,XL1)
58300      ZCORD(0)=XCORD(0)*YCORD(0)
58400      XCUM(0)=0.0
58500      AREA=0.0
58600      DO 3 KI=1,50
58700      XU=XL+PART2

```

```

58800      XCORD(KI)=XU
58900      YCORD(KI)=DIFF2(AMV,XU)
59000      ZCORD(KI)=XCORD(KI)*YCORD(KI)
59100      IF(IWHICH.EQ.0) GO TO 6
59200      CALL QG10(XL,XU,Y,AMV)
59300      STAR(KI)=Y
59400      AREA=AREA+STAR(KI)
59500      6      XL=XU
59600      3      CONTINUE
59700      IF(IWHICH.EQ.0) GO TO 5
59800      C      TO CALCULATE NORMALISED CUMULATIVE DISTRIBUTION
59900      TOTAL=AREA
60000      AREA=0.0
60100      DO 4 KI=1,50
60200      AREA=AREA+STAR(KI)
60300      XCUM(KI)=AREA/TOTAL
60400      4      CONTINUE
60500      5      RETURN
60600      END
60700      C
60800      C
60900      SUBROUTINE FUNXX(X,RES,K,DASHI,
61000      1ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,C556,
61100      1C56,C7,CT1,C65, DUMMY,THETR,SAI,XFD,X66)
61200      C***** THIS SUBROUTINE IS SIMILAR TO FUNX, FOR SECOND APPROACH
61300      C
61400      C      CHANGE THE NAME OF SUBROUTINE TO FUNX.
61500      C***** DOUBLE PRECISION X(13),XFD(13),F(6),FFD(6),FOLD(6)
61600      DOUBLE PRECISION X(13),XFD(13),F(6),FFD(6),FOLD(6)
61700      DOUBLE PRECISION RES,DASHI,ONETH,TWOTH
61800      DOUBLE PRECISION BRACE1,BRACE2,BRACE3,DUMMY,THETR,SAI
61900      DOUBLE PRECISION C50,C51,C52,C53,C54,C55,C556,C56,C7,CT1,C65
62000      DOUBLE PRECISION FZERO,FFDZER,X66
62100      C      COMMON ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,
62200      C      1C556,C56,CT1,C7,THETR,DUMMY
62300      NMAX=4
62400      C
62500      C
62600      C
62700      W1=8.08E-04
62800      W2=47.69
62900      W3=3.807E-02
63000      W4=2.937E+06
63100      W5=7.0998E-01
63200      W6=0.63
63300      W7=0.1981
63400      FOLD(1)=8.5111E+02
63500      FOLD(2)=7.3E+06
63600      FOLD(3)=9.432E+10
63700      FOLD(4)=1.622E+15

```

```

63800 C
63900 DUMMY1=1.0E+06
64000 DUMMY2=1.0E-02
64100 DUMMY3=1.0E-03
64200 DUMMY4=4.2985E+03
64300 C
64400 C X ARE THE OUTPUT VARIABLES FROM THE REACTOR
64500 C XFD ARE THE FEED CHARECTARISTIC OF AFORESAID VARIABLES
64600 N=K=7
64700 FZERO=X(6)
64800 FFDZER=XFD(6)
64900 DO 210 INTEMP=1,NMAX
65000 NTEM=INTEMP
65100 F(NTEM)=X(NTEM+6)
65200 FFD(NTEM)=XFD(NTEM+6)
65300 210 CONTINUE
65400 IF(K.GT.6) GO TO 205
65500 GO TO (1,2,3,4,5,6),K
65600 1 CONH=CONFR(TWOTH,F,FZERO)
65700 RES=X(11)-CONH
65800 RETURN
65900 2 RES=(XFD(1)-X(1))/THETR
66000 RES=RES/W1
66100 RETURN
66200 3 RES=C56*(XFD(2)-X(2))/THETR-X(2)
66300 RES=RES/W2
66400 RETURN
66500 4 RES=(XFD(3)-X(3))/(THETR*DUMMY2)-C53*X66*(FZERO/DUMMY3)*DASHI
66600 RES=RES/W3
66700 RETURN
66800 5 RES=C50*X(1)-X(4)-C51-X(11)*(DUMMY4**TWOTH)/
66900 1DUMMY3-C52*X(5)
67000 RES=RES/W4
67100 RETURN
67200 6 RES=C54*X(3)/DUMMY2-(X(5))-C55*(F(1)*DUMMY4/DUMMY3)*
67300 1X66-C556*X(4)
67400 RES=RES/W5
67500 RETURN
67600 205 IF (K.GT.7) GO TO 206
67700 RES=C7*X(2)*X(4)/(X(4)+(X(11)*(DUMMY4**TWOTH)/
67800 1DUMMY3))+FFDZER-FZERO)/(THETR*DUMMY3)
67900 CCC RES=RES/W7
68000 RETURN
68100 206 IF(K.GT.8)GO TO 207
68200 RES=C7*X(2)*X(4)/(X(4)+(X(11)*(DUMMY4**TWOTH)/
68300 1DUMMY3))+CT1*DASHI*N*X66*FZERO/(DUMMY3*(1-X66))+
68400 1*(FFD(N)-F(N))*(DUMMY4**N)/(THETR*DUMMY3)
68500 CCC RES=RES/FOLD(N)
68600 RETURN
68700 207 RES=C7*X(2)*X(4)/(X(4)+(X(11)*(DUMMY4**TWOTH)/

```

```

68800      1DUMMY3))+CT1*DASHI*N*X66*F(N-1)*(DUMMY4**N)/(DUMMY3*
68900      1(1-X66))+FFD(N)=F(N))*(DUMMY4**N)/(THETR*DUMMY3)
69000 CCC   RES=RES/FOLD(N)
69100      RETURN
69200      END
69300 C
69400 C
69500      SUBROUTINE GOLD(X01,X02,C,XMAX,D,X7,SAI,C65,DUMMY4,ONETH)
69600 C*****THIS SUBROUTINE SOLVES EQN.4.21 BY GOLDEN SECTION METHOD
69700 C
69800 C      THIS SUBROUTINE SOLVES EQN.4.21 BY GOLDEN SECTION METHOD
69900 C      FOR OPTIMIZATION.
70000 C*****
70100      DOUBLE PRECISION X01,X02,C,XMAX,D,X7,SAI,C65,C1,ONETH
70200      DOUBLE PRECISION X3,X4,TEMPX2,Y3,Y4,X1,X2
70300      C1=C/10.0
70400      IF(ABS(X01-X02).LT.C) GO TO 100
70500      X1=X01
70600      X2=X02
70700      X3=X1+0.3820*(X2-X1)
70800      X4=X2-0.3820*(X2-X1)
70900      CALL OBJECT(Y3,X3,D,X7,SAI,C65,DUMMY4,ONETH)
71000      CALL OBJECT(Y4,X4,D,X7,SAI,C65,DUMMY4,ONETH)
71100 50    IF(Y3.GE,Y4) GO TO 30
71200      X1=X3
71300      IF(ABS(X1-X2).LT.C) GO TO 100
71400      TEMPX2=X1+0.3820*(X2-X1)
71500      IF(ABS(TEMPX2-X4).GT., C1) GO TO 40
71600      Y3=Y4
71700      GO TO 45
71800 40    X3=TEMPX2
71900      CALL OBJECT(Y3,X3,D,X7,SAI,C65,DUMMY4,ONETH)
72000 45    X4=X2-0.3820*(X2-X1)
72100      CALL OBJECT(Y4,X4,D,X7,SAI,C65,DUMMY4,ONETH)
72200      GO TO 50
72300
72400 30    X2=X4
72500      IF(ABS(X1-X2).LT.C) GO TO 100
72600      TEMPX2=X2-0.3820*(X2-X1)
72700      IF(ABS(TEMPX2-X3).GT., C1) GO TO 60
72800      Y4=Y3
72900      GO TO 65
73000 60    X4= TEMPX2
73100      CALL OBJECT(Y4,X4,D,X7,SAI,C65,DUMMY4,ONETH)
73200 65    X3=X1+0.3820 *(X2-X1)
73300      CALL OBJECT(Y3,X3,D,X7,SAI,C65,DUMMY4,ONETH)
73400      GO TO 50
73500 100   XMAX=(X1+X2)/2.0
73600      CALL OBJECT(YMAX,XMAX,D,X7,SAI,C65,DUMMY4,ONETH)
73700      YMAXX=YMAX

```

```
73800      RETURN
73900      END
74000 C
74100 C
74200 C
74300 C
74400      SUBROUTINE OBJECT(YOPT,XOPT,D,X7,SAI,C65,DUMMY4,ONETH)
74500 C***** ****
74600 C
74700 C      THIS SUBROUTINE REPRESENTS THE OBJECTIVE FUNCTION TO BE
74800 C      OPTIMIZED FOR THE SOLUTION OF EQN. 4.21.
74900 C
75000 C***** ****
75100 C
75200      DOUBLE PRECISION YOPT,XOPT,D,X7,SAI,C65,ONETH
75300      YOPT=1.0-XOPT+DLOG(XOPT)+SAI*(1.0-XOPT)*(1.0-XOPT)+C65*X7/
75400      1(D*(DUMMY4**ONETH))
75500      YOPT=-ABS(YOPT)
75600      RETURN
75700      END
```

DATA FOR THE PROBLEM

DATA FOR CONSTRAINED OPTIMIZATION

NOPT = 11 M = 16 K = 14 ITMAX = 2000 IPRIN = 0
ALPHA = 1.300000 BETA = 0.000100 DELTA = 0.001000
GAMMA = 5 DASHI = 0.900000

INITIAL GUESSES

XSTART(1) = 0.4000000E+01 XSTART(2) = 0.9900000E+00
XSTART(3) = 0.1630000E+01 XSTART(4) = 0.9900000E+00
XSTART(5) = 0.9900000E+00 XSTART(6) = 0.6400000E+00
XSTART(7) = 0.9900000E+00 XSTART(8) = 0.1400000E+01
XSTART(9) = 0.3000000E+01 XSTART(10) = 0.6000000E+01
XSTART(11) = 0.2400000E+02

INPUT PARAMETERS FOR LINEAR EQUATIONS AND THE CONVERGENCE CRITERION

MAXIT = 200 NUMSIG = 6 IPT = 1
NUMEQ = 6
EPS = 0.1000000E-07 DELNON = 0.1000000E-06 PREC = 0.5000000E-05
ZETA = 0.1000000E-08

RATE CONSTANTS

KP = 0.2650000E+06 KD = 0.2330000E-04 KZERO = 0.0000000E+00
KIP = 0.0000000E+00 KTD = 0.0000000E+00 KTC = 0.7200000E+10
KFT = 0.0000000E+00 KFM = 0.2120000E+02

MONOMER-POLYMER CHARECTARISTICS

MW = 0.1040000E+03 DP = 0.10526000E+01 DM = 0.8780000E+00
VMOL = 0.11845000E+03 SAI = 0.2560000E+00

REACTOR CHARECTARISTICS

THETA = 0.9000000E+03 TEMP = 0.3430000E+03 VW = 0.64940000E+00

OTHER CONSTANTS

AIOW = 0.1540000E-04 AMZERO = 0.16611000E-12 VD = 0.41888000E-11
VM = 0.65449800E-19 CE = 0.36720000E+10 AMM = 0.63000000E-02
FEF = 0.10000000E+01 SWC = 0.89000000E-05 YETA = 0.75000000E+01

FEED CONDITIONS

RADICAL NUMBER DTSTRIBUTION

FFD(1) = 0.000000 FFD(2) = 0.000000
FFD(3) = 0.000000 FFD(4) = 0.000000
FFD(5) = 0.000000 FFD(6) = 0.000000
FFD(7) = 0.000000 FFD(8) = 0.000000

LIVE POLYMER CHAINS

FDLIV(0, 1) = 0.0000	FDLIV(0, 2) = 0.0000	FDLIV(0, 3) = 0.0000
FDLIV(0, 4) = 0.0000	FDLIV(0, 5) = 0.0000	FDLIV(0, 6) = 0.0000
FDLIV(0, 7) = 0.0000	FDLIV(0, 8) = 0.0000	FDLIV(0, 9) = 0.0000
FDLIV(0,10) = 0.0000	FDLIV(
FDLIV(1, 1) = 0.0000	FDLIV(1, 2) = 0.0000	FDLIV(1, 3) = 0.0000
FDLIV(1, 4) = 0.0000	FDLIV(1, 5) = 0.0000	FDLIV(1, 6) = 0.0000
FDLIV(1, 7) = 0.0000	FDLIV(1, 8) = 0.0000	FDLIV(1, 9) = 0.0000
FDLIV(1,10) = 0.0000	FDLIV(
FDLIV(2, 1) = 0.0000	FDLIV(2, 2) = 0.0000	FDLIV(2, 3) = 0.0000
FDLIV(2, 4) = 0.0000	FDLIV(2, 5) = 0.0000	FDLIV(2, 6) = 0.0000
FDLIV(2, 7) = 0.0000	FDLIV(2, 8) = 0.0000	FDLIV(2, 9) = 0.0000
FDLIV(2,10) = 0.0000	FDLIV(
FDLIV(3, 1) = 0.0000	FDLIV(3, 2) = 0.0000	FDLIV(3, 3) = 0.0000
FDLIV(3, 4) = 0.0000	FDLIV(3, 5) = 0.0000	FDLIV(3, 6) = 0.0000
FDLIV(3, 7) = 0.0000	FDLIV(3, 8) = 0.0000	FDLIV(3, 9) = 0.0000
FDLIV(3,10) = 0.0000	FDLIV(
FDLIV(4, 1) = 0.0000	FDLIV(4, 2) = 0.0000	FDLIV(4, 3) = 0.0000
FDLIV(4, 4) = 0.0000	FDLIV(4, 5) = 0.0000	FDLIV(4, 6) = 0.0000
FDLIV(4, 7) = 0.0000	FDLIV(4, 8) = 0.0000	FDLIV(4, 9) = 0.0000
FDLIV(4,10) = 0.0000	FDLIV(
FDLIV(5, 1) = 0.0000	FDLIV(5, 2) = 0.0000	FDLIV(5, 3) = 0.0000
FDLIV(5, 4) = 0.0000	FDLIV(5, 5) = 0.0000	FDLIV(5, 6) = 0.0000
FDLIV(5, 7) = 0.0000	FDLIV(5, 8) = 0.0000	FDLIV(5, 9) = 0.0000
FDLIV(5,10) = 0.0000	FDLIV(

DEAD POLYMER CHAINS

FDDED(0, 1) = 0.0000	FDDED(0, 2) = 0.0000	FDDED(0, 3) = 0.0000
FDDED(0, 4) = 0.0000	FDDED(0, 5) = 0.0000	FDDED(0, 6) = 0.0000
FDDED(0, 7) = 0.0000	FDDED(0, 8) = 0.0000	FDDED(0, 9) = 0.0000
FDDED(0,10) = 0.0000	FDDED(
FDDED(1, 1) = 0.0000	FDDED(1, 2) = 0.0000	FDDED(1, 3) = 0.0000
FDDED(1, 4) = 0.0000	FDDED(1, 5) = 0.0000	FDDED(1, 6) = 0.0000
FDDED(1, 7) = 0.0000	FDDED(1, 8) = 0.0000	FDDED(1, 9) = 0.0000
FDDED(1,10) = 0.0000	FDDED(
FDDED(2, 1) = 0.0000	FDDED(2, 2) = 0.0000	FDDED(2, 3) = 0.0000
FDDED(2, 4) = 0.0000	FDDED(2, 5) = 0.0000	FDDED(2, 6) = 0.0000
FDDED(2, 7) = 0.0000	FDDED(2, 8) = 0.0000	FDDED(2, 9) = 0.0000
FDDED(2,10) = 0.0000	FDDED(
FDDED(3, 1) = 0.0000	FDDED(3, 2) = 0.0000	FDDED(3, 3) = 0.0000
FDDED(3, 4) = 0.0000	FDDED(3, 5) = 0.0000	FDDED(3, 6) = 0.0000
FDDED(3, 7) = 0.0000	FDDED(3, 8) = 0.0000	FDDED(3, 9) = 0.0000
FDDED(3,10) = 0.0000	FDDED(
FDDED(4, 1) = 0.0000	FDDED(4, 2) = 0.0000	FDDED(4, 3) = 0.0000
FDDED(4, 4) = 0.0000	FDDED(4, 5) = 0.0000	FDDED(4, 6) = 0.0000
FDDED(4, 7) = 0.0000	FDDED(4, 8) = 0.0000	FDDED(4, 9) = 0.0000
FDDED(4,10) = 0.0000	FDDED(
FDDED(5, 1) = 0.0000	FDDED(5, 2) = 0.0000	FDDED(5, 3) = 0.0000

FDDED(5, 4) = 0.0000 FDDED(5, 5) = 0.0000 FDDED(5, 6) = 0.0000
FDDED(5, 7) = 0.0000 FDDED(5, 8) = 0.0000 FDDED(5, 9) = 0.0000
FDDED(5,10) = 0.0000 FDDED(

FEED COND. FOR VARIABLES OF INITIAL SET

1	0.4080E+01	2	0.1000E+01	3	0.1920E+01
4	0.0000E+00	5	0.0000E+00	6	0.0000E+00
7	0.0000E+00	8	0.0000E+00	9	0.0000E+00
10	0.0000E+00	11	0.0000E+00		

OUTPUT FROM THE PROGRAM

(1) NUMBER OF PARTICLES (/CC EMULSION) 0.11466412E+16
(2) AVERAGE VOLUME OF PARTICLES (CC) 0.26767643E-15
(3) AVERAGE DIA ;VOLUME BASIS ("A) 0.79956124E+03
(4) AVERAGE DIA ;NUMBER BASIS ("A) 0.71403866E+03

(5) CONVERSION OBTAINED 0.40438988E+00
(6) RATE (MOLES/(LIT.HOUR)) 0.47828000E+01

(7) NUMBER AVERAGE MOLECULAR WEIGHT 0.39254813E+06
(8) WEIGHT AVERAGE MOLECULAR WEIGHT 0.92157680E+06
(9) POLYDISPERSITY INDEX 0.23476785E+01

DETAILS OF THE DISTRIBUTIONS FOLLOW

LIVE POLYMER CHAIN LENGTH DISTRIBUTION

NO	MOL. WT.	NUMB.BASIS	WT.BASIS(NORMAL)
0	0.1000E+01	0.1134E-03	0.6037E-07
1	0.7010E+03	0.1116E-03	0.4164E-04
2	0.1401E+04	0.9412E-04	0.7020E-04
3	0.2101E+04	0.7868E-04	0.8801E-04
4	0.2801E+04	0.6548E-04	0.9765E-04
5	0.3501E+04	0.5434E-04	0.1013E-03
6	0.4201E+04	0.4501E-04	0.1007E-03
7	0.4901E+04	0.3722E-04	0.9712E-04
8	0.5601E+04	0.3074E-04	0.9168E-04
9	0.6301E+04	0.2537E-04	0.8511E-04
10	0.7001E+04	0.2092E-04	0.7799E-04
11	0.7701E+04	0.1725E-04	0.7071E-04
12	0.8401E+04	0.1421E-04	0.6356E-04
13	0.9101E+04	0.1171E-04	0.5673E-04
14	0.9801E+04	0.9646E-05	0.5033E-04
15	0.1050E+05	0.7947E-05	0.4443E-04
16	0.1120E+05	0.6549E-05	0.3906E-04
17	0.1190E+05	0.5399E-05	0.3421E-04
18	0.1260E+05	0.4452E-05	0.2987E-04
19	0.1330E+05	0.3673E-05	0.2601E-04
20	0.1400E+05	0.3032E-05	0.2260E-04
21	0.1470E+05	0.2504E-05	0.1960E-04
22	0.1540E+05	0.2069E-05	0.1697E-04
23	0.1610E+05	0.1711E-05	0.1467E-04
24	0.1680E+05	0.1416E-05	0.1266E-04
25	0.1750E+05	0.1172E-05	0.1092E-04
26	0.1820E+05	0.9712E-06	0.9412E-05
27	0.1890E+05	0.8052E-06	0.8103E-05
28	0.1960E+05	0.6680E-06	0.6971E-05
29	0.2030E+05	0.5545E-06	0.5993E-05
30	0.2100E+05	0.4605E-06	0.5148E-05
31	0.2170E+05	0.3826E-06	0.4420E-05
32	0.2240E+05	0.3180E-06	0.3793E-05
33	0.2310E+05	0.2644E-06	0.3252E-05
34	0.2380E+05	0.2199E-06	0.2787E-05
35	0.2450E+05	0.1829E-06	0.2386E-05
36	0.2520E+05	0.1522E-06	0.2042E-05
37	0.2590E+05	0.1266E-06	0.1746E-05
38	0.2660E+05	0.1053E-06	0.1491E-05
39	0.2730E+05	0.8758E-07	0.1273E-05
40	0.2800E+05	0.7281E-07	0.1085E-05
41	0.2870E+05	0.6050E-07	0.9245E-06
42	0.2940E+05	0.5025E-07	0.7865E-06
43	0.3010E+05	0.4170E-07	0.6683E-06
44	0.3080E+05	0.3458E-07	0.5671E-06
45	0.3150E+05	0.2865E-07	0.4805E-06
46	0.3220E+05	0.2371E-07	0.4064E-06
47	0.3290E+05	0.1959E-07	0.3432E-06
48	0.3360E+05	0.1617E-07	0.2893E-06
49	0.3430E+05	0.1332E-07	0.2433E-06
50	0.3500E+05	0.1096E-07	0.2042E-06

DEAD POLYMER CHAIN LENGTH DISTRIBUTION

NU.	MOL. WT.	NUMB. BASIS	WT. BASIS(NORMAL)
0	0.1000E+01	0.4579E+00	0.4217E-06
1	0.5830E+03	0.7485E-01	0.4060E-04
2	0.1177E+04	0.5342E-01	0.5790E-04
3	0.1765E+04	0.4136E-01	0.6723E-04
4	0.2353E+04	0.3326E-01	0.7206E-04
5	0.2941E+04	0.2739E-01	0.7417E-04
6	0.3529E+04	0.2295E-01	0.7460E-04
7	0.4117E+04	0.1951E-01	0.7396E-04
8	0.4705E+04	0.1677E-01	0.7267E-04
9	0.5293E+04	0.1456E-01	0.7095E-04
10	0.5881E+04	0.1273E-01	0.6896E-04
11	0.6469E+04	0.1121E-01	0.6680E-04
12	0.7057E+04	0.9928E-02	0.6452E-04
13	0.7645E+04	0.8828E-02	0.6215E-04
14	0.8233E+04	0.7877E-02	0.5972E-04
15	0.8821E+04	0.7046E-02	0.5723E-04
16	0.9409E+04	0.6315E-02	0.5471E-04
17	0.9997E+04	0.5666E-02	0.5216E-04
18	0.1059E+05	0.5087E-02	0.4959E-04
19	0.1117E+05	0.4559E-02	0.4701E-04
20	0.1176E+05	0.4102E-02	0.4443E-04
21	0.1235E+05	0.3682E-02	0.4187E-04
22	0.1294E+05	0.3301E-02	0.3933E-04
23	0.1352E+05	0.2957E-02	0.3682E-04
24	0.1411E+05	0.2645E-02	0.3437E-04
25	0.1470E+05	0.2362E-02	0.3198E-04
26	0.1529E+05	0.2106E-02	0.2965E-04
27	0.1588E+05	0.1874E-02	0.2740E-04
28	0.1647E+05	0.1664E-02	0.2523E-04
29	0.1705E+05	0.1475E-02	0.2316E-04
30	0.1764E+05	0.1304E-02	0.2118E-04
31	0.1823E+05	0.1150E-02	0.1930E-04
32	0.1882E+05	0.1011E-02	0.1752E-04
33	0.1941E+05	0.8870E-03	0.1585E-04
34	0.1999E+05	0.7757E-03	0.1428E-04
35	0.2058E+05	0.6762E-03	0.1282E-04
36	0.2117E+05	0.5875E-03	0.1145E-04
37	0.2176E+05	0.5086E-03	0.1019E-04
38	0.2234E+05	0.4385E-03	0.9024E-05
39	0.2293E+05	0.3766E-03	0.7952E-05
40	0.2352E+05	0.3218E-03	0.6971E-05
41	0.2411E+05	0.2737E-03	0.6076E-05
42	0.2470E+05	0.2314E-03	0.5263E-05
43	0.2529E+05	0.1944E-03	0.4527E-05
44	0.2587E+05	0.1622E-03	0.3865E-05
45	0.2646E+05	0.1342E-03	0.3270E-05
46	0.2705E+05	0.1100E-03	0.2739E-05
47	0.2764E+05	0.8908E-04	0.2267E-05
48	0.2823E+05	0.7116E-04	0.1850E-05
49	0.2881E+05	0.5586E-04	0.1482E-05
50	0.2940E+05	0.4287E-04	0.1161E-05

PARTICLE SIZE DISTRIBUTION [CUMULATIVE]

NO	DIA("A)	FRACTION
0	0.5000E+02	0.0000E+00
1	0.4553E+03	0.1683E+00
2	0.5735E+03	0.3083E+00
3	0.6564E+03	0.4248E+00
4	0.7225E+03	0.5216E+00
5	0.7782E+03	0.6022E+00
6	0.8270E+03	0.6692E+00
7	0.8706E+03	0.7249E+00
8	0.9102E+03	0.7713E+00
9	0.9466E+03	0.8098E+00
10	0.9805E+03	0.8419E+00
11	0.1012E+04	0.8685E+00
12	0.1042E+04	0.8907E+00
13	0.1070E+04	0.9091E+00
14	0.1097E+04	0.9244E+00
15	0.1122E+04	0.9372E+00
16	0.1147E+04	0.9478E+00
17	0.1170E+04	0.9566E+00
18	0.1193E+04	0.9639E+00
19	0.1214E+04	0.9700E+00
20	0.1235E+04	0.9751E+00
21	0.1256E+04	0.9793E+00
22	0.1275E+04	0.9828E+00
23	0.1294E+04	0.9857E+00
24	0.1313E+04	0.9881E+00
25	0.1331E+04	0.9901E+00
26	0.1348E+04	0.9918E+00
27	0.1365E+04	0.9932E+00
28	0.1382E+04	0.9944E+00
29	0.1398E+04	0.9953E+00
30	0.1414E+04	0.9961E+00
31	0.1430E+04	0.9968E+00
32	0.1445E+04	0.9974E+00
33	0.1460E+04	0.9978E+00
34	0.1474E+04	0.9982E+00
35	0.1489E+04	0.9985E+00
36	0.1503E+04	0.9988E+00
37	0.1516E+04	0.9990E+00
38	0.1530E+04	0.9992E+00
39	0.1543E+04	0.9993E+00
40	0.1556E+04	0.9995E+00
41	0.1569E+04	0.9996E+00
42	0.1582E+04	0.9997E+00
43	0.1594E+04	0.9997E+00
44	0.1607E+04	0.9998E+00
45	0.1619E+04	0.9998E+00
46	0.1631E+04	0.9999E+00
47	0.1642E+04	0.9999E+00
48	0.1654E+04	0.1000E+01
49	0.1665E+04	0.1000E+01
50	0.1676E+04	0.1000E+01

CHE-1979-M-BAL-M 59503

59503

Date Slip 59503

This book is to be returned on the
date last stamped.

CD 6.72.9